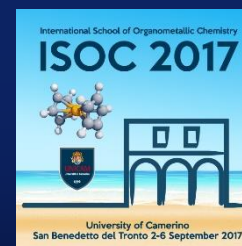




UNIVERSITÀ DEGLI STUDI
DI MILANO



Catalytic carbophilic activation in synthetic organic chemistry: catalysis by gold π acids

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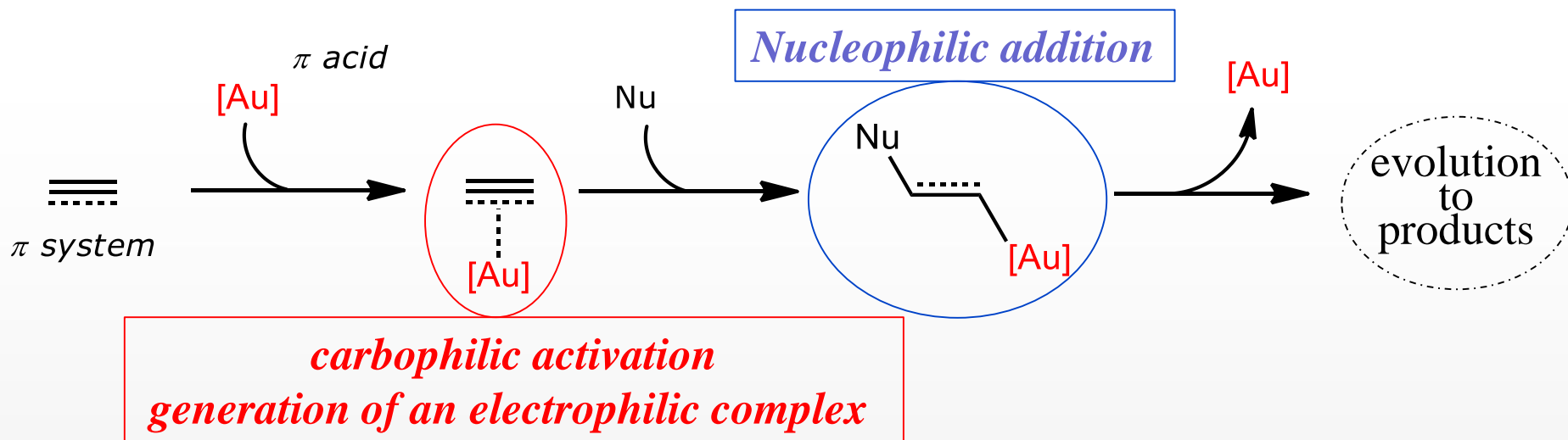
Activation of π -systems (alkenes, alkynes and allenes)
towards nucleophilic additions



Catalytic *carbophilic activation* in synthetic organic chemistry:
catalysis by *gold π acids*

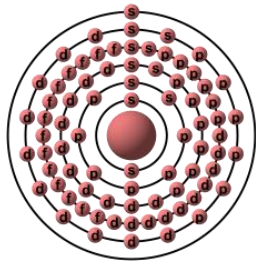


X^- = non-coordinating counterion



Basic mechanism of π acid catalysis exemplified by the addition of a generic nucleophile across a π system

Gold as element



electronic configuration $[Xe] 4f^{14}5d^{10}6s^1$
 atomic number 79
 atomic weight 197

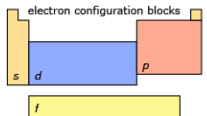
The Periodic Table of the Elements

by Robert Cawthon version 1.13

Legend:

- alkali metals
- alkaline metals
- other metals
- transition metals
- lanthanoids
- actinoids
- metalloids
- nonmetals
- halogens
- noble gases
- unknown elements
- radioactive elements have masses in parentheses

																		18				
1																	2					
1	H Hydrogen 1.00794 101.08																	He Helium 4.002602 2972.3				
2	Li Lithium 6.941 690.5	Be Beryllium 9.012182 898.9															B Boron 10.811 1080.6	C Carbon 12.0107 1080.5	N Nitrogen 14.0067 1400.3	O Oxygen 15.9994 1513.9	F Fluorine 18.998403 1081.0	Ne Neon 20.1797 2060.7
3	Na Sodium 22.98976 468.9	Mg Magnesium 24.3050 738.7															Al Aluminium 26.98153 577.5	Si Silicon 28.0855 785.5	P Phosphorus 30.97396 1011.6	S Sulfur 32.065 390.9	Cl Chlorine 35.453 1051.2	Ar Argon 39.948 1038.6
4	K Potassium 39.0983 418.8	Ca Calcium 40.078 790.9	Sc Scandium 44.95591 895.4	Ti Titanium 47.867 898.8	V Vanadium 50.9415 890.9	Cr Chromium 51.9962 862.9	Mn Manganese 54.938025 717.3	Fe Iron 55.845 762.5	Co Cobalt 58.93319 760.4	Ni Nickel 58.6934 737.1	Cu Copper 63.546 745.5	Zn Zinc 65.38 900.4	Ga Gallium 69.723 578.9	Ge Germanium 72.64 762.0	As Arsenic 74.92160 947.0	Se Selenium 78.96 941.0	Br Bromine 79.904 1139.9	Kr Krypton 83.798 1100.6				
5	Rb Rubidium 85.4678 400.0	Sr Strontium 87.62 548.5	Y Yttrium 88.90585 900.0	Zr Zirconium 91.224 90.1	Nb Niobium 92.90638 909.2	Mo Molybdenum 95.96 959.6	Tc Technetium (98) 98.9062	Ru Ruthenium 101.07 101.1	Rh Rhodium 102.9055 101.1	Pd Palladium 106.42 106.4	Ag Silver 107.8682 107.9	Cd Cadmium 112.414 112.4	In Indium 114.818 114.8	Sn Tin 118.710 118.7	Sb Antimony 121.760 121.8	Te Tellurium 127.60 127.6	I Iodine 126.9044 126.9	Xe Xenon 131.293 131.3				
6	Cs Caesium 132.9054 502.7	Ba Barium 137.327 562.5	Lu Lutetium 174.9668 563.8	Hf Hafnium 178.49 563.8	Ta Tantalum 180.9478 563.8	W Tungsten 183.84 563.8	Re Rhenium 186.207 563.8	Os Osmium 190.23 563.8	Ir Iridium 192.222 563.8	Pt Platinum 195.084 563.8	Au Gold 196.9665 563.8	Hg Mercury 200.59 563.8	Tl Thallium 204.3833 563.8	Pb Lead 207.2 563.8	Bi Bismuth 208.9804 563.8	Po Polonium (210) 210	At Astatine (210) 210	Rn Radon (222) 222				
7	Fr Francium (223) 223	Ra Radium (226) 226	Lr Lawrencium (262) 262	Rf Rutherfordium (261) 261	Db Dubnium (262) 262	Sg Seaborgium (266) 266	Bh Bohrium (264) 264	Hs Hassium (277) 277	Mt Meitnerium (268) 268	Ds Darmstadtium (271) 271	Rg Roentgenium (272) 272	Cn Copernicium (285) 285	Uut Ununtrium (284) 284	Fl Flerovium (289) 289	Uup Ununpentium (288) 288	Lv Livermorium (292) 292	Uus Ununseptium (117) 117	Uuo Ununoctium (118) 118				

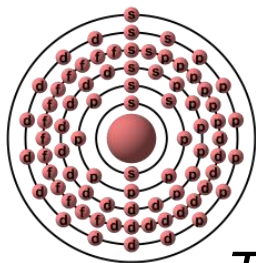


notes

- * as of yet, elements 113,115,117 and 118 have no official name designated by the IUPAC.
- * 1 kJ/mol ≈ 96.485 eV.
- * all elements are implied to have an oxidation state of zero.

138.9054 588.1	140.116 594.4	140.9076 627.0	144.242 593.1	(145) 540.0	150.36 544.5	151.964 547.1	157.25 589.4	158.9253 565.8	162.500 579.0	164.9303 581.0	167.259 589.3	168.9342 586.7	173.054 603.4
La Lanthanum [Xe] 5d 6s	Ce Cerium [Xe] 4f 5d 6s	Pr Praseodymium [Xe] 4f 6s	Nd Neodymium [Xe] 4f 6s	Pm Promethium [Xe] 4f 6s	Sm Samarium [Xe] 4f 6s	Eu Europium [Xe] 4f 6s	Gd Gadolinium [Xe] 4f 6s	Tb Terbium [Xe] 4f 6s	Dy Dysprosium [Xe] 4f 6s	Ho Holmium [Xe] 4f 6s	Er Erbium [Xe] 4f 6s	Tm Thulium [Xe] 4f 6s	Yb Ytterbium [Xe] 4f 6s
(227) 227	232.0380 232	231.0358 231	238.0289 238	(237) 237	(244) 244	(243) 243	(247) 247	(247) 247	(251) 251	(252) 252	(257) 257	(258) 258	(259) 259
Ac Actinium [Rn] 5f 7s	Th Thorium [Rn] 6d 7s	Pa Protactinium [Rn] 5f 6d 7s	U Uranium [Rn] 5f 6d 7s	Np Neptunium [Rn] 5f 6d 7s	Pu Plutonium [Rn] 5f 7s	Am Americium [Rn] 6d 7s	Cm Curium [Rn] 6f 7s	Bk Berkelium [Rn] 6f 7s	Cf Californium [Rn] 6f 7s	Es Einsteinium [Rn] 6f 7s	Fm Fermium [Rn] 6f 7s	Md Mendelevium [Rn] 6f 7s	No Nobelium [Rn] 6f 7s

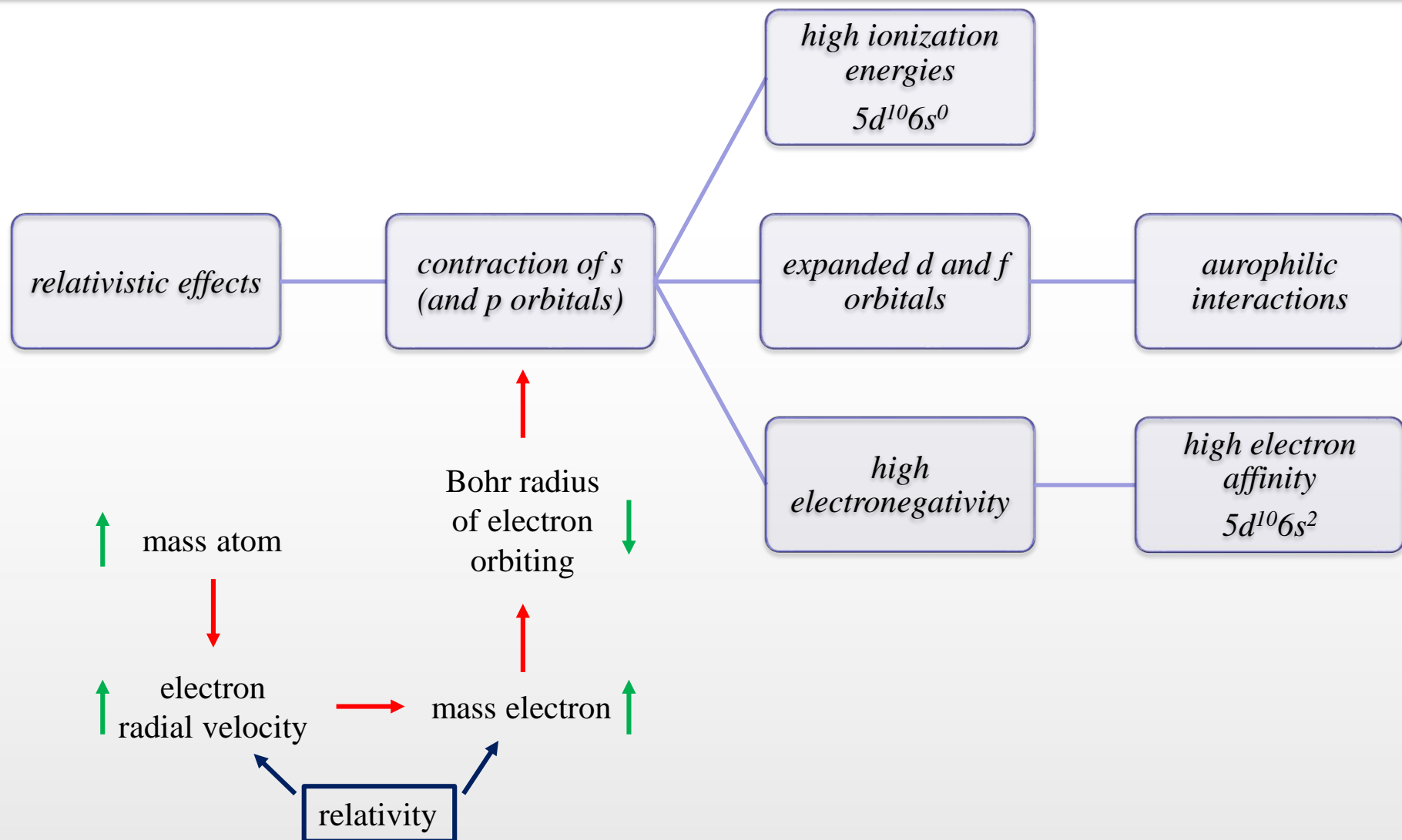
Gold as element: valence electron configuration $5d^{10}6s^1$



TM properties (ionization potential, electronegativity)

3	4	5	6	7	8	9	10	11	12
44.95591 21 633.1 1.36 Sc Scandium [Ar] 3d ¹ 4s ²	47.867 22 658.8 1.54 Ti Titanium [Ar] 3d ² 4s ²	50.9415 23 650.9 1.63 V Vanadium [Ar] 3d ³ 4s ²	51.9962 24 652.9 1.66 Cr Chromium [Ar] 3d ⁵ 4s ¹	54.93804 25 717.3 1.55 Mn Manganese [Ar] 3d ⁵ 4s ²	55.845 26 762.5 1.83 Fe Iron [Ar] 3d ⁶ 4s ²	58.93319 27 760.4 1.91 Co Cobalt [Ar] 3d ⁷ 4s ²	58.6934 28 737.1 1.88 Ni Nickel [Ar] 3d ⁸ 4s ²	63.546 29 745.5 1.90 Cu Copper [Ar] 3d ¹⁰ 4s ¹	65.38 30 906.4 1.65 Zn Zinc [Ar] 3d ¹⁰ 4s ²
88.90585 39 800.0 1.22 Y Yttrium [Kr] 4d ¹ 5s ²	91.224 40 640.1 1.33 Zr Zirconium [Kr] 4d ² 5s ²	92.90638 41 852.1 1.60 Nb Niobium [Kr] 4d ⁴ 5s ¹	95.96 42 684.3 2.16 Mo Molybdenum [Kr] 4d ⁵ 5s ¹	(98) 43 702.0 1.90 Tc Technetium [Kr] 4d ⁵ 5s ²	101.07 44 710.2 2.20 Ru Ruthenium [Kr] 4d ⁷ 5s ¹	102.9055 45 719.7 2.28 Rh Rhodium [Kr] 4d ⁸ 5s ¹	106.42 46 804.4 2.20 Pd Palladium [Kr] 4d ¹⁰	107.8682 47 731.0 1.93 Ag Silver [Kr] 4d ¹⁰ 5s ¹	112.411 48 867.8 1.69 Cd Cadmium [Kr] 4d ¹⁰ 5s ²
174.9668 71 523.5 1.27 Lu Lutetium [Xe] 4f ¹⁴ 5d ¹ 6s ²	178.49 72 658.5 1.30 Hf Hafnium [Xe] 4f ¹⁴ 5d ² 6s ²	180.9478 73 761.0 1.50 Ta Tantalum [Xe] 4f ¹⁴ 5d ³ 6s ²	183.84 74 770.0 2.36 W Tungsten [Xe] 4f ¹⁴ 5d ⁴ 6s ²	186.207 75 760.0 1.90 Re Rhenium [Xe] 4f ¹⁴ 5d ⁵ 6s ²	190.23 76 840.0 2.20 Os Osmium [Xe] 4f ¹⁴ 5d ⁶ 6s ²	192.217 77 880.0 2.20 Ir Iridium [Xe] 4f ¹⁴ 5d ⁷ 6s ²	195.084 78 870.0 2.28 Pt Platinum [Xe] 4f ¹⁴ 5d ⁹ 6s ¹	196.9665 79 890.1 2.54 Au Gold [Xe] 4f ¹⁴ 5d ¹⁰ 6s ¹	200.59 80 1007.1 2.00 Hg Mercury [Xe] 4f ¹⁴ 5d ¹⁰ 6s ²
(262) 103 470.0 Lr Lawrencium [Rn] 5f ¹⁴ 7s ² 7p ¹	(261) 104 580.0 Rf Rutherfordium [Rn] 5f ¹⁴ 6d ² 7s ²	(262) 105 Db Dubnium	(266) 106 Sg Seaborgium	(264) 107 Bh Bohrium	(277) 108 Hs Hassium	(268) 109 Mt Meitnerium	(271) 110 Ds Darmstadtium	(272) 111 Rg Roentgenium	(285) 112 Cn Copernicium

Gold as element: relativistic effects



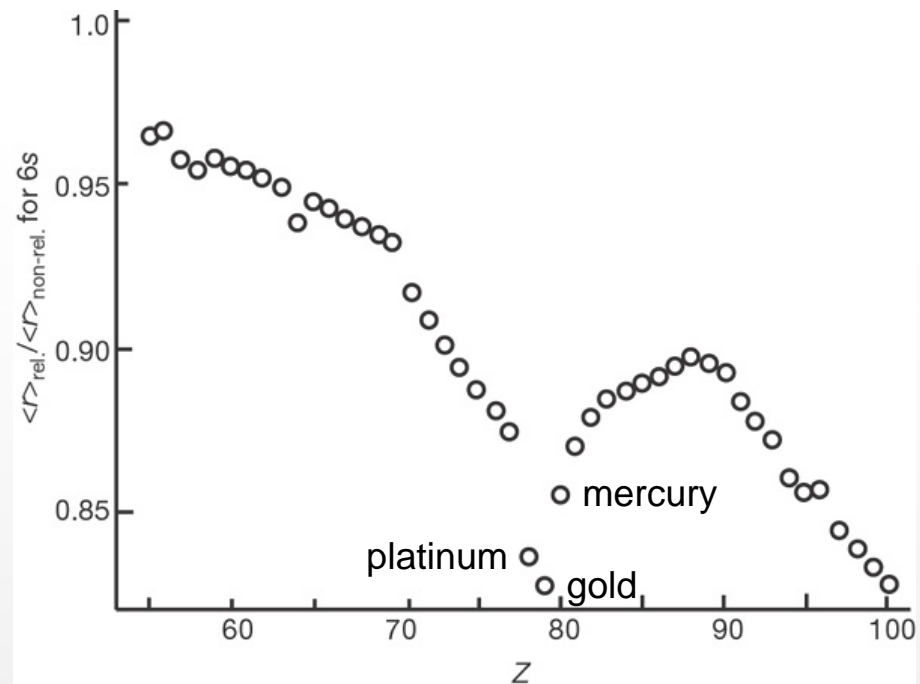
S. Kramer, F. Gagosz in "Gold Catalysis An Homogeneous Approach", Eds. F. D. Toste, V. Michelet, 2014 World Scientific Publishing, ch. 1.

D. J. Gorin, F. D. Toste *Nature* **2007**, 446, 395.

A. Fürstner, P. W. Davies *Angew. Chem. Int. Ed.* **2007**, 46, 3410.

relativistic effects

contraction of *s*
(and *p* orbitals)



Calculated relativistic contraction of the 6s orbital.

R. Coquet, K. L. Howard, D. J. Willock *Chem. Soc. Rev.* **2008**, 37, 2046.

E. Pykko *Chem. Rev.* **1988**, 88, 563.



counterions



Ligand(s)



Ligand



- A. Fürstner *Chem. Soc. Rev.* **2009**, 38, 3208.
A. Fürstner, P. W. Davies *Angew. Chem. Int. Ed.* **2007**, 46, 3410.
A. Fürstner *Acc. Chem. Res.* **2014**, 47, 925.
B. Ranieri, I. Escofeta, A. M. Echavarren *Org. Biomol. Chem.* **2015**, 13, 7103.
A. S. K. Hashmi *Chem. Rev.* **2007**, 107, 3180.
A. S. K. Hashmi, G. J. Hutchings *Angew. Chem. Int. Ed.* **2006**, 45, 7896.
C. Obradorsa, A. M. Echavarren *Chem. Commun.* **2014**, 50, 16.
D. J. Gorin, B. D. Sherry, F. D. Toste *Chem. Rev.* **2008**, 108, 3351.
A. Arcadi *Chem. Rev.* **2008**, 108, 3266

gold(I) and gold(III) salts, aurous and auric compounds

$AuCl_3$ and $AuBr_3$



*Less coordinating X
More electrophilic Au*

counterions



$NaAuCl_4$ and $KAuCl_4$

$AuCl$

gold(I) complexes



ligand Gold(I) counterion

Tuning of electronic
properties

Tuning of steric
properties

Optimized
reactivity and
selectivity

gold(I) complexes

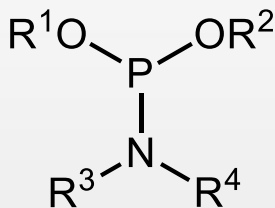


gold chloride complex

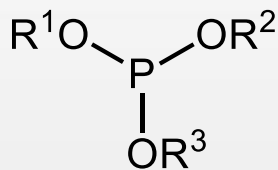


gold chloride complex

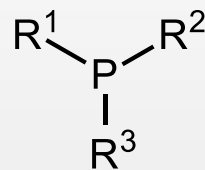
General structure of most commonly used ligands



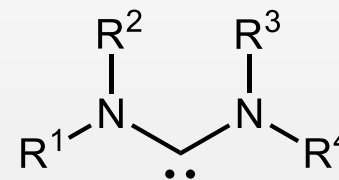
phosphoroamidite



phosphite



phosphine

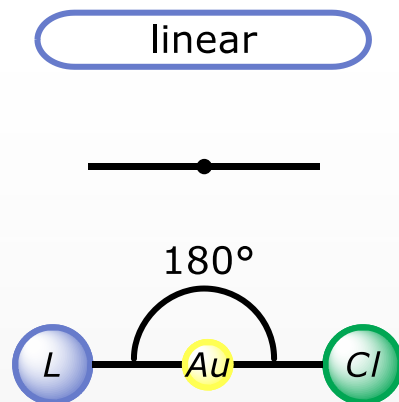


carbene

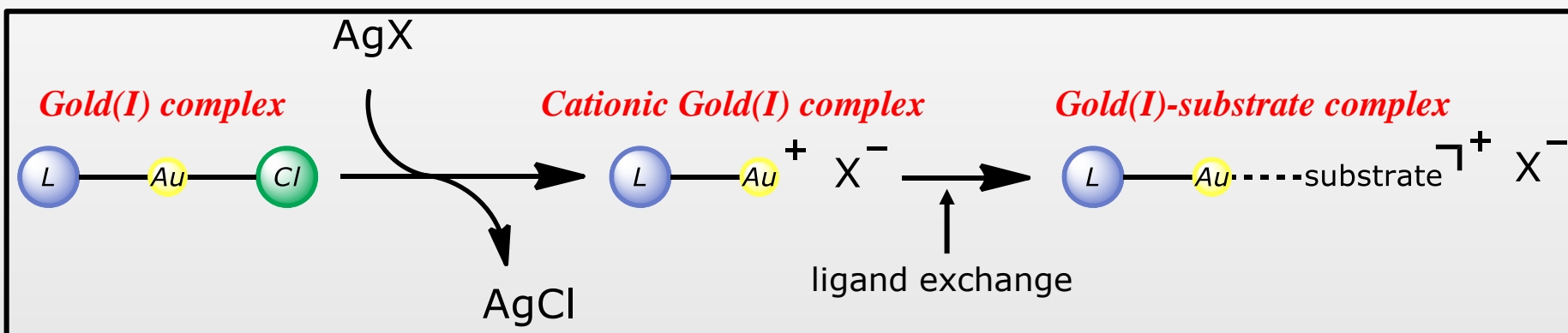
electrodonor properties

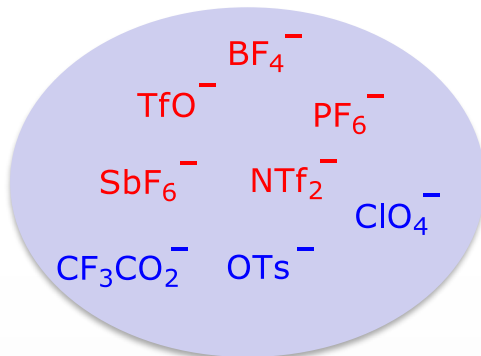
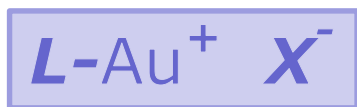


gold(I) complexes

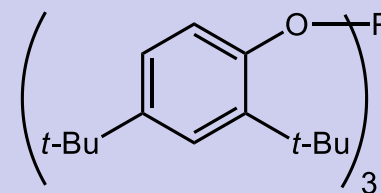
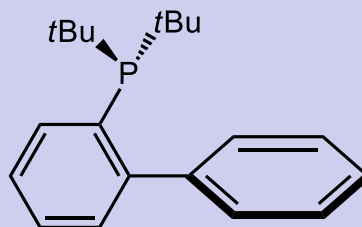
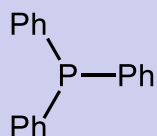
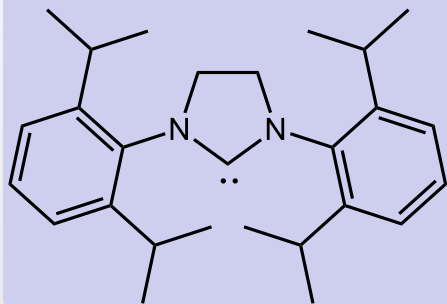
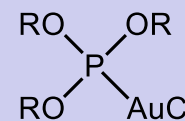
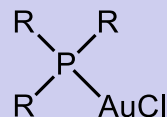
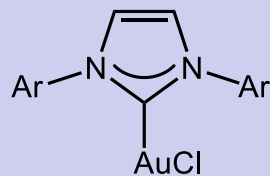


Gold(I) complexes are d^{10} complexes and present a linear, bi-coordinated geometry comprising the metal center within the ligand and the chloride atom.





Less and more coordinating ions



iPr carbene ligand

triphenylphosphine

JohnPhos

tri-(2,4-ditBu)phenyl phosphite

electrophilicity

Properties of cationic gold(I) species



- ✓ strength of the $L-Au$ bond
- ✓ Lewis acidity, π acids
- ✓ do not tend to undergo oxidative addition
- ✓ act as big soft proton

P. Schwerdtfeger, H. L. Hermann, H. Schmidbaur *Inorg. Chem.* **2003**, *42*, 1334.

Properties of cationic gold(I) species

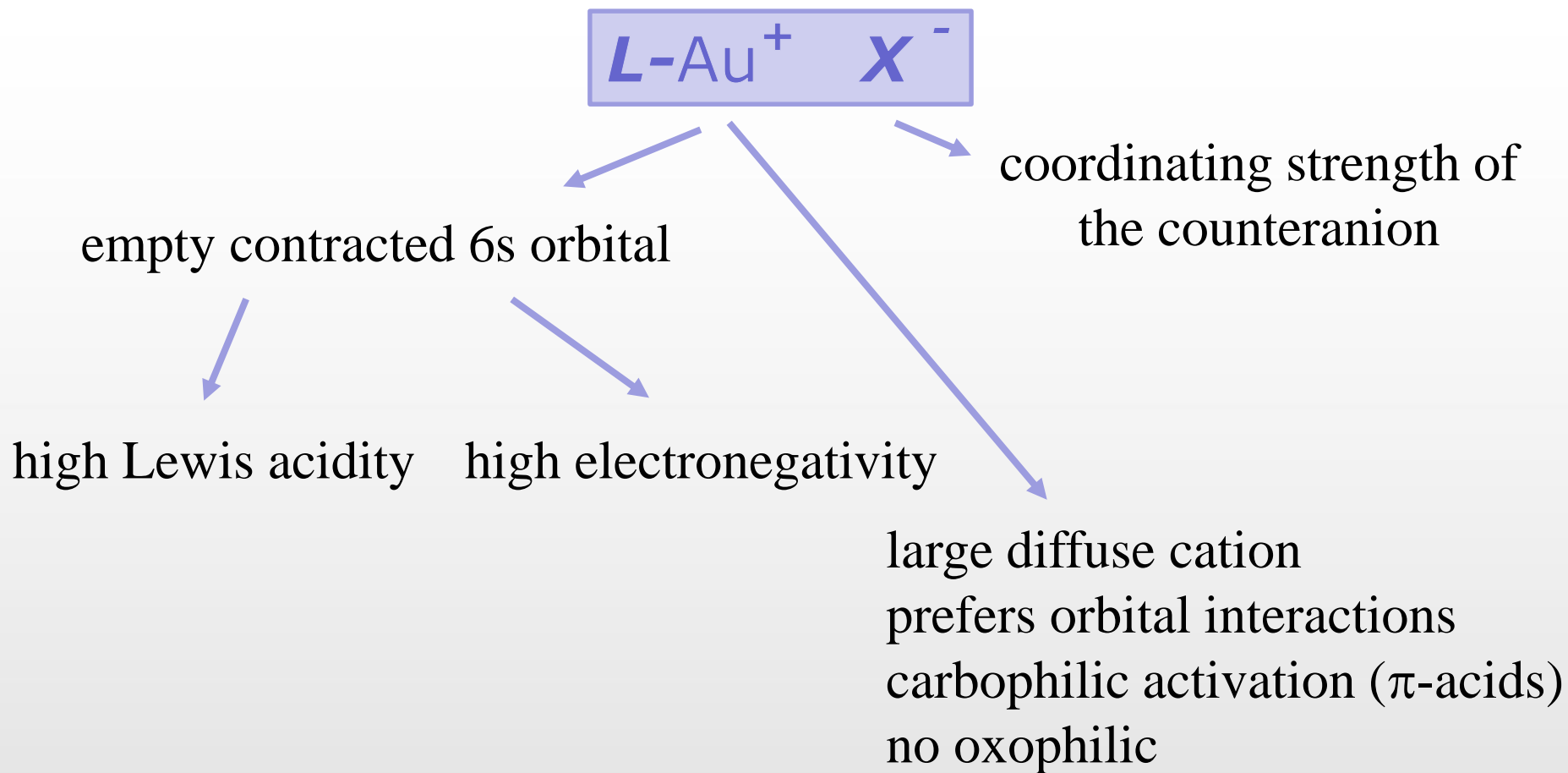


- ✓ the strength of the L-Au bond depends on relativistic contraction of the 6s orbital
- ✓ the strength of the L-Au bond is greater than in Ag and Cu
- ✓ ligand exchange in solution is not observed

P. Schwerdtfeger, H. L. Hermann, H. Schmidbaur *Inorg. Chem.* **2003**, *42*, 1334.

Properties of cationic gold(I) species

- ✓ Lewis acidity, 'soft' Lewis acid, preferring 'soft' electrophiles



Properties of cationic gold(I) species

- ✓ do not tend to undergo oxidative addition



- ✓ not particularly nucleophilic,
- ✓ do not tend to undergo oxidative addition,
- ✓ tolerant to oxygen.

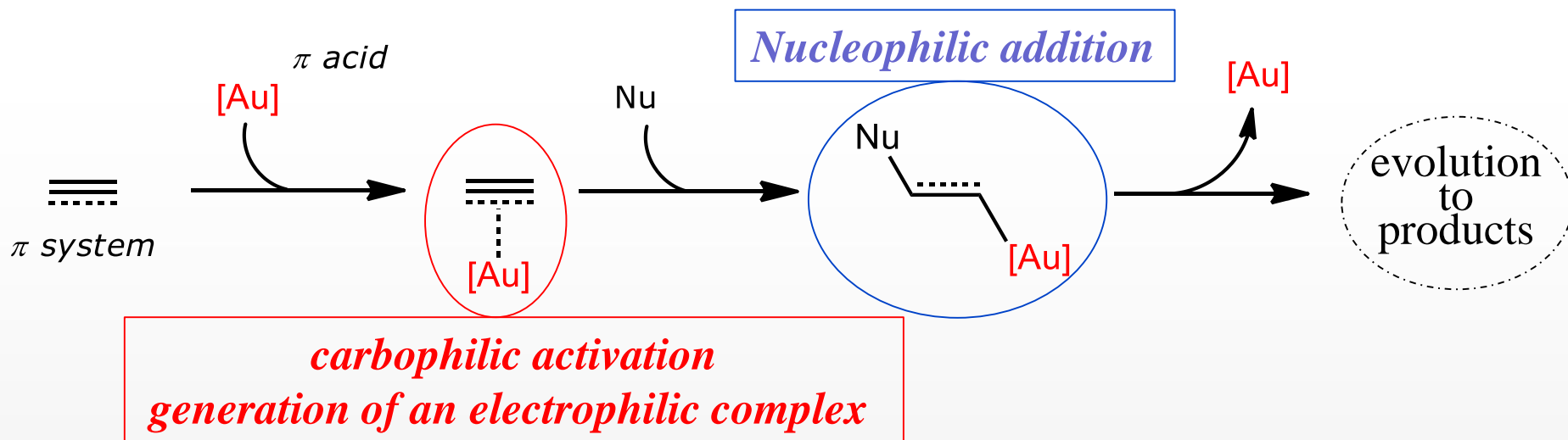
Properties of cationic gold(I) species

- ✓ big soft proton

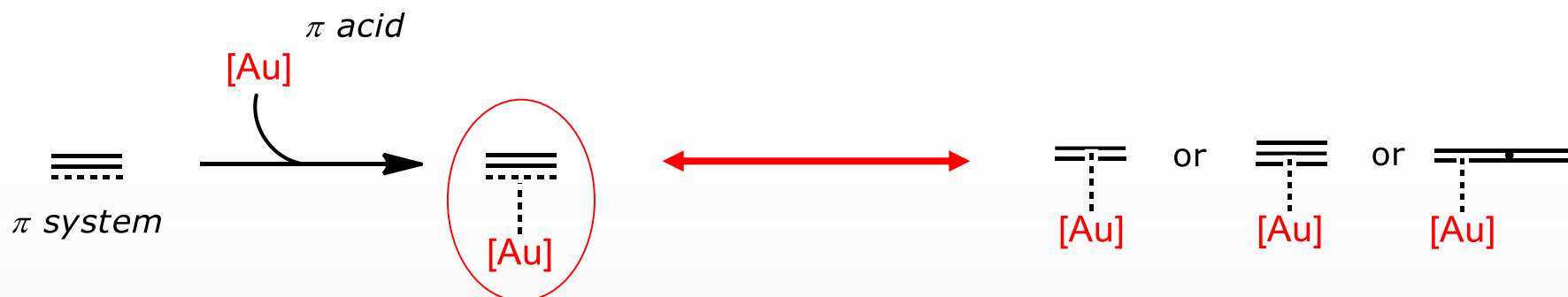


- ✓ isolobal to H^+
- ✓ catalysis under “acidic” conditions

cfr Pt(II) and Hg(II)

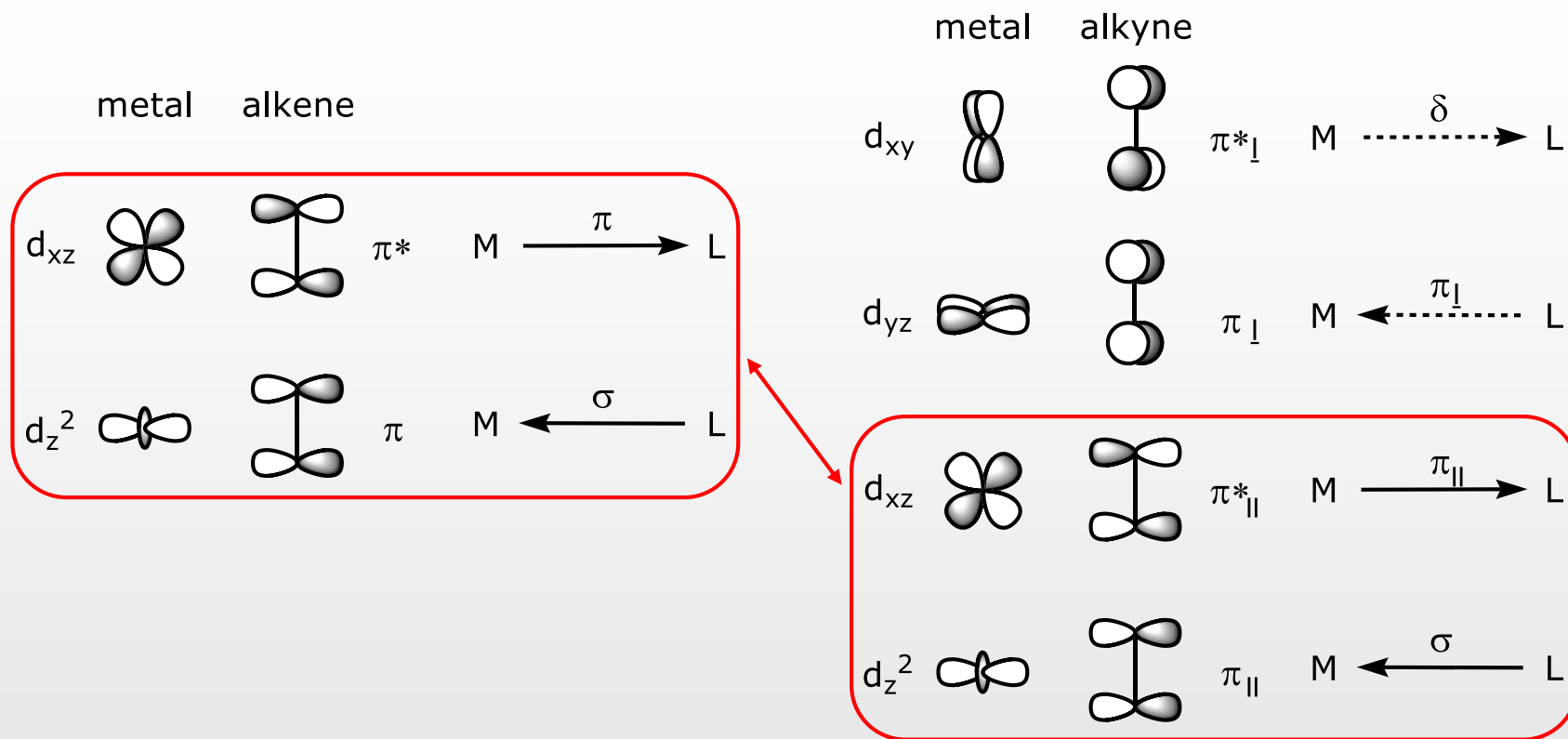
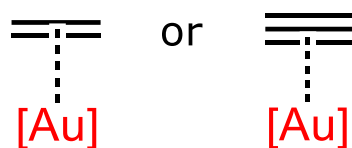


Basic mechanism of π acid catalysis exemplified by the addition of a generic nucleophile across a π system



carbophilic activation
generation of an electrophilic complex

π acid coordination to unsaturated systems





Au⁺ acetylene complex

contribution

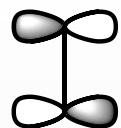
metal

alkyne

1 %



d_{xy}



π^*_{\perp}

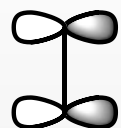


metal to ligand back-donation

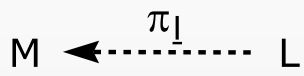
7 %



d_{yz}



π_{\perp}

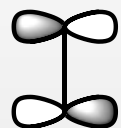
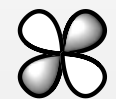


ligand to metal π donation

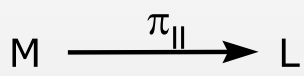
27 %



d_{xz}



π^*_{\parallel}

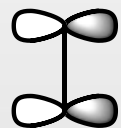


in-plane back-donation

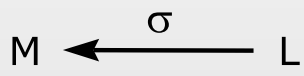
65 %



d_z^2

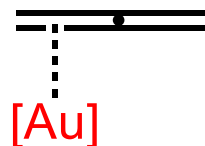


π_{\parallel}

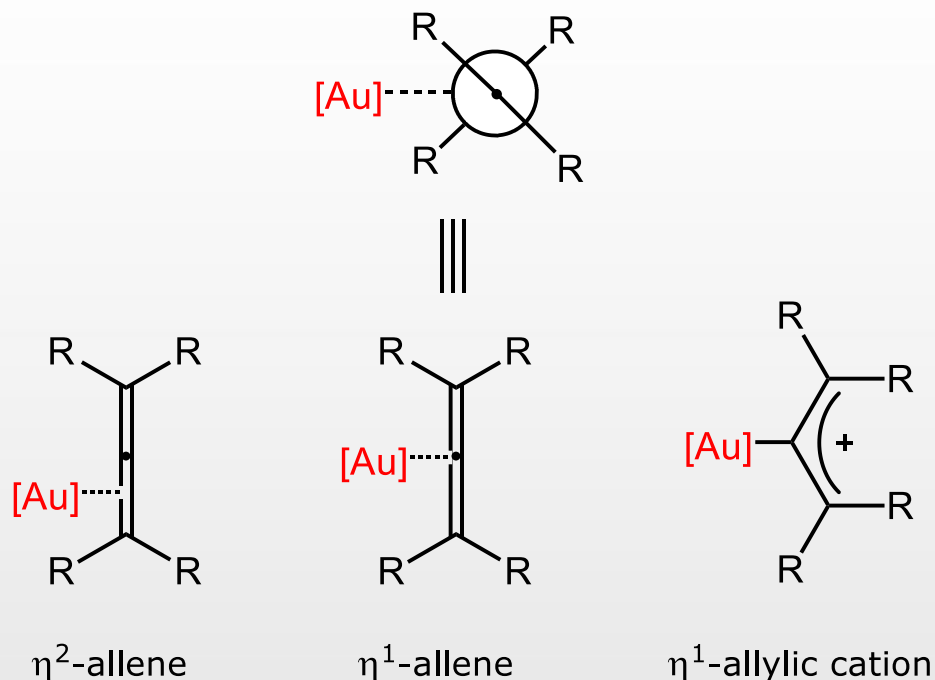


σ interaction

π acid coordination to unsaturated systems



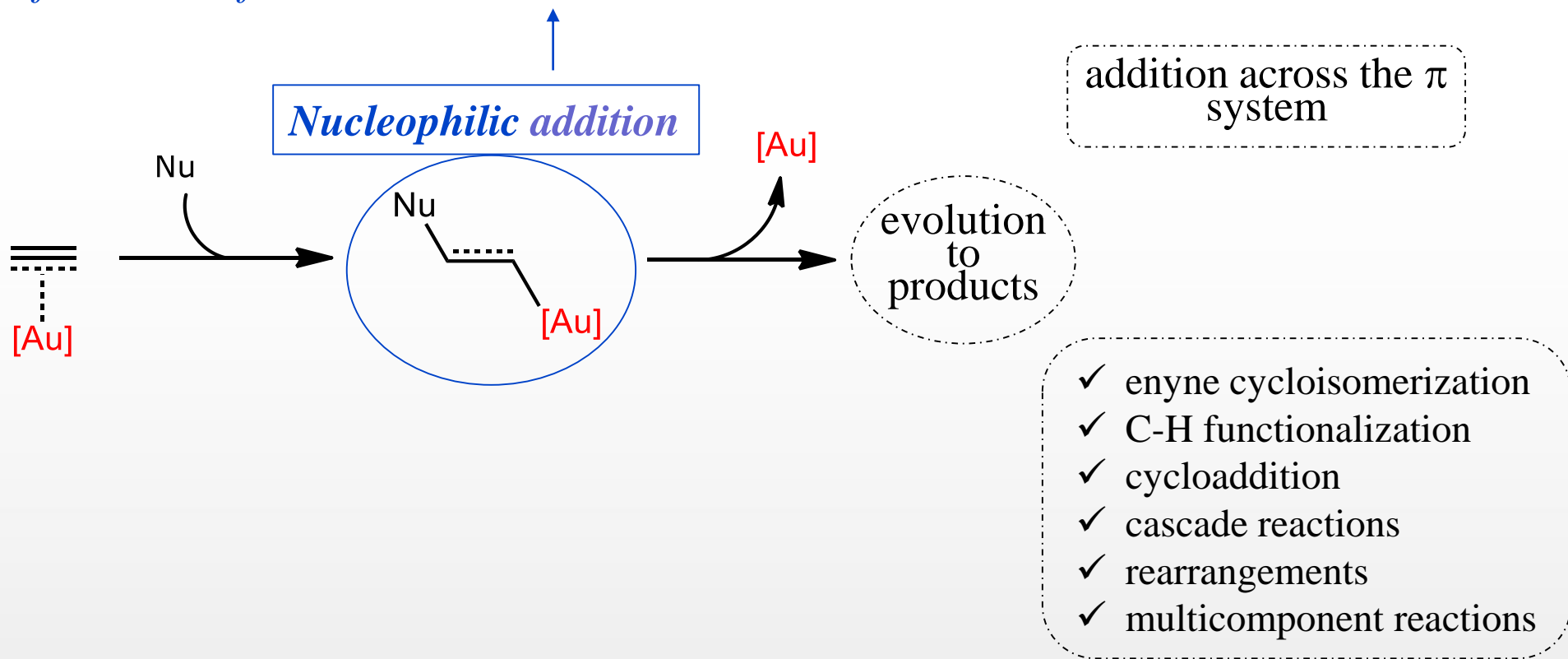
Potential coordination mode for gold π allene complexes

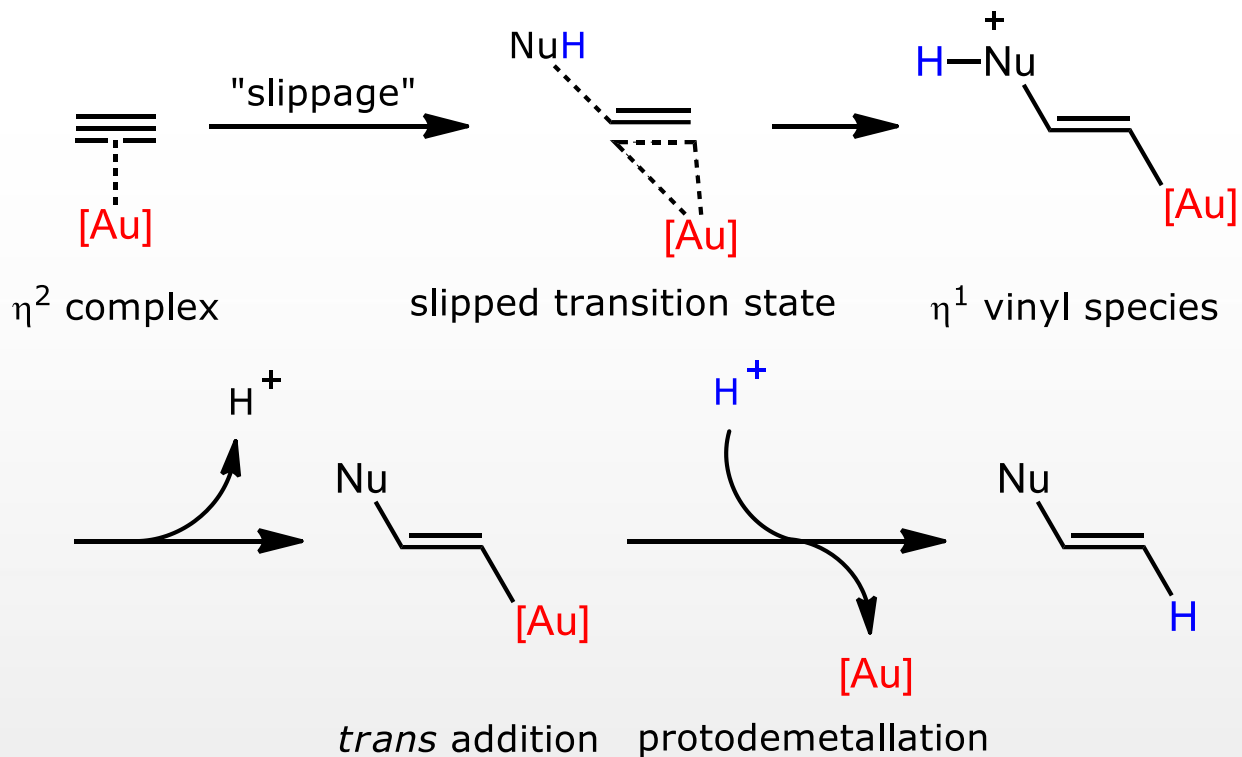


T. J. Brown, A. Sugie, M. G. Dickens, R. A. Widenhoefer, *Organometallics* **2010**, 29, 4207.

T. J. Brown, A. Sugie, M. G. Dickens, R. A. Widenhoefer, *Chem. Eur. J.* **2012**, 18, 6959.

formation of carbon-heteroatom and carbon-carbon bonds

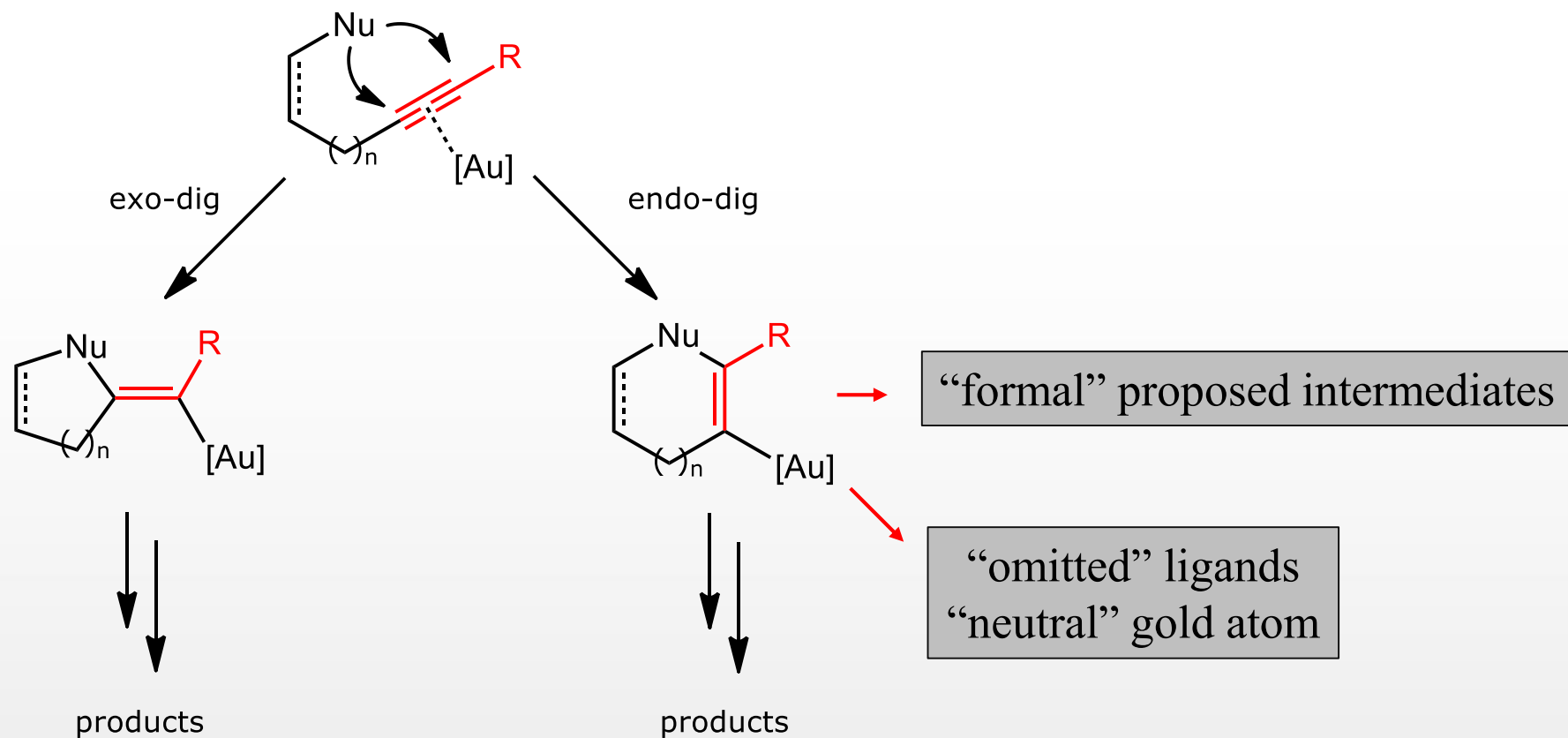




R. Dorel, A. M. Echavarren *Chem. Rev.* **2015**, *115*, 9028.

W. Debrouwer, T. S. A. Heugebaert, B. I. Roman, C. V. Stevens *Adv. Synth. Catal.* **2015**, *357*, 2975.

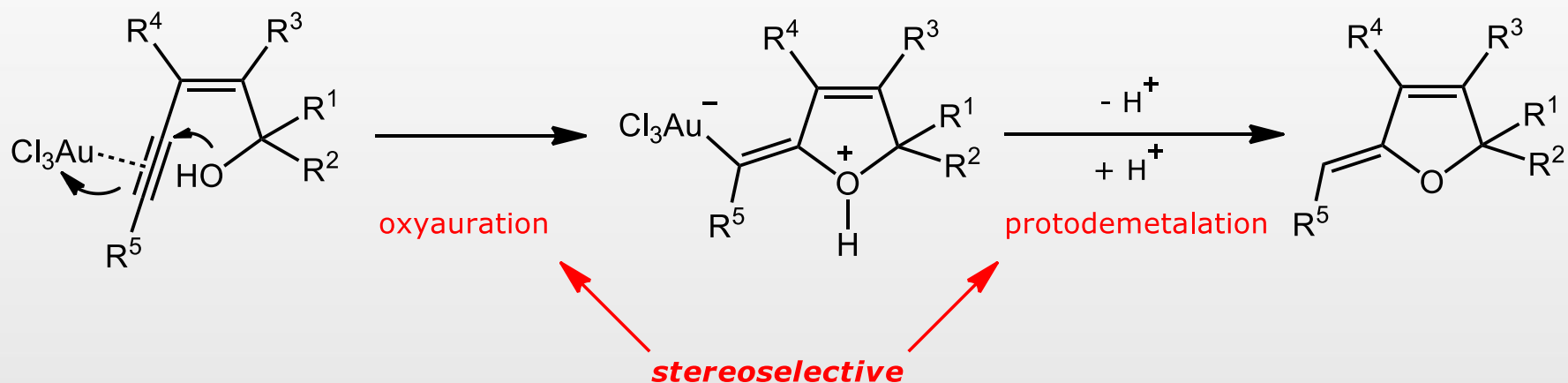
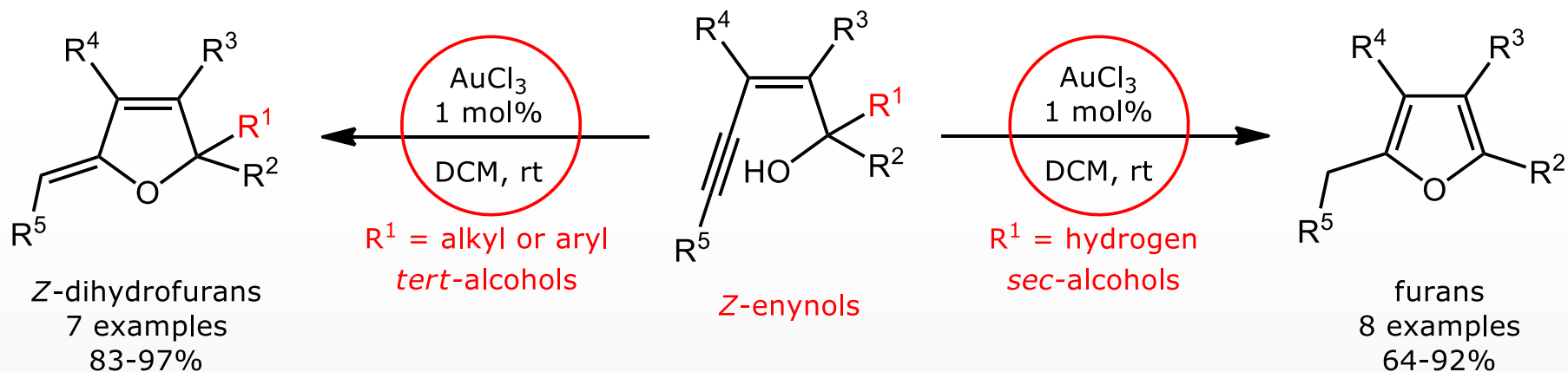
Nucleophilic addition to π activated carbon-carbon triple bonds, synthesis of heterocycles



R. Dorel, A. M. Echavarren *Chem. Rev.* **2015**, *115*, 9028.

W. Debrouwer, T. S. A. Heugebaert, B. I. Roman, C. V. Stevens *Adv. Synth. Catal.* **2015**, *357*, 2975.

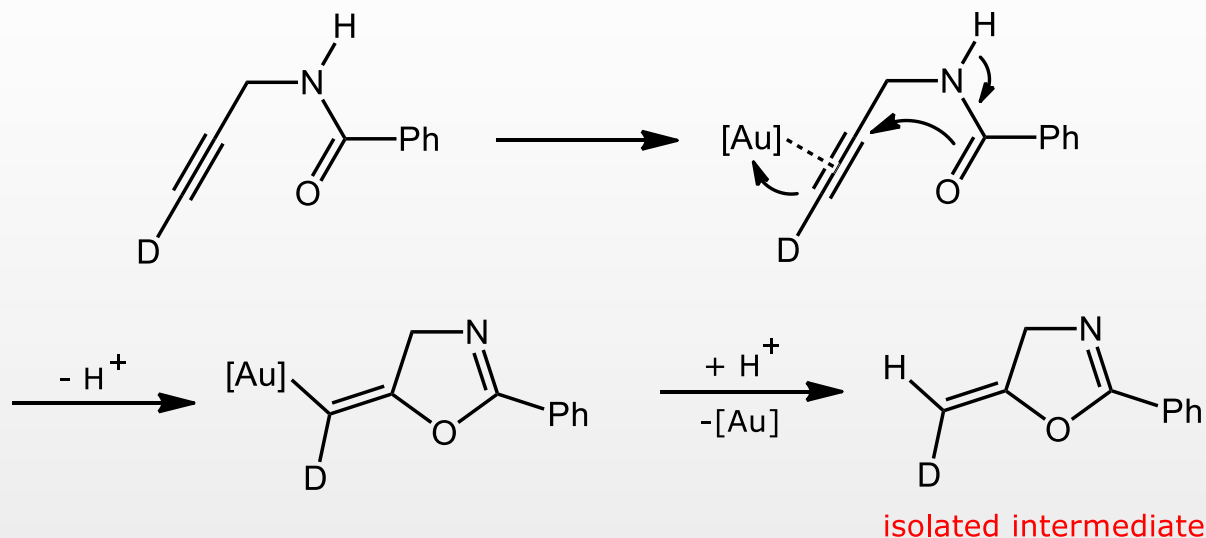
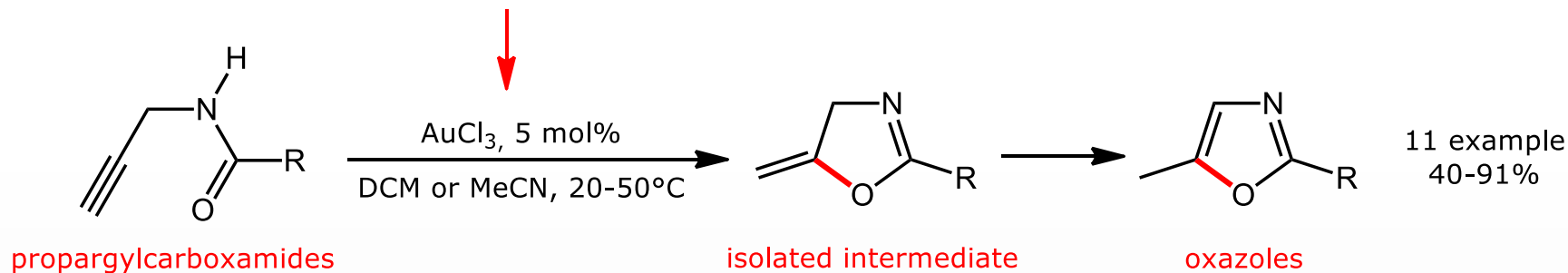
gold-catalyzed 5-exo-dig cyclization of (Z)-enynols



A. S. K. Hashmi, L. Schwarz, J. H. Choi, T. M. Frost *Angew. Chem., Int. Ed.* **2000**, 39, 2285.

Y. Liu, F. Song, Z. Song, M. Liu, B. Yan *Org. Lett.* **2005**, 7, 5409.

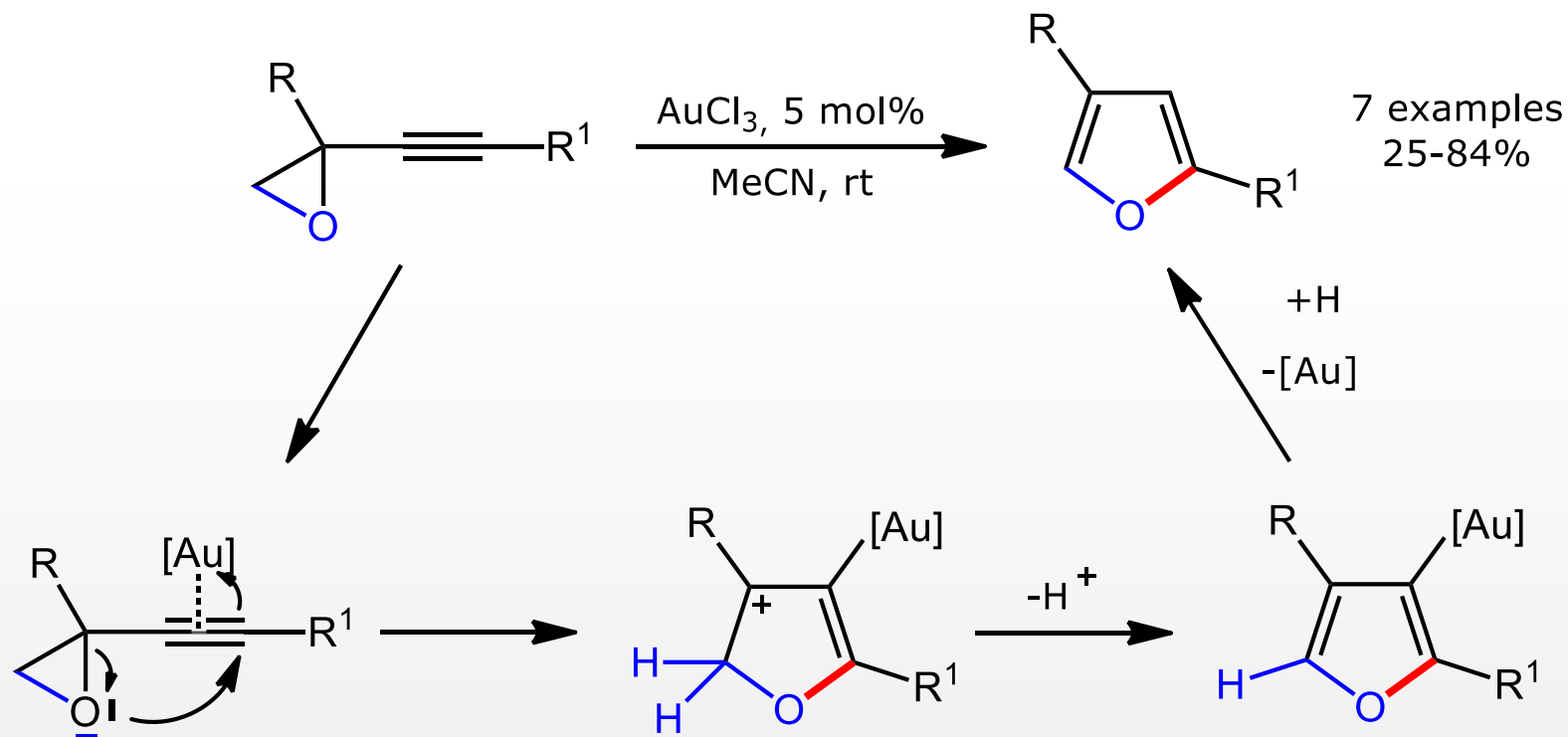
gold-catalyzed 5-exo-dig cyclization of propargylcarboxamides.



Alternative methodologies: basic conditions, Pt(II) or Hg(II) salts in stoichiometric amount at high temperature

A. S. K. Hashmi, J. P. Weyrauch, W. Frey, J. W. Bats *Org. Lett.*, **2004**, 6, 4391.

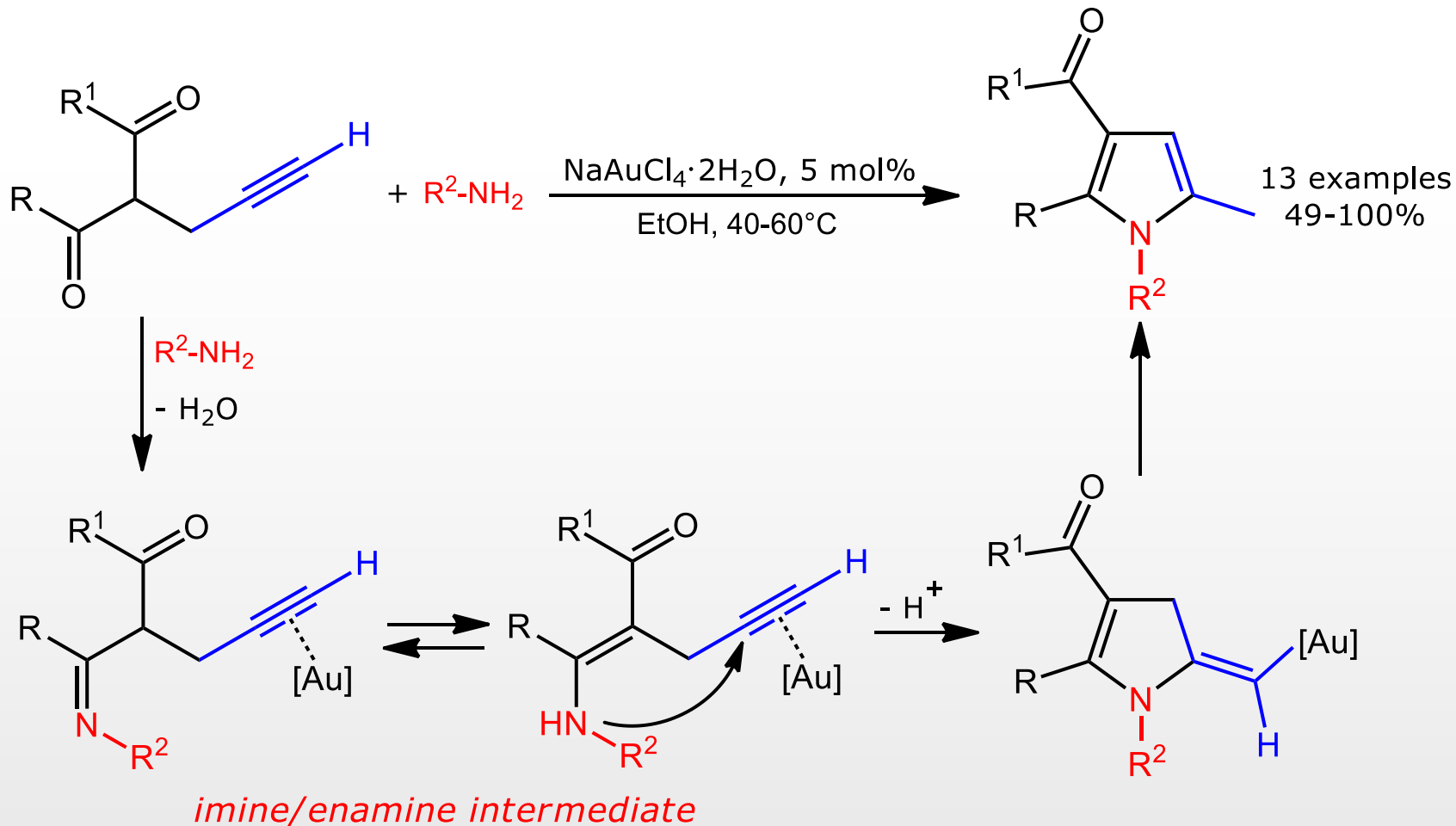
isomerization of alkynyl epoxides to furans



Alternative promoters: mercury salts *toxic*,
bases, $R = H$
molybdenum and ruthenium, $R^1 = H$

A. S. K. Hashmi, P. Sinha *Adv. Synth. Catal.* **2004**, 346, 432.

Domino addition/annulation reaction for the synthesis of pyrroles

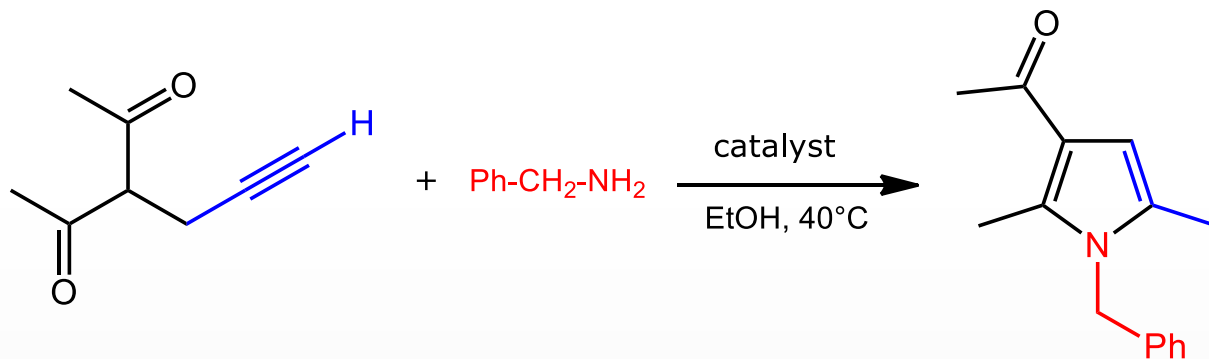


A. Arcadi, S. Di Giuseppe, F. Marinelli, E. Rossi *Adv. Synth. Catal.* **2001**, 343, 443.

A. Arcadi, S. Di Giuseppe, F. Marinelli, E. Rossi *Tetrahedron: asymmetry* **2001**, 12, 2715.

A. Arcadi, G. Abbiati, E. Rossi *J. Organomet. Chem.* **2011**, 696, 87.

Domino addition/annulation reaction for the synthesis of pyrroles



<i>catalyst (mol%)</i>	<i>time (h)</i>	<i>yield (%)</i>
NaAuCl_4 (5)	1	100
CuI (5)	24	100
ZnCl_2 (5)	24	34
Na_2PdCl_4 (5)	1	100
AgNO_3 (5)	6	71

→ Strictly anaerobic conditions

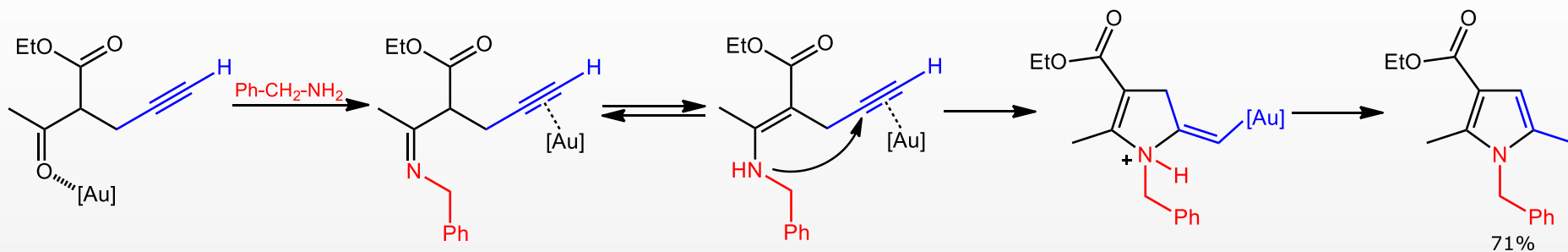
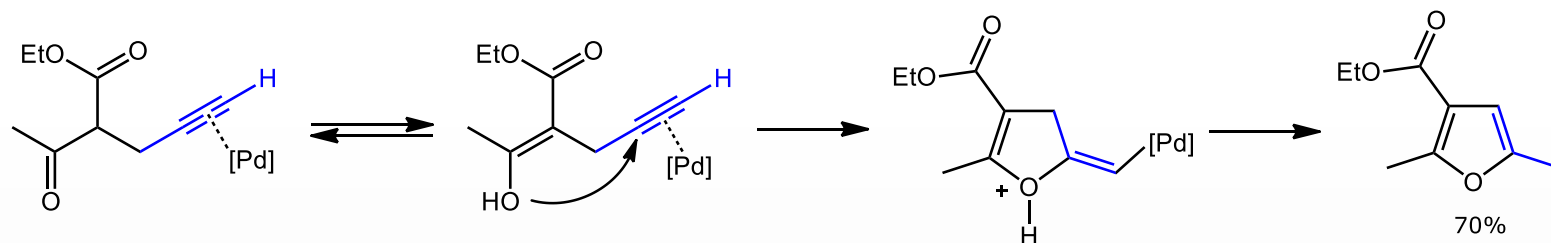
→ Poorly selective

A. Arcadi, S. Di Giuseppe, F. Marinelli, E. Rossi *Adv. Synth. Catal.* **2001**, 343, 443.

A. Arcadi, S. Di Giuseppe, F. Marinelli, E. Rossi *Tetrahedron: asymmetry* **2001**, 12, 2715.

A. Arcadi, G. Abbiati, E. Rossi *J. Organomet. Chem.* **2011**, 696, 87.

Domino addition/annulation reaction for the synthesis of pyrroles



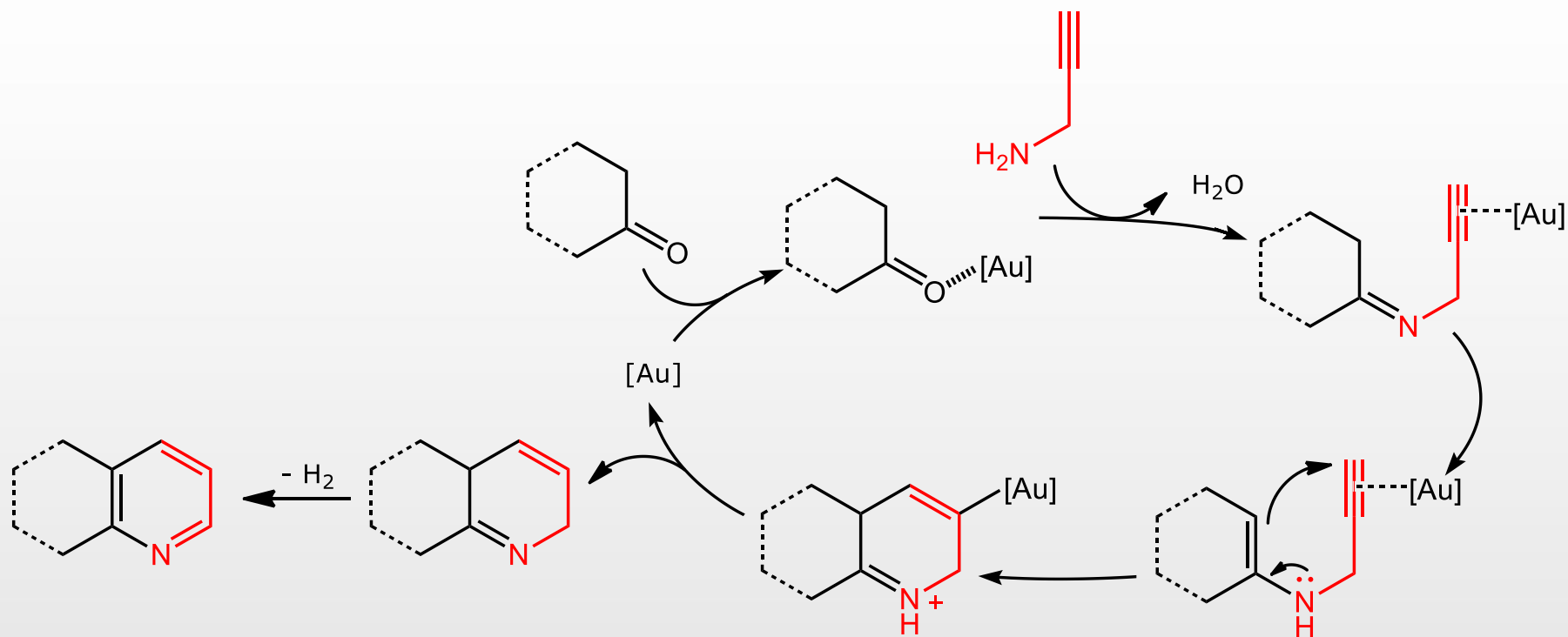
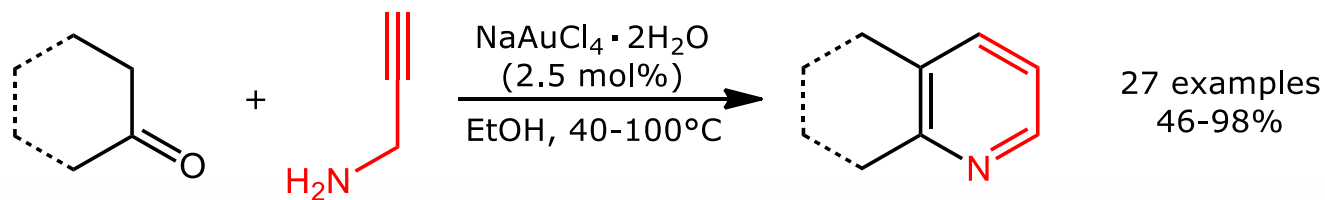
*Reaction performed with β -ketoesters
selectivity could be related to the σ -philic property of gold(III) salts*

A. Arcadi, S. Di Giuseppe, F. Marinelli, E. Rossi *Adv. Synth. Catal.* **2001**, 343, 443.

A. Arcadi, S. Di Giuseppe, F. Marinelli, E. Rossi *Tetrahedron: asymmetry* **2001**, 12, 2715.

A. Arcadi, G. Abbiati, E. Rossi *J. Organomet. Chem.* **2011**, 696, 87.

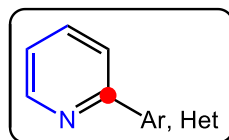
Domino addition/annulation reaction for the synthesis of pyridines



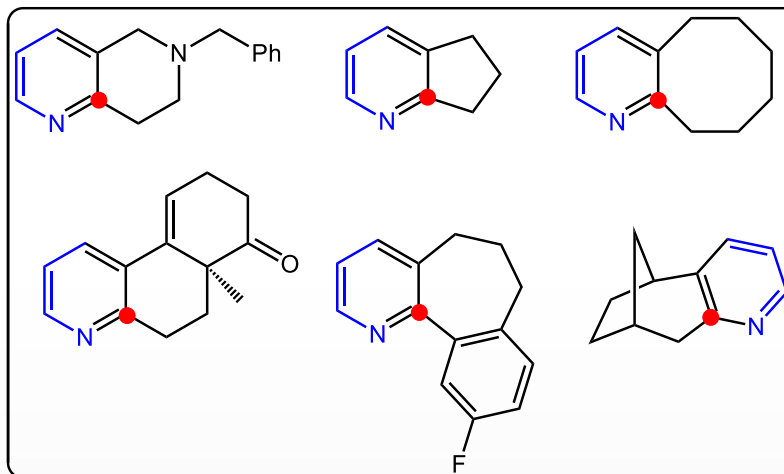
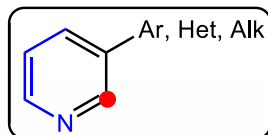
G. Abbiati, A. Arcadi, G. Bianchi, S. Di Giuseppe, F. Marinelli, E. Rossi *J. Org. Chem.* **2003**, 68, 6959.

Nucleophilic addition to π activated carbon-carbon triple bonds, synthesis of heterocycles

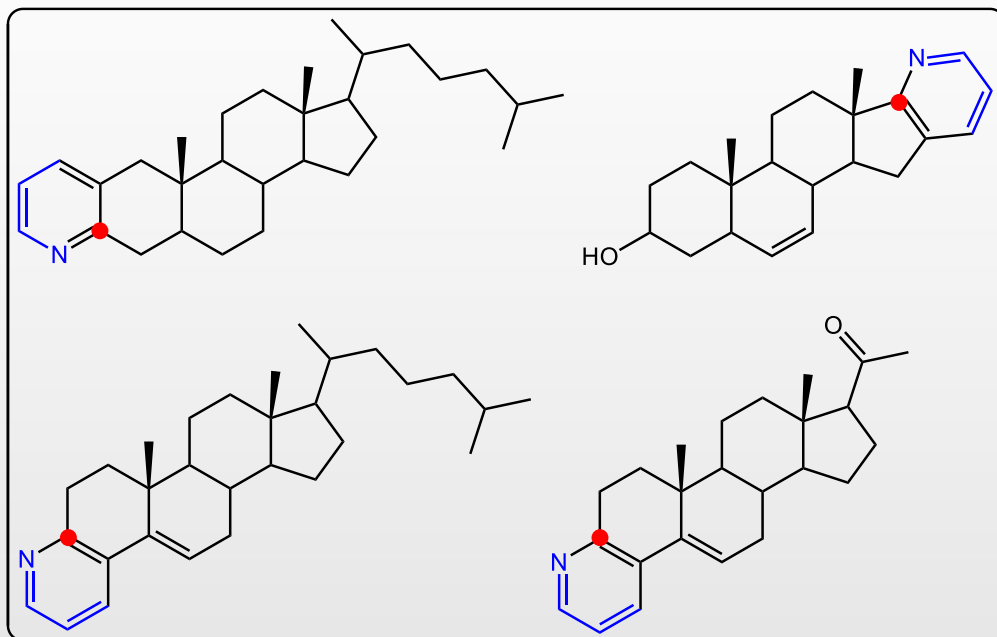
from methyl ketones



from aldehydes



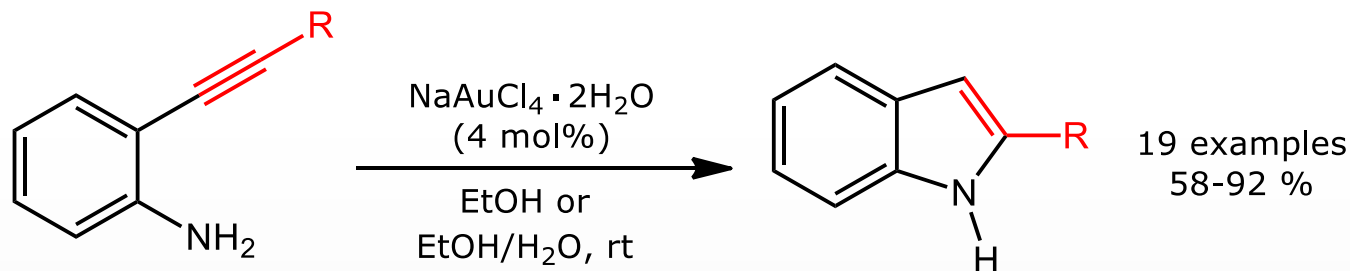
from cyclic ketones



linear or angular steroidal derivatives

G. Abbiati, A. Arcadi, G. Bianchi, S. Di Giuseppe, F. Marinelli, E. Rossi *J. Org. Chem.* **2003**, 68, 6959.

annulation of 2-alkynylanilines

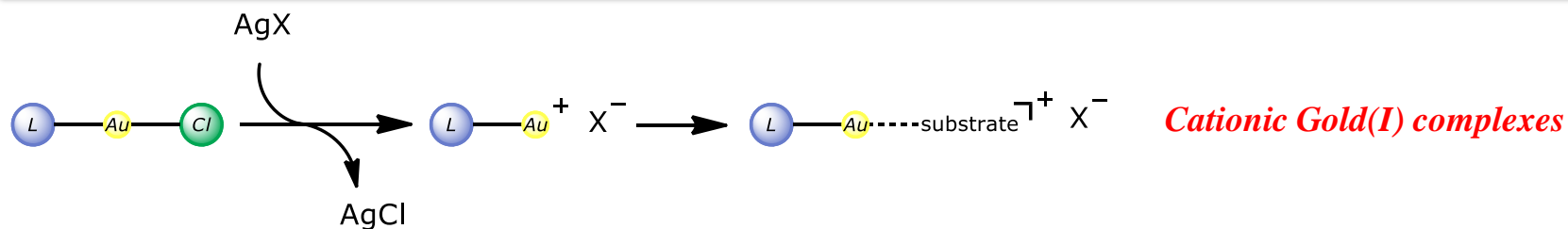


$R = Ph$

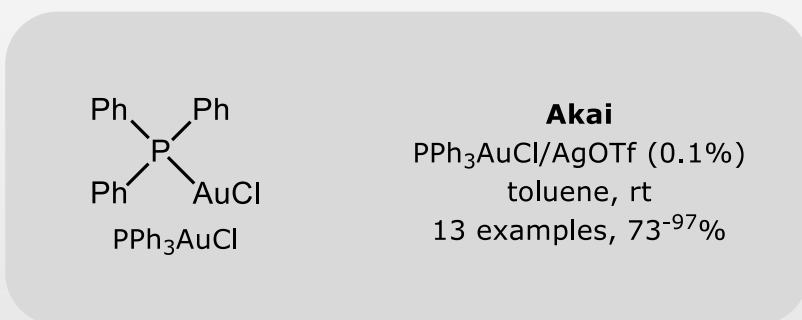
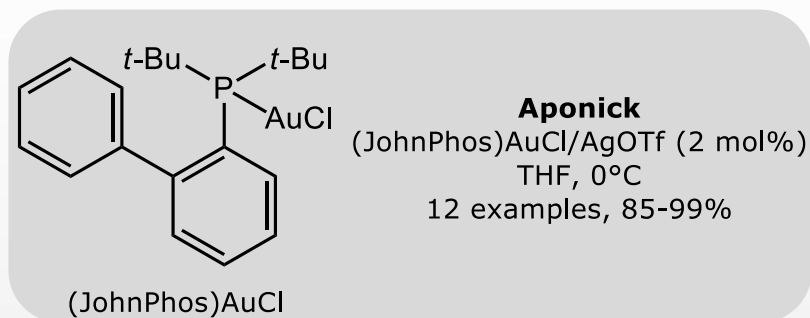
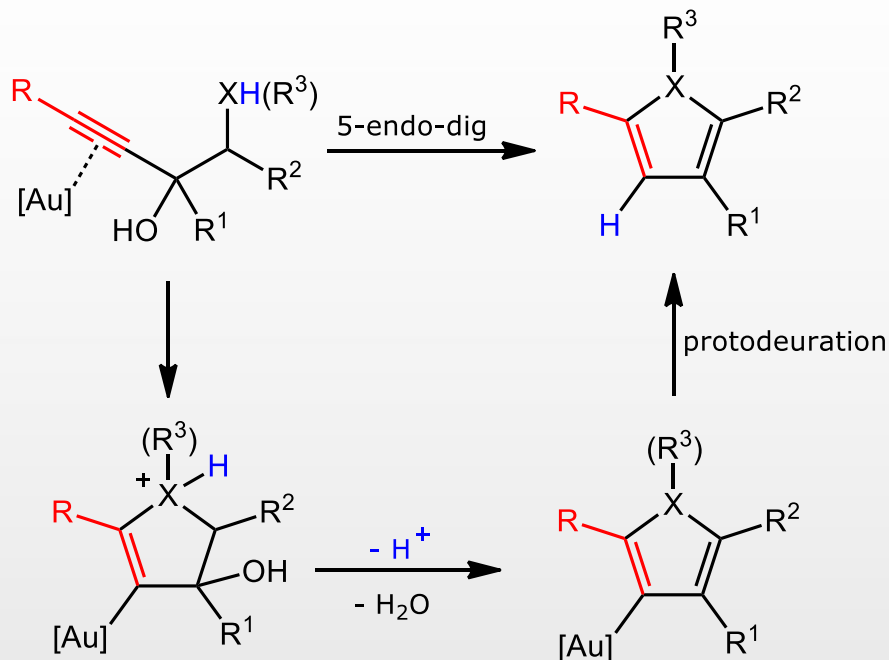
<i>catalyst (mol%)</i>	<i>solvent</i>	<i>yield (%)</i>
NaAuCl ₄ (4)	EtOH	83
AuCl (4)	EtOH	50
Na ₂ PdCl ₄ (5)	EtOH	7
PdCl ₂ (5)	EtOH	6
Pd(OAc) ₂ (5)	EtOH	8
Cu(OTf) ₂ (5)	EtOH	10
Cu(OAc) ₂ (5)	EtOH	0

A. Arcadi, G. Bianchi, F. Marinelli *Synthesis* **2004**, 610.

Nucleophilic addition to π activated carbon-carbon triple bonds, synthesis of heterocycles

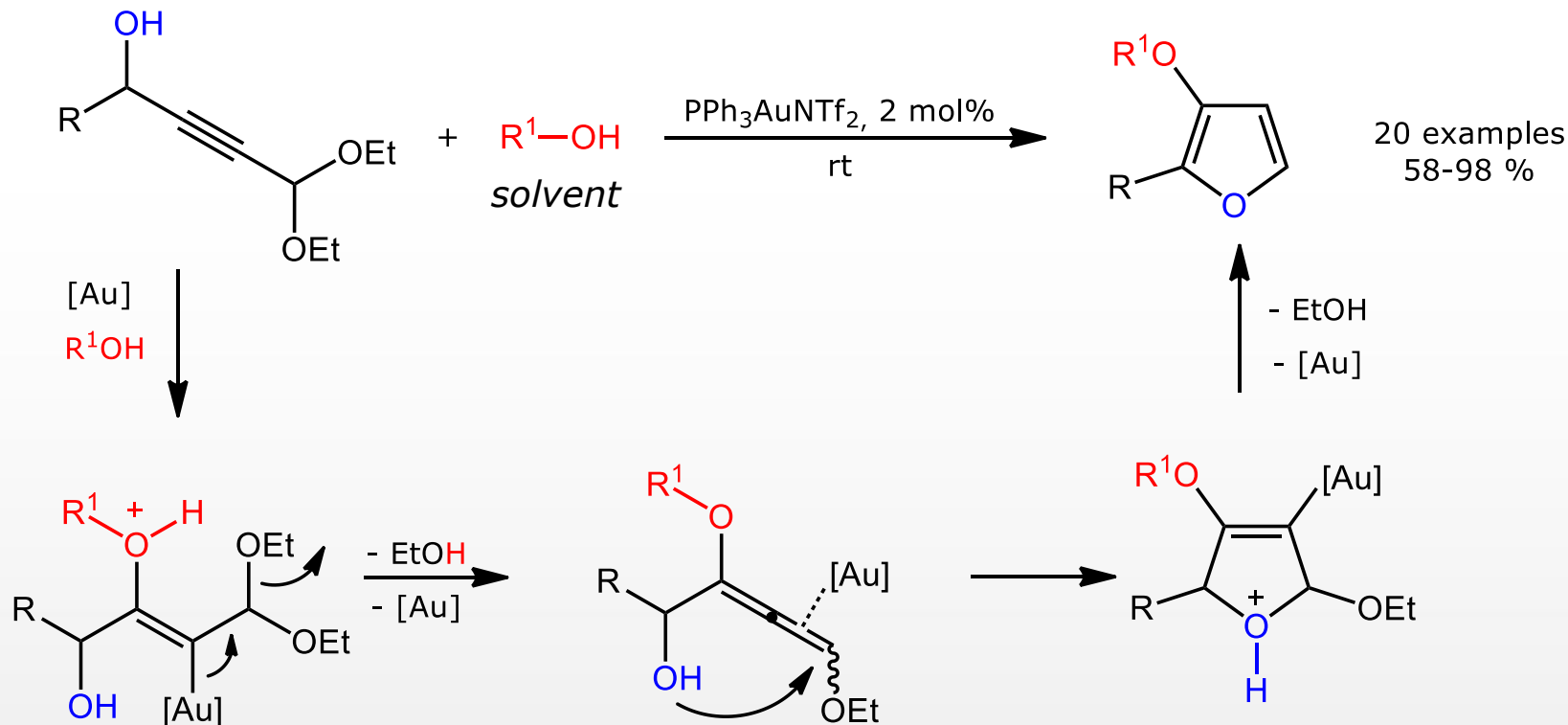


Synthesis of pyrroles, furans and thiophenes from 2-hydroxyhomopropargylic alcohols, amines or sulphides



A. Aponick, C.-Y. Li, J. Malinge, E. F. Marques *Org. Lett.* **2009**, *11*, 4624.
M. Egi, K. Azechi, S. Akai *Org. Lett.* **2009**, *11*, 5002.

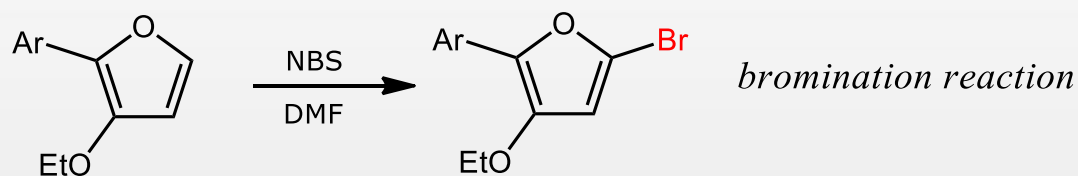
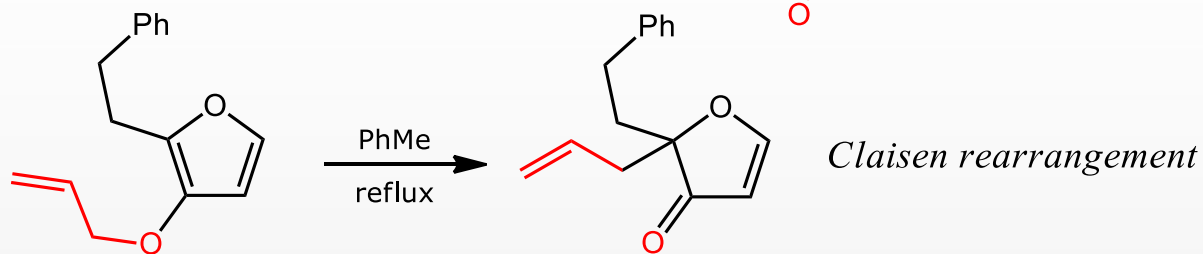
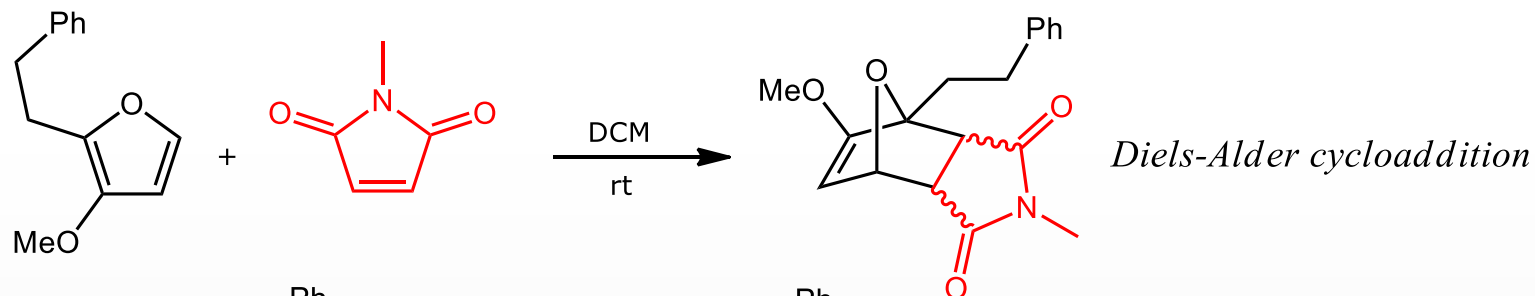
Synthesis of 3-alkoxyfurans from propargylic alcohols containing an acetal moiety



It is worth to underline that firm guidelines are not yet available for the optimum choice of ligands and counterions used in these processes, and so a certain degree of screening for the best reaction conditions (ligands, counterions, solvent, temperature) may be necessary.

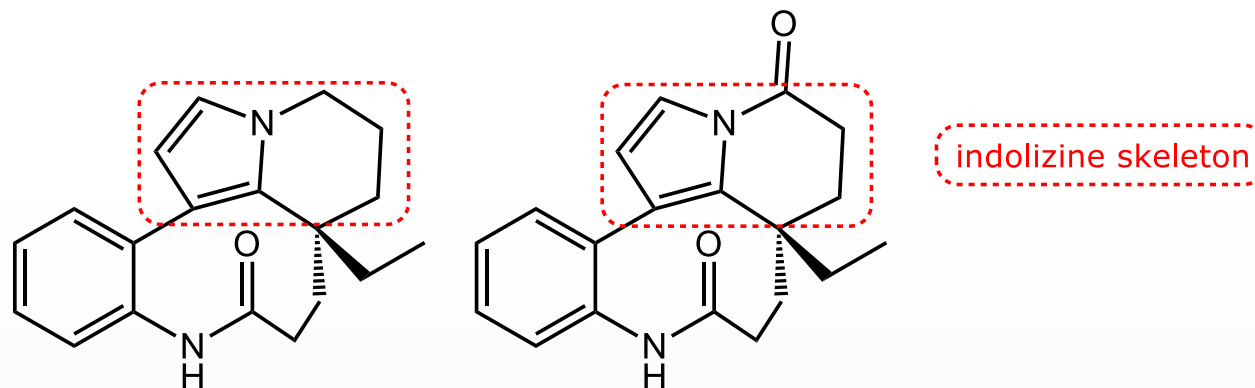
M. N. Pennell, R. W. Foster, P. G. Turner, H. C. Hailes, C. J. Tame, T. D. Sheppard *Chem. Commun.* **2014**, 50, 1302.

Nucleophilic addition to π activated carbon-carbon triple bonds, synthesis of heterocycles

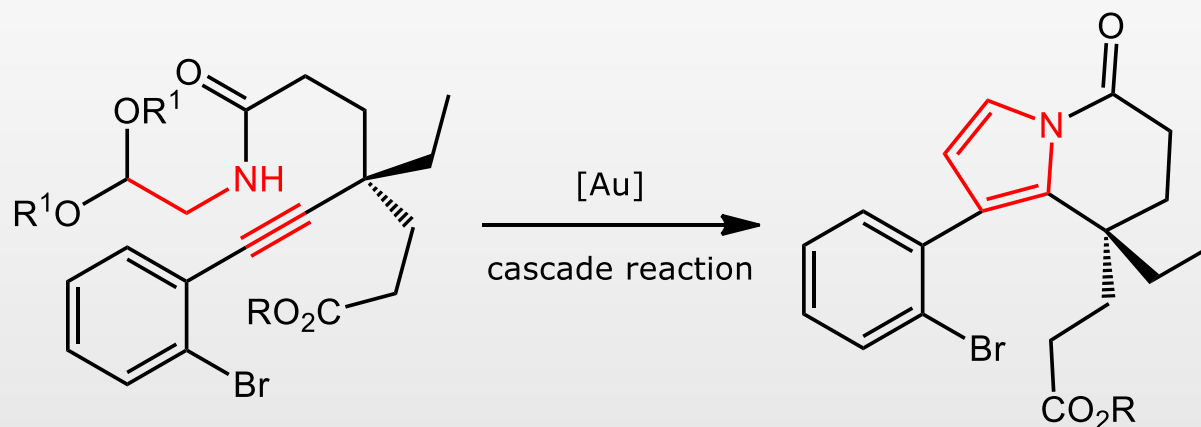


M. N. Pennell, R. W. Foster, P. G. Turner, H. C. Hailes, C. J. Tame, T. D. Sheppard *Chem. Commun.* **2014**, 50, 1302.

total syntheses of (-)-Rhazinilam and (-)-Rhazinicine



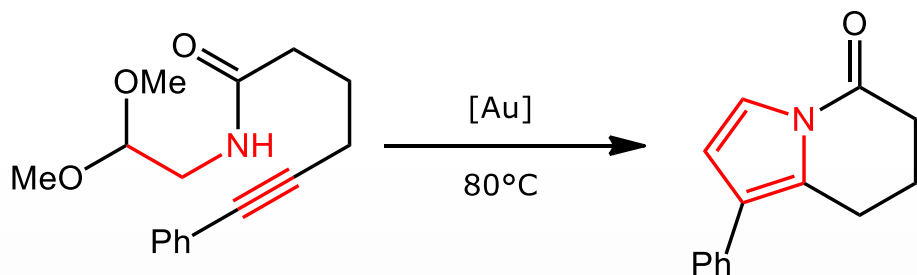
construction of indolizine skeleton from linear intermediates



K. Sugimoto, K. Toyoshima, S. Nonaka, K. Kotaki, H. Ueda, H. Tokuyama *Angew. Chem. Int. Ed.* **2013**, *52*, 7168.
H. Ueda, M. Yamaguchi, H. Kameya, K. Sugimoto, H. Tokuyama *Org. Lett.* **2014**, *16*, 4948.

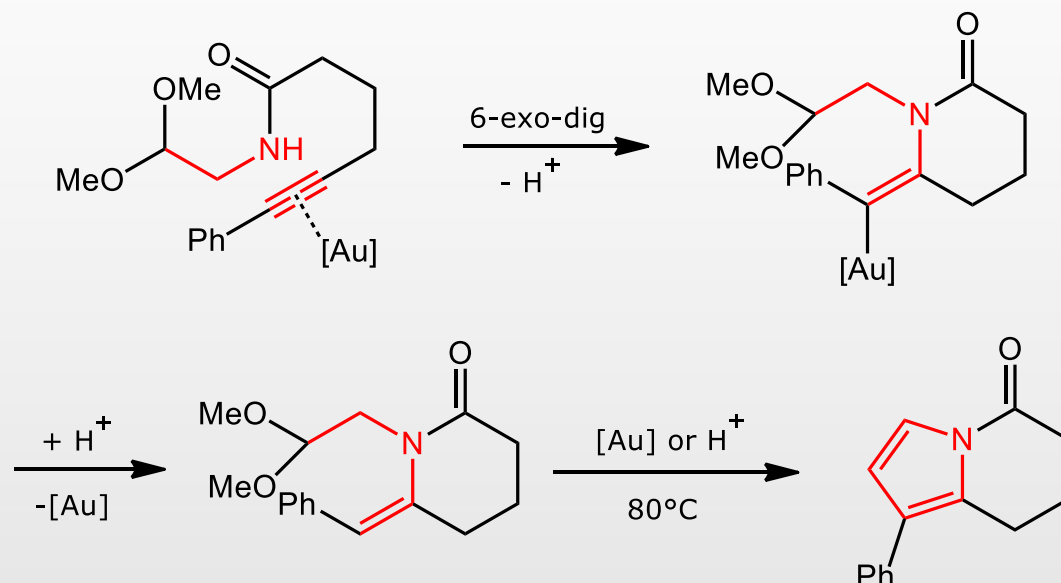
Nucleophilic addition to π activated carbon-carbon triple bonds, synthesis of heterocycles

model reaction



catalyst (mol%)	solvent	yield (%)
AuCl (10)	DCE	-
AuCl ₃ (10)	DCE	-
Au(PPh ₃)Cl (10)	DCE	-
AuCl/AgOTf (10)	DCE	-
Au(PPh ₃)Cl/AgOTf (10)	DCE	20
Au(PPh ₃)Cl/AgNTf ₂ (10)	DCE	20
Au(PPh ₃)NTf ₂ (10)	DCE	50
Au(PPh ₃)NTf ₂ (10)	1,4-dioxane	69

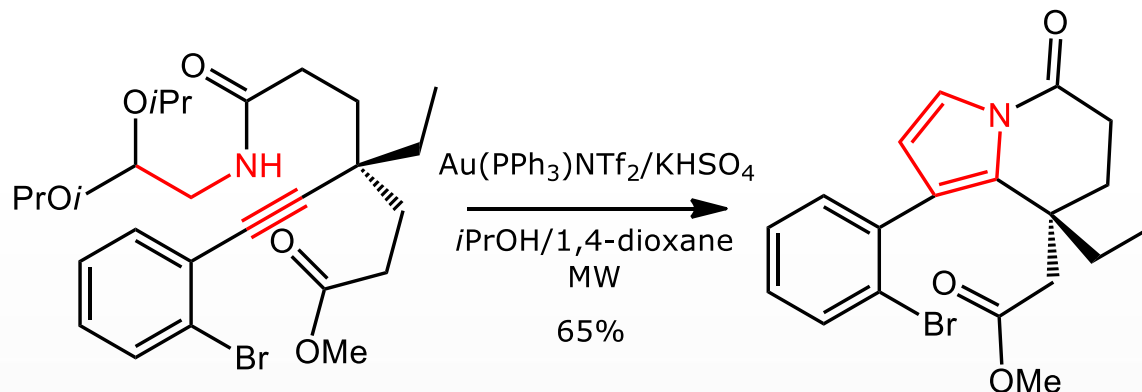
proposed reaction mechanism



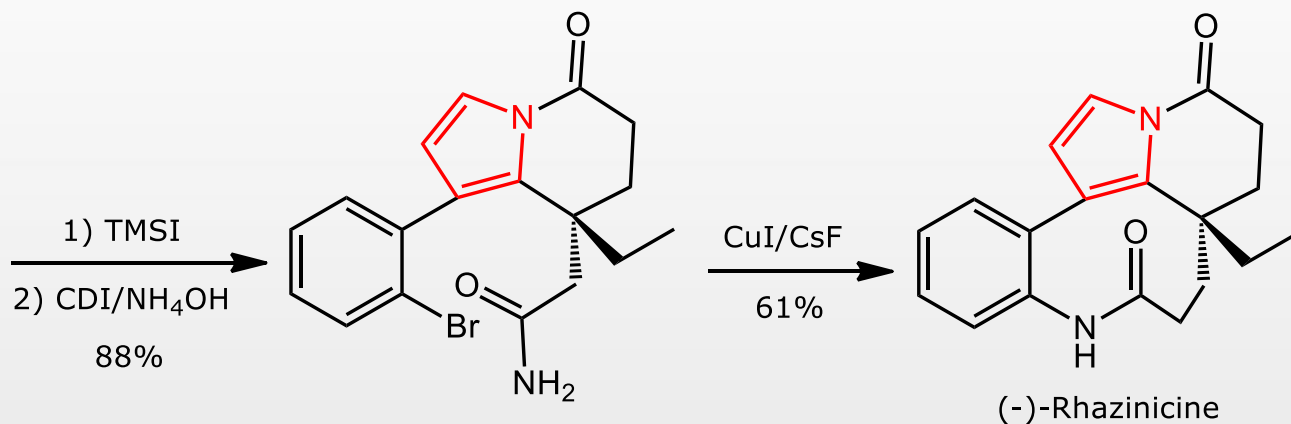
K. Sugimoto, K. Toyoshima, S. Nonaka, K. Kotaki, H. Ueda, H. Tokuyama *Angew. Chem. Int. Ed.* **2013**, *52*, 7168.

H. Ueda, M. Yamaguchi, H. Kameya, K. Sugimoto, H. Tokuyama *Org. Lett.* **2014**, *16*, 4948.

application of the selected conditions to the synthesis of (-)-Rhazinicine



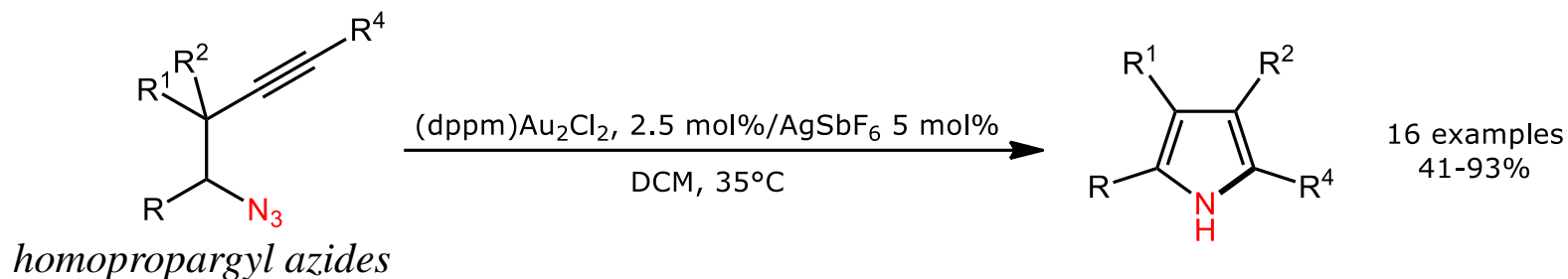
formation of indolizine skeleton



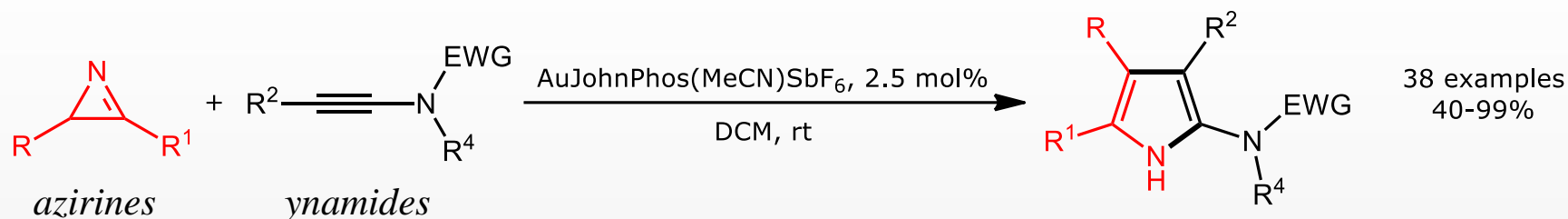
formation of the lactam ring

K. Sugimoto, K. Toyoshima, S. Nonaka, K. Kotaki, H. Ueda, H. Tokuyama *Angew. Chem. Int. Ed.* **2013**, *52*, 7168.
H. Ueda, M. Yamaguchi, H. Kameya, K. Sugimoto, H. Tokuyama *Org. Lett.* **2014**, *16*, 4948.

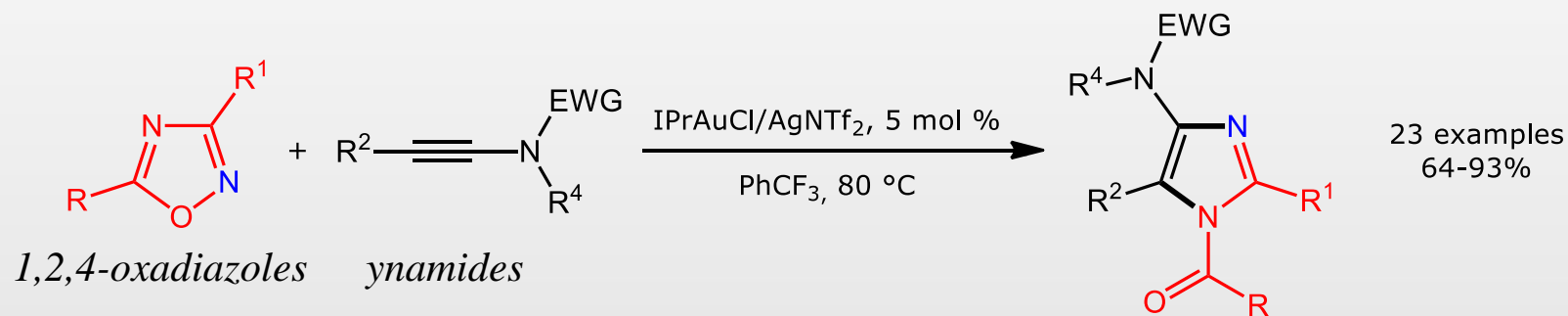
Nucleophilic addition to π activated carbon-carbon triple bonds, synthesis of heterocycles



D. J. Gorin, N. R. Davis, F. D. Toste *J. Am. Chem. Soc.* **2005**, *127*, 11260.

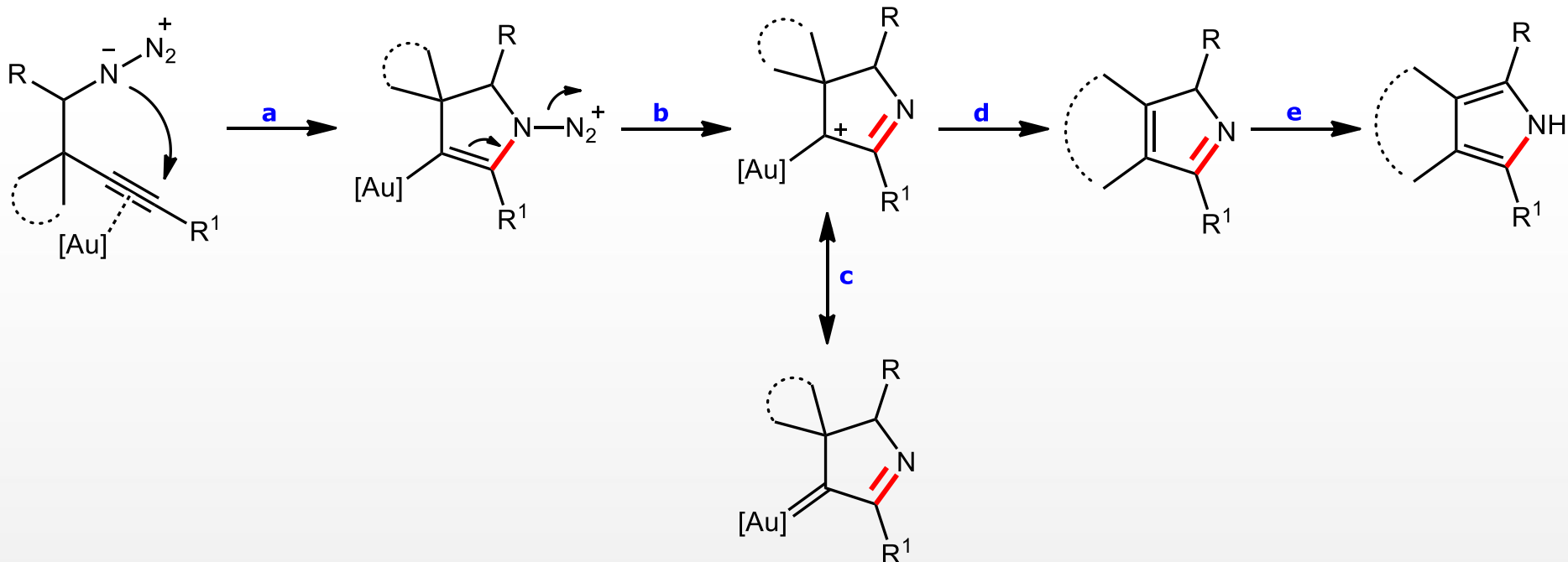


L. Zhu, Y. Yu, Z. Mao, X. Huang *Org. Lett.* **2015**, *17*, 30.



Z. Zeng, H. Jin, J. Xie, B. Tian, M. Rudolph, F. Rominger, A. S. K. Hashmi *Org. Lett.* **2017**, *19*, 1020.

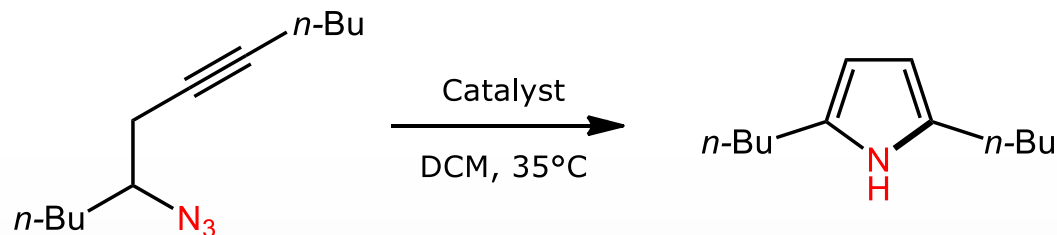
*intramolecular Au(I)-catalyzed Schmidt reaction
reaction mechanism*



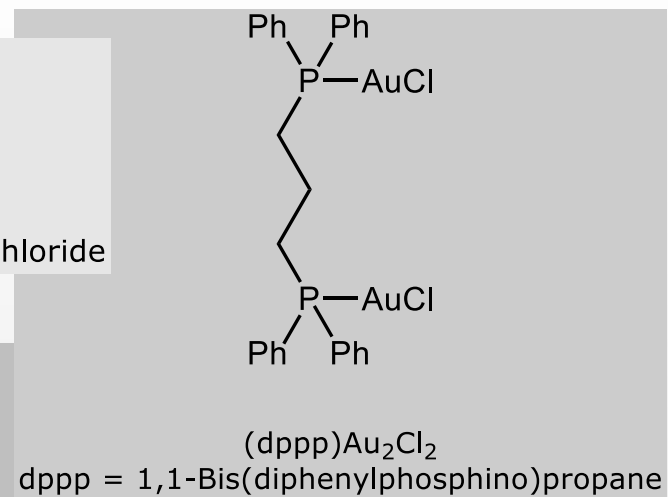
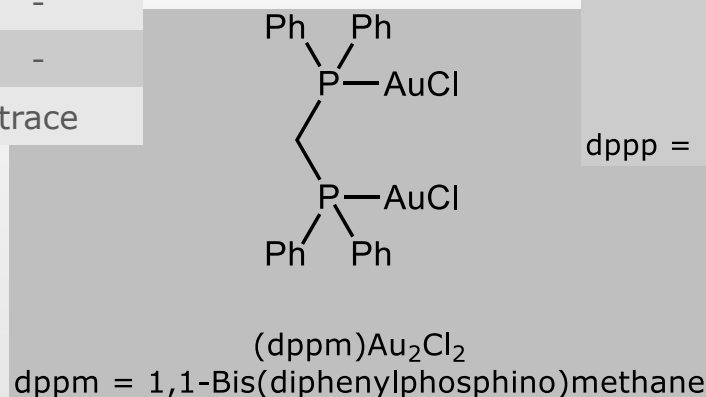
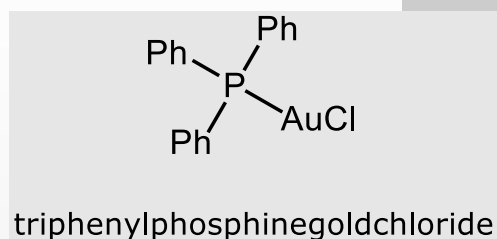
- a) attack of the proximal N-atom of the azide on gold(I) activated triple bond,
 b) loss of nitrogen, production of a cationic intermediate,
 c) stabilization by electron donation from gold(I),
 d) formal 1,2-shift and regeneration of the cationic gold(I) catalyst and produces a 2H-pyrrole,
 e) tautomerization to the 1H-pyrrole.

D. J. Gorin, N. R. Davis, F. D. Toste *J. Am. Chem. Soc.* **2005**, *127*, 11260.

intramolecular Au(I)-catalyzed Schmidt reaction
reaction conditions

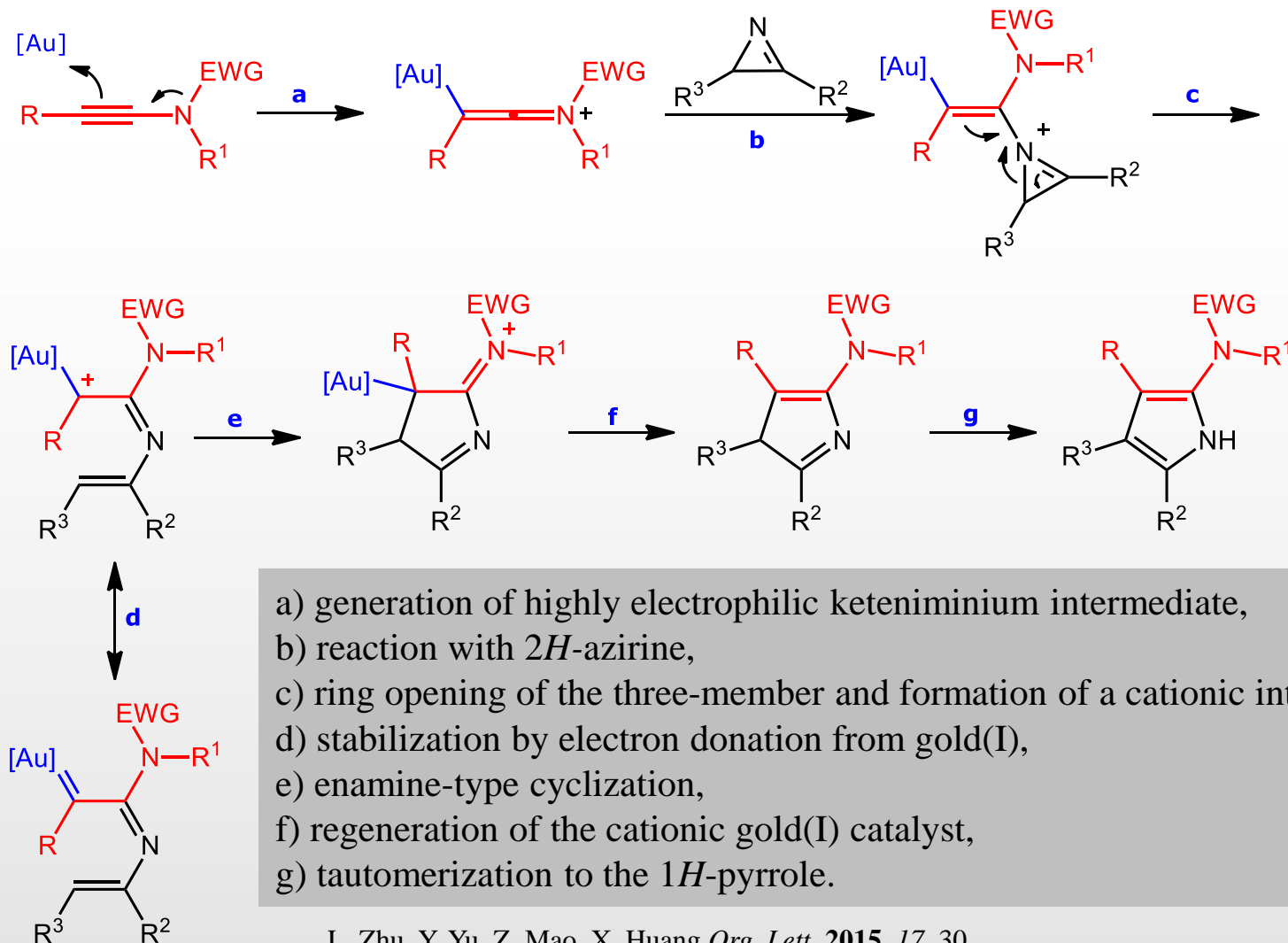


<i>catalyst (mol%)</i>	<i>yield (%)</i>
Au(PPh ₃)Cl (5), AgSbF ₆ (5)	72
(dppm)Au ₂ Cl ₂ (2.5), AgSbF ₆ (5)	93
(dppp)Au ₂ Cl ₂ (2.5), AgSbF ₆ (5)	86
AuCl ₃ (5)	-
CuI (5)	-
AgSbF ₆ (5)	trace



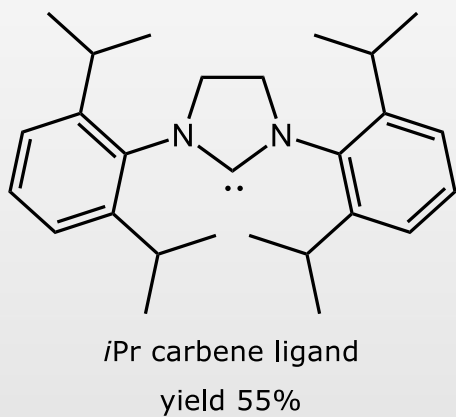
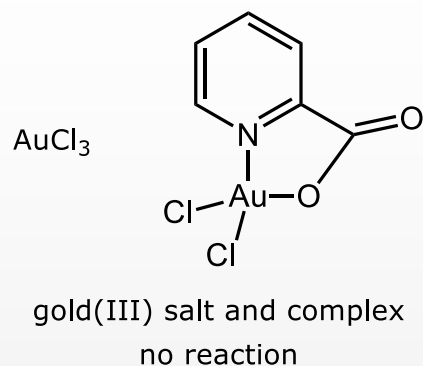
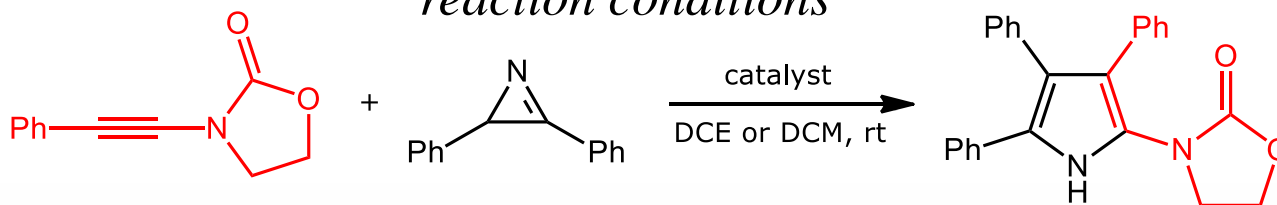
D. J. Gorin, N. R. Davis, F. D. Toste *J. Am. Chem. Soc.* **2005**, *127*, 11260.

substituted pyrroles in via formal [3 + 2] cycloaddition
reaction mechanism

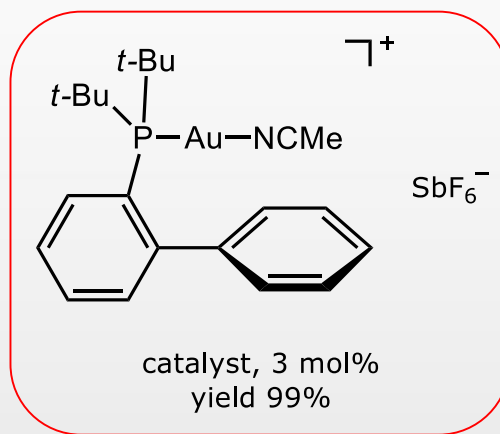


L. Zhu, Y. Yu, Z. Mao, X. Huang *Org. Lett.* **2015**, *17*, 30.

*substituted pyrroles in via formal [3 + 2] cycloaddition
reaction conditions*



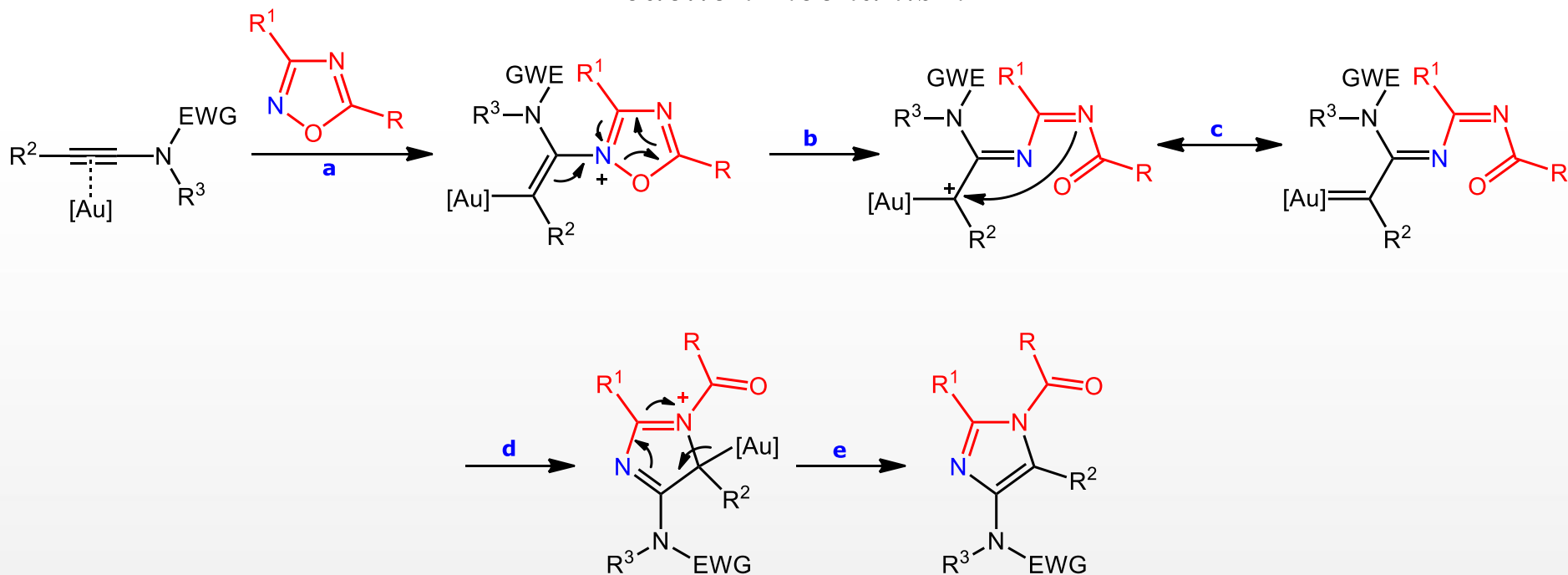
*electrophilicity of the corresponding gold(I) complex
with chloride as counterion*



optimized reaction conditions

L. Zhu, Y. Yu, Z. Mao, X. Huang *Org. Lett.* **2015**, *17*, 30.

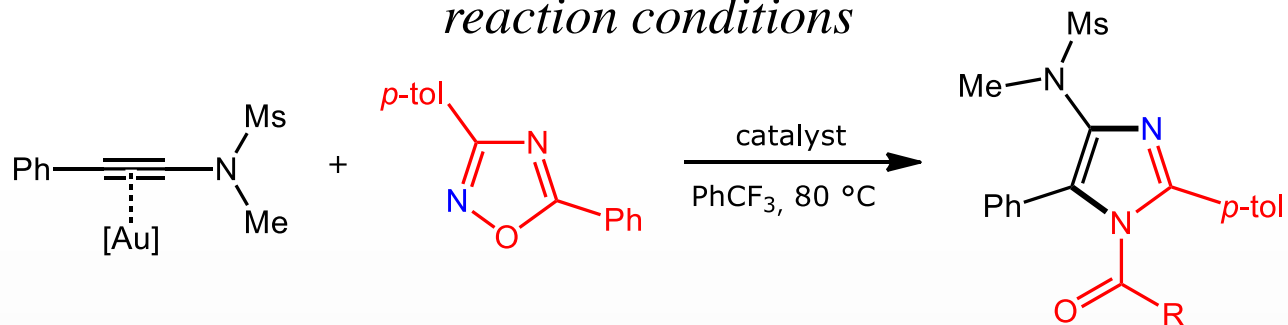
4-aminoimidazoles in via formal [3 + 2] cycloaddition
reaction mechanism



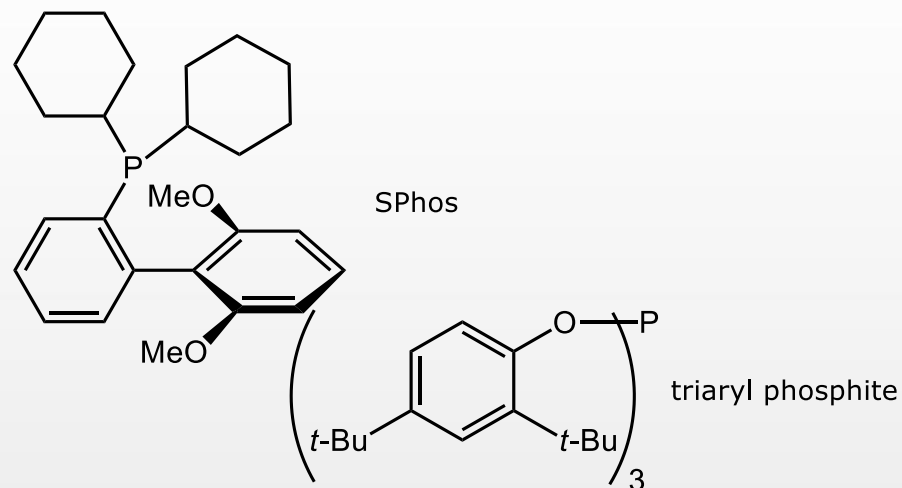
a) regioselective addition to gold-activated ynamide,
 b) ring opening of the oxadiazole, formation of a cationic intermediate,
 c) stabilization by electron donation from gold(I),
 d) imine-type cyclization,
 e) regeneration of the catalyst.

Z. Zeng, H. Jin, J. Xie, B. Tian, M. Rudolph, F. Rominger, A. S. K. Hashmi *Org. Lett.* **2017**, *19*, 1020.

4-aminoimidazoles in via formal [3 + 2] cycloaddition
reaction conditions



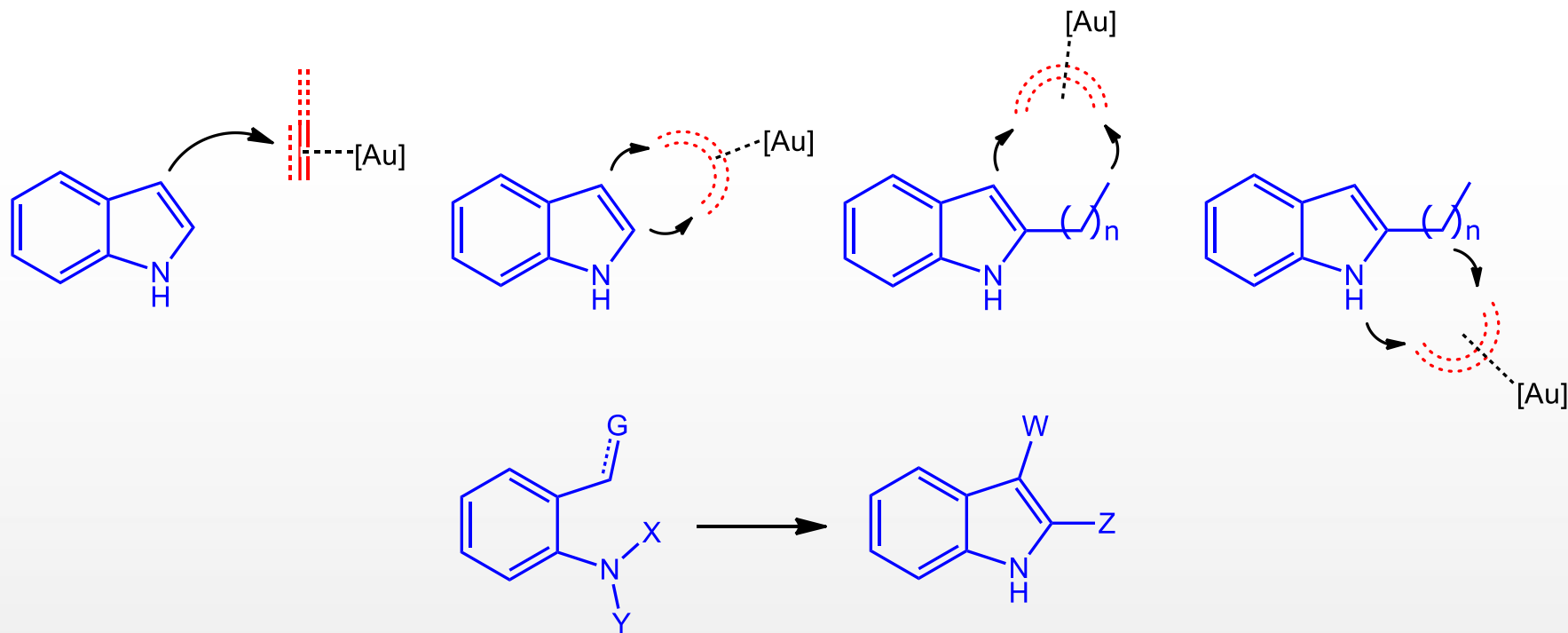
<i>catalyst (mol%)</i>	<i>imidazole (%)</i>
IPrAuCl/AgNTf ₂ (5)	95
IPrAuCl/AgOTf (5)	67
PPh ₃ AuNTf ₂ (5)	52
SPhosAuNTf ₂ (5)	48
KAuBr ₄	39
(2,4- <i>t</i> Bu ₂ PhO) ₃ PAuCl/AgNTf ₂ (5)	30



*electrophilicity of the corresponding
gold(I) complex
with chloride as counterion*

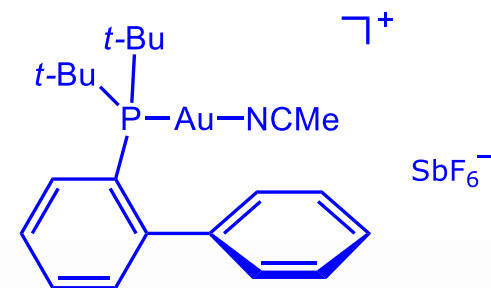
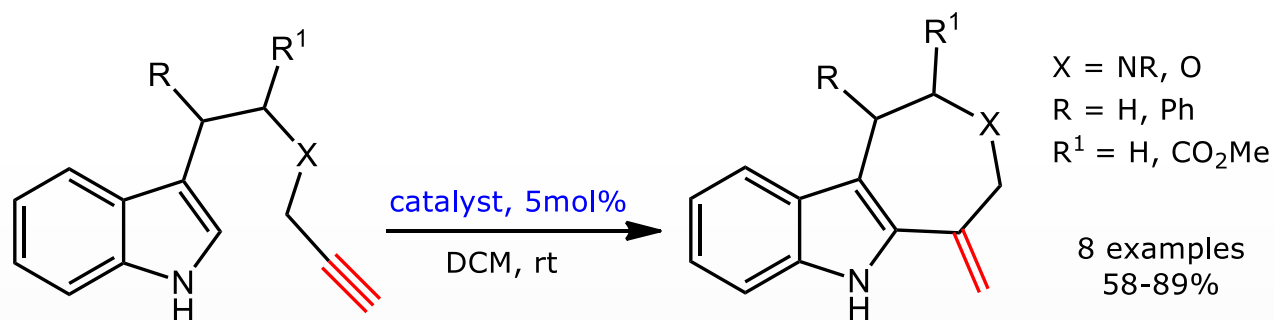
Z. Zeng, H. Jin, J. Xie, B. Tian, M. Rudolph, F. Rominger, A. S. K. Hashmi *Org. Lett.* **2017**, *19*, 1020.

indoles as nucleophilic partners in reactions with activated π -systems



alkynes and allenes in functionalization, cyclization and cycloaddition reactions

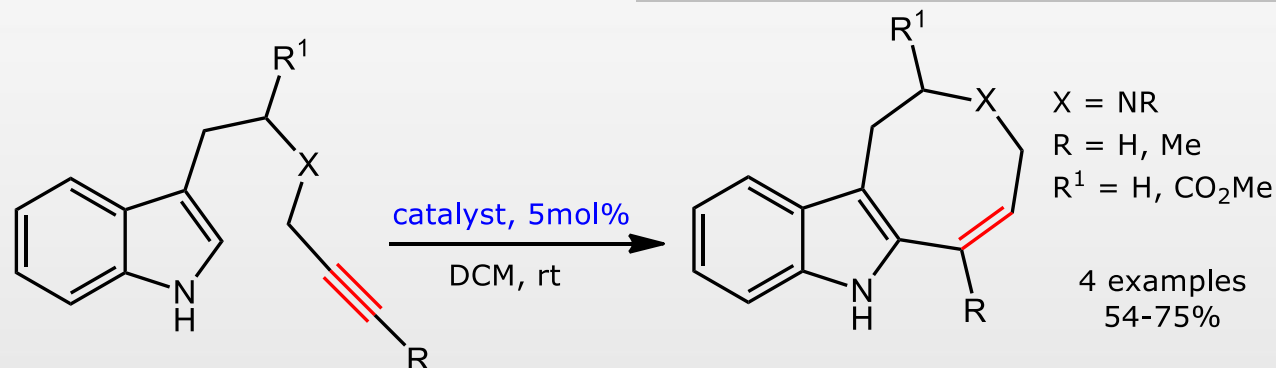
intramolecular hydroarylations of alkyne tethered indoles



same substrate
different catalyst!

formal 7-exo-dig cyclization

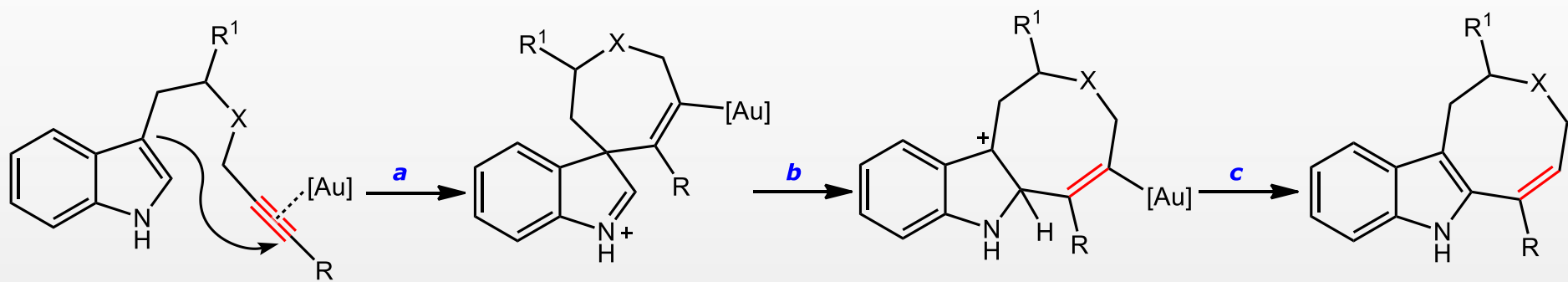
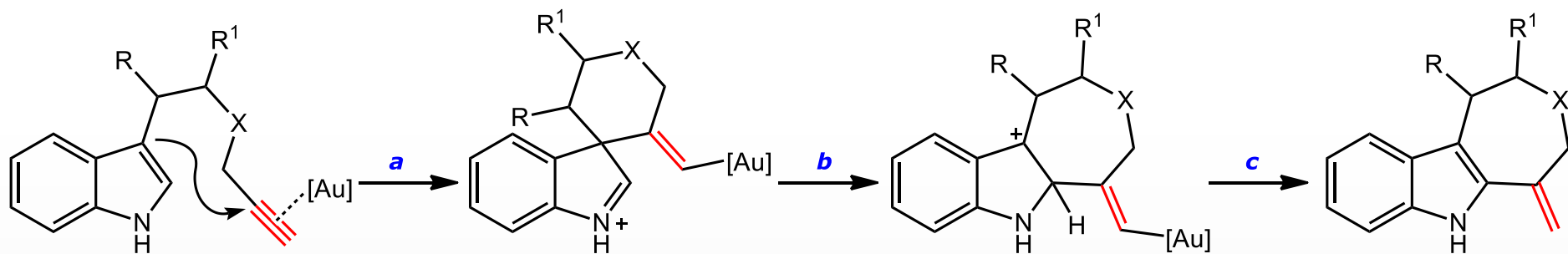
formal 8-endo-dig cyclization



AuCl_3

C. Ferrer, A. M. Echavarren *Angew. Chem., Int. Ed.* **2006**, *45*, 1105.
 C. Ferrer, C. H. M. Amijs, A. M. Echavarren *Chem. Eur. J.* **2007**, *13*, 1358.

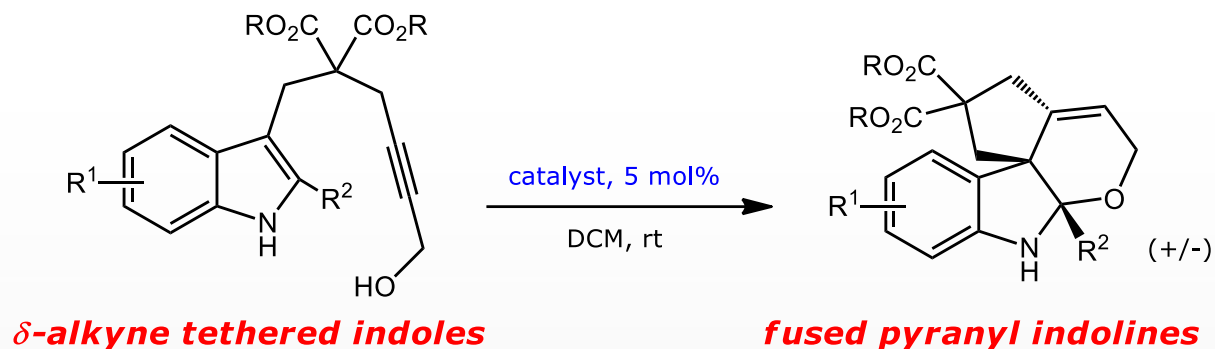
intramolecular hydroarylations of alkyne tethered indoles



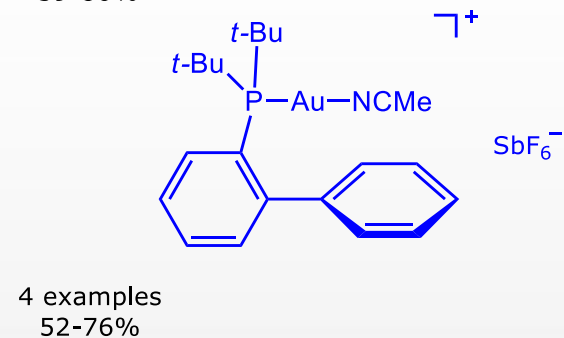
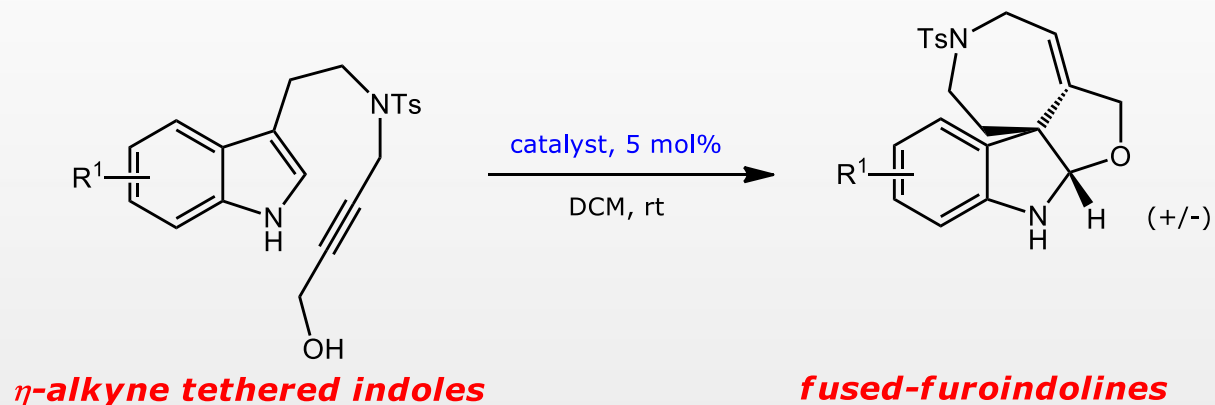
- a) intramolecular hydroarylation
- b) 1,2 migration
- c) aromatization, protodeauration

C. Ferrer, A. M. Echavarren *Angew. Chem., Int. Ed.* **2006**, *45*, 1105.
 C. Ferrer, C. H. M. Amijs, A. M. Echavarren *Chem. Eur. J.* **2007**, *13*, 1358.

Cascade reactions of δ - and η -alkyne tethered indoles



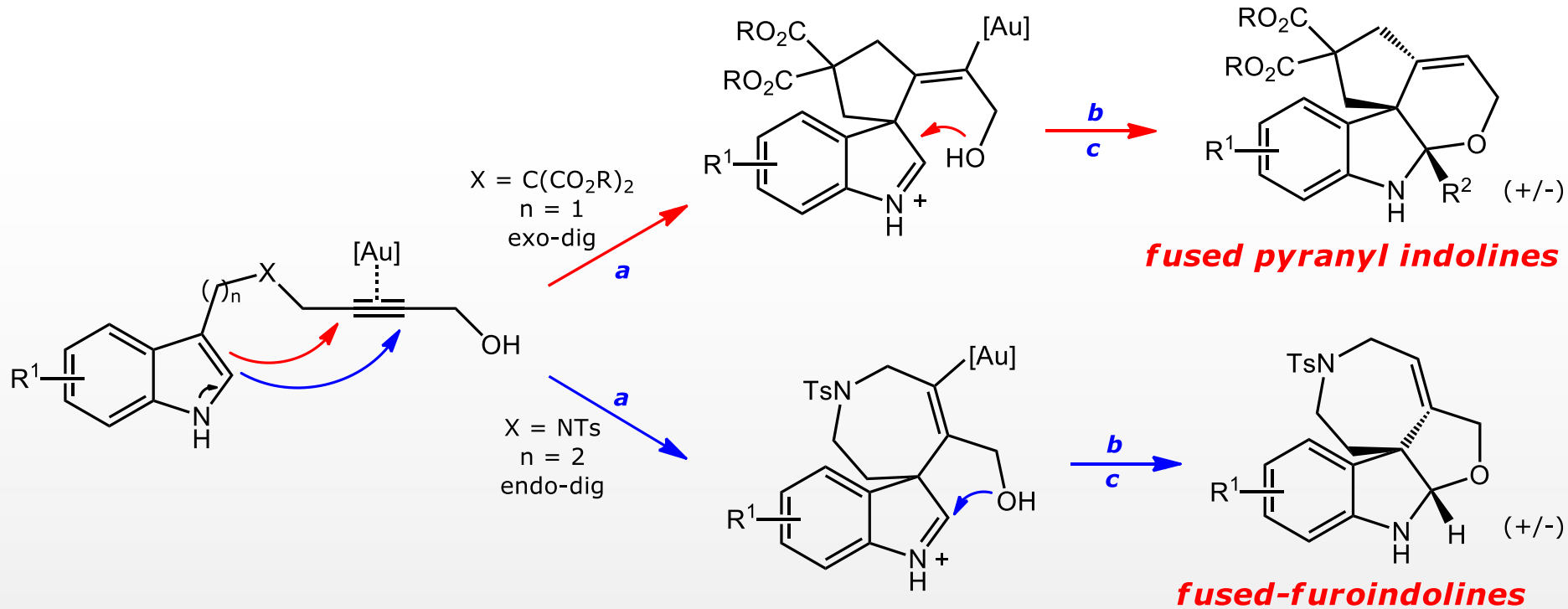
10 examples
59-86%



G. Cera, P. Crispino, M. Monari, M. Bandini *Chem. Commun.* **2011**, 47, 7803.

G. Cera, P. Crispino, M. Chiarucci, A. Mazzani, M. Mancinelli, M. Bandini *Org. Lett.* **2012**, 14, 1350.

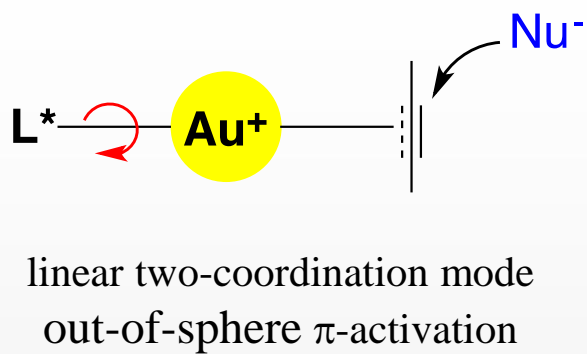
Cascade reactions of alkyne tethered indoles
reaction mechanism



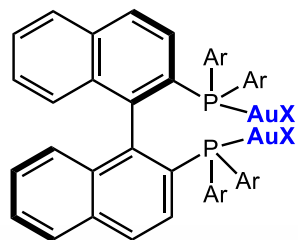
- Intramolecular hydroarylation
- trapping of the iminium group by the hydroxyl group
- protodeauration

G. Cera, P. Crispino, M. Monari, M. Bandini *Chem. Commun.* **2011**, 47, 7803.

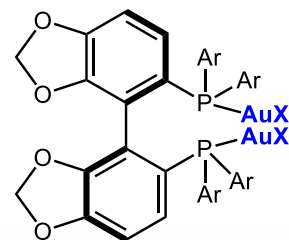
G. Cera, P. Crispino, M. Chiarucci, A. Mazzani, M. Mancinelli, M. Bandini *Org. Lett.* **2012**, 14, 1350.



Bidentate catalyst

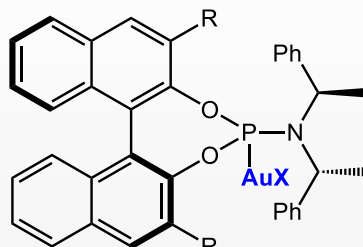


BINAP phosphines

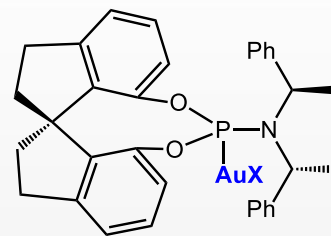


SEGPHOS phosphines

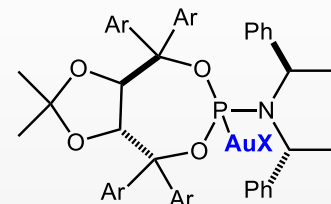
Monodentate catalyst



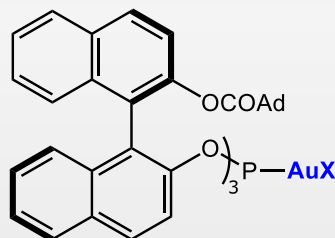
BINOL phosphoramidites



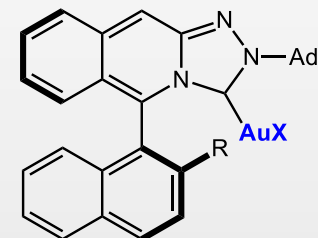
SPIINOL phosphoramidites



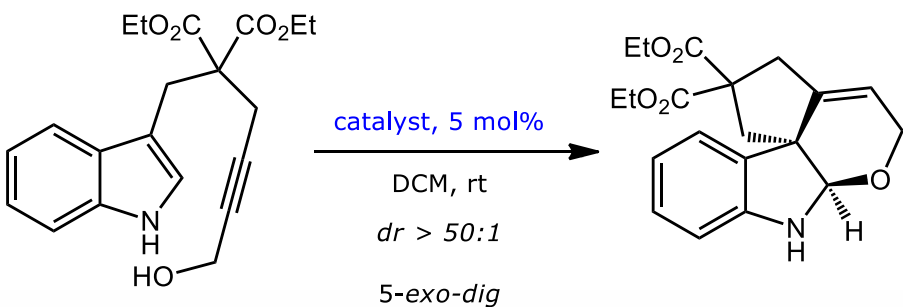
TADDOL phosphoramidites



BINOL phosphites

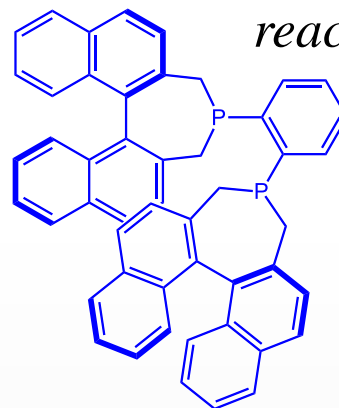


Chiral NHCs

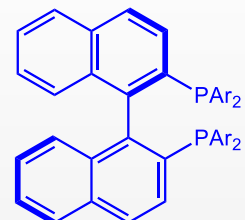


Catalyst (5 mol%)	AgX	Yield (%)	ee (%)
(R,R)-L1	AgSbF ₆	21	46 (+)
(S)-L2	AgSbF ₆	26	20 (-)
(R)-L3	AgSbF₆	65	56 (-)
(S)-L4	AgSbF ₆	36	14 (-)
(R)-L5	AgSbF ₆	61	0
(S)-L6	AgSbF ₆	65	17 (-)
(R)-L3	AgOTf	59	72 (-)
(R)-L3	AgNTf₂	80	27 (-)
(R)-L3	AgBF₄	60	74 (-)
(R)-L3	AgPF₆	70	70 (-)
(R)-L3 0°C + 4A MS	AgBF₄	89	86 (-)

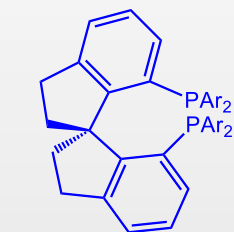
enantioselective cascade reaction reaction conditions



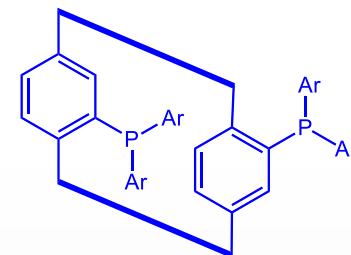
(R,R)-L1: binaphane



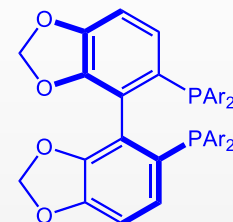
(R)-L3: Ar = 3,5-Me₂-Ph



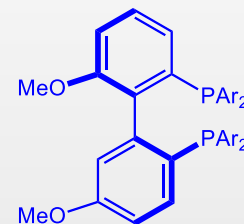
(R)-L5: Ar = 3,5-Me₂-Ph



(S)-L2: Ar = 3,5-Me₂-Ph



(S)-L4: Ar = 3,5-(tBu)₂-4-OMe-Ph



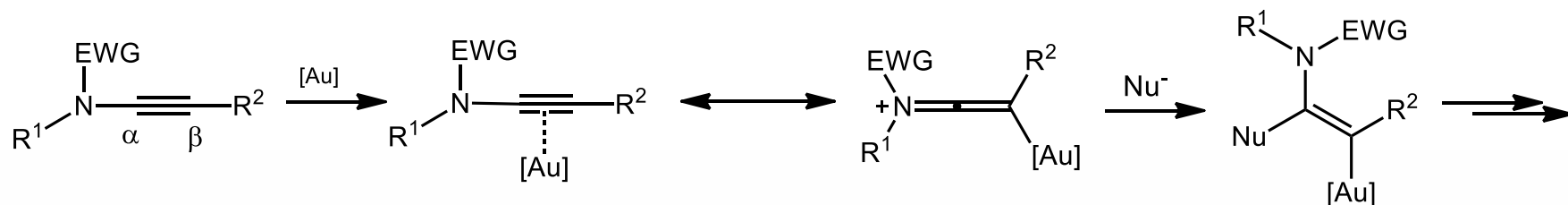
(S)-L6: Ar = 3,5-(tBu)₂-4-OMe-Ph

9 examples
50-89%, ee 75-87%

G. Cera, P. Crispino, M. Monari, M. Bandini *Chem. Commun.* **2011**, 47, 7803.

G. Cera, P. Crispino, M. Chiarucci, A. Mazzani, M. Mancinelli, M. Bandini *Org. Lett.* **2012**, 14, 1350.

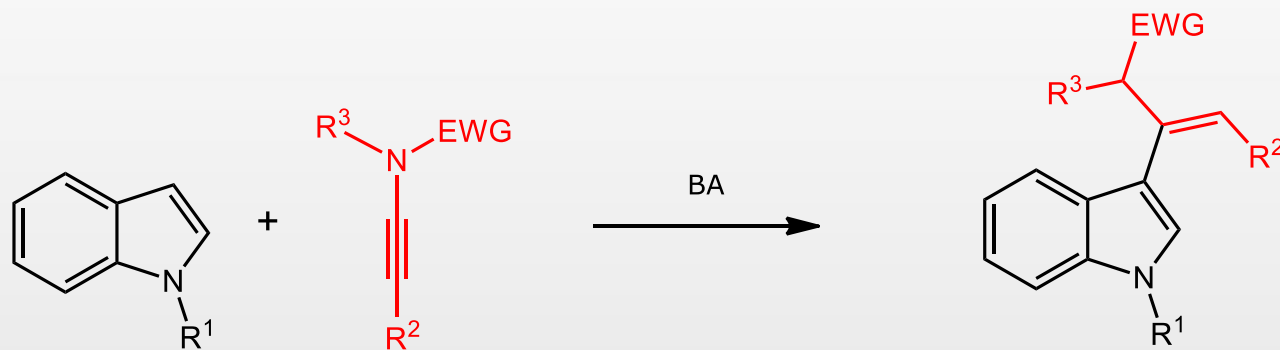
ynamides and gold catalysis



S. Nayak, B. Prabagar, A. K. Sahoo *Org. Biomol. Chem.* **2016**, *14*, 803.

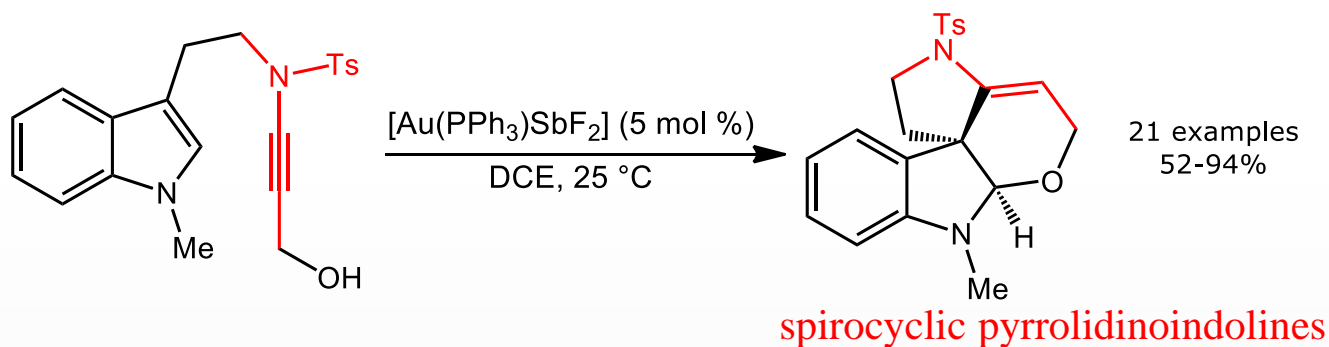
C. Theunissen, M. Lecomte *Aldrichimica Acta* **2015**, *48*, 59.

reaction of indoles with ynamides under Broensted acid catalysis

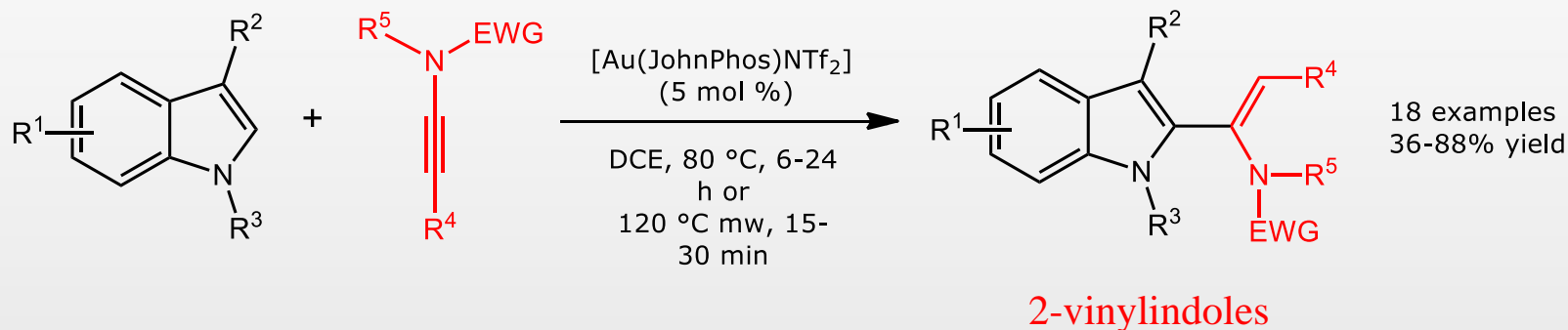


Y. Zhang *Tetrahedron* **2006**, *62*, 3917.

intra- and intermolecular reactions of ynamides and indoles under gold catalysis

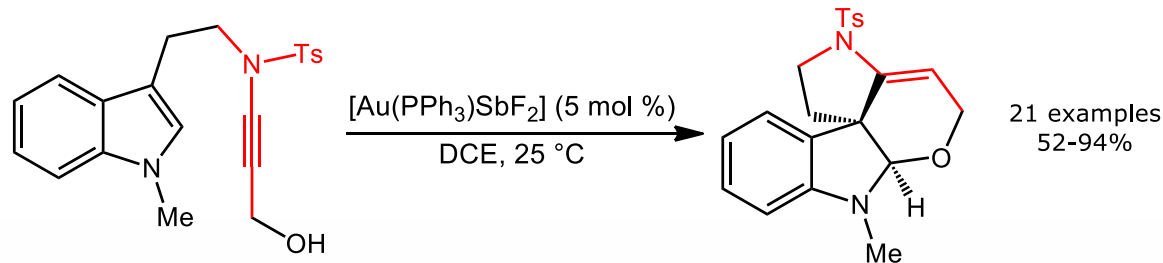


N. Zheng, Y.-Y. Chang, L.-J. Zhang, J.-X. Gong, Z. Yang *Chem. Asian J.* **2016**, *11*, 371.



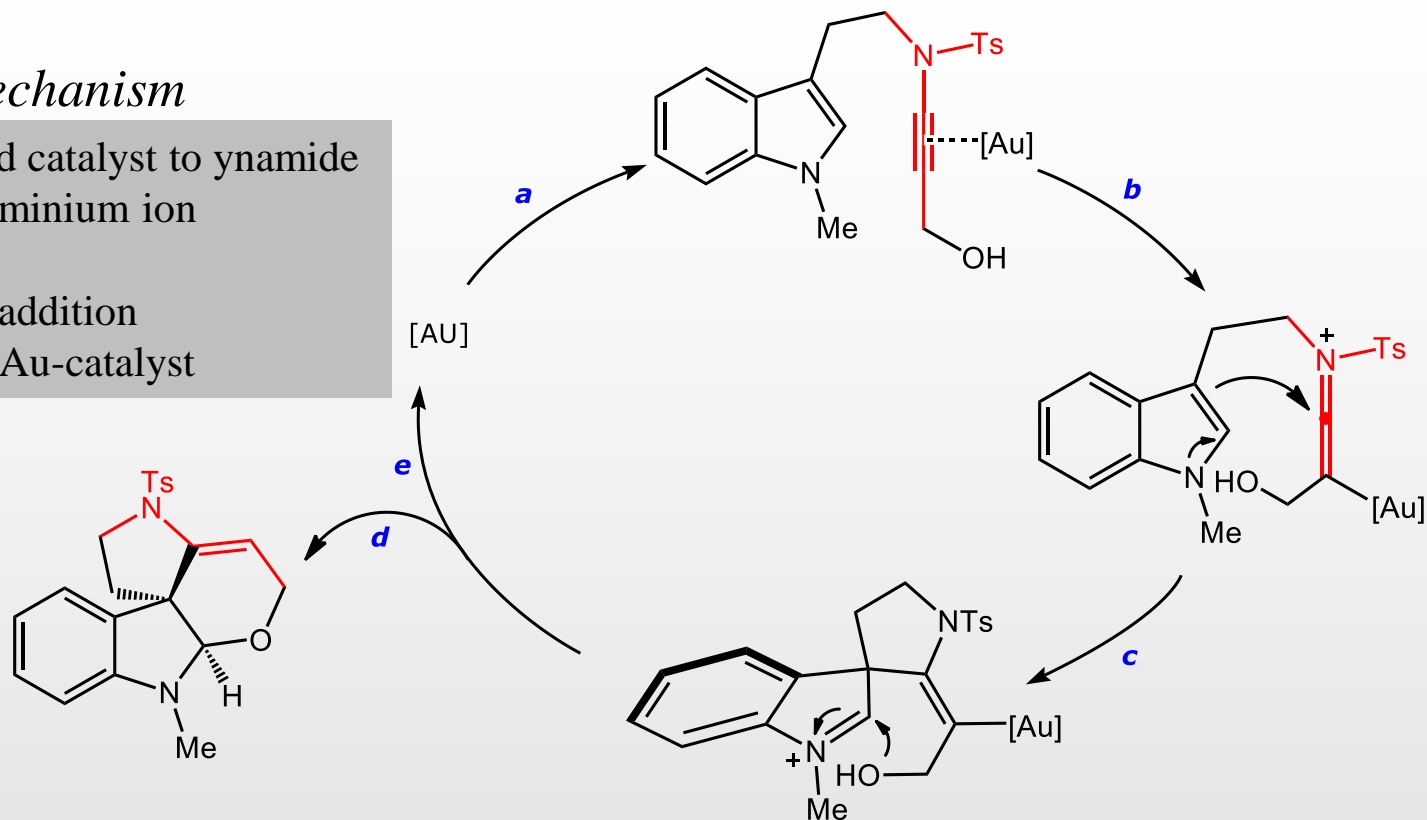
V. Pirovano, M. Negrato, G. Abbiati, M. Dell'Acqua, E. Rossi *Org. Lett.* **2016**, *18*, 4798.

synthesis of spirocyclic pyrrolidinoindolines



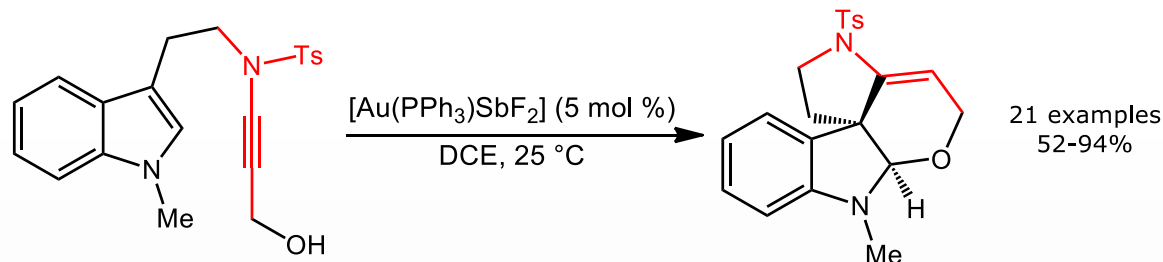
reaction mechanism

- coordination of gold catalyst to ynamide
- formation of keteniminium ion
- 5-endo cyclization
- 6-exo nucleophilic addition
- regeneration of the Au-catalyst



N. Zheng, Y.-Y. Chang, L.-J. Zhang, J.-X. Gong, Z. Yang *Chem. Asian J.* **2016**, *11*, 371.

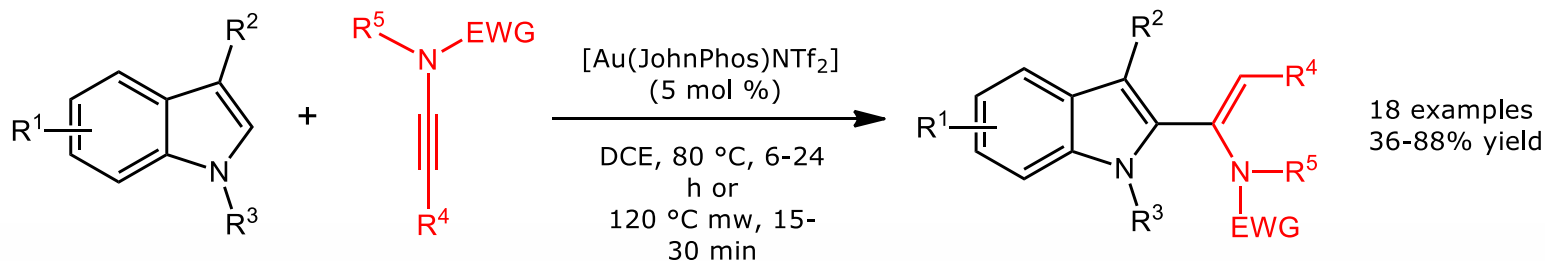
synthesis of spirocyclic pyrrolidinoindolines
reaction conditions



<i>catalyst (mol%)</i>	<i>Yield (%)</i>
TsOH (5)	-
TfOH (5)	13
Tf ₂ OH (5)	14
AgOTf (5)	-
AuCl ₃ (5)	-
PtCl ₂ (5)	-
iPrAuCl (5), AgNTf ₂ (5)	60
iPrAuCl (5), AgSbF ₆ (5)	76
PPh ₃ AuCl (5), AgNTf ₂ (5)	77
PPh ₃ AuCl (5), AgSbF ₆ (5)	94

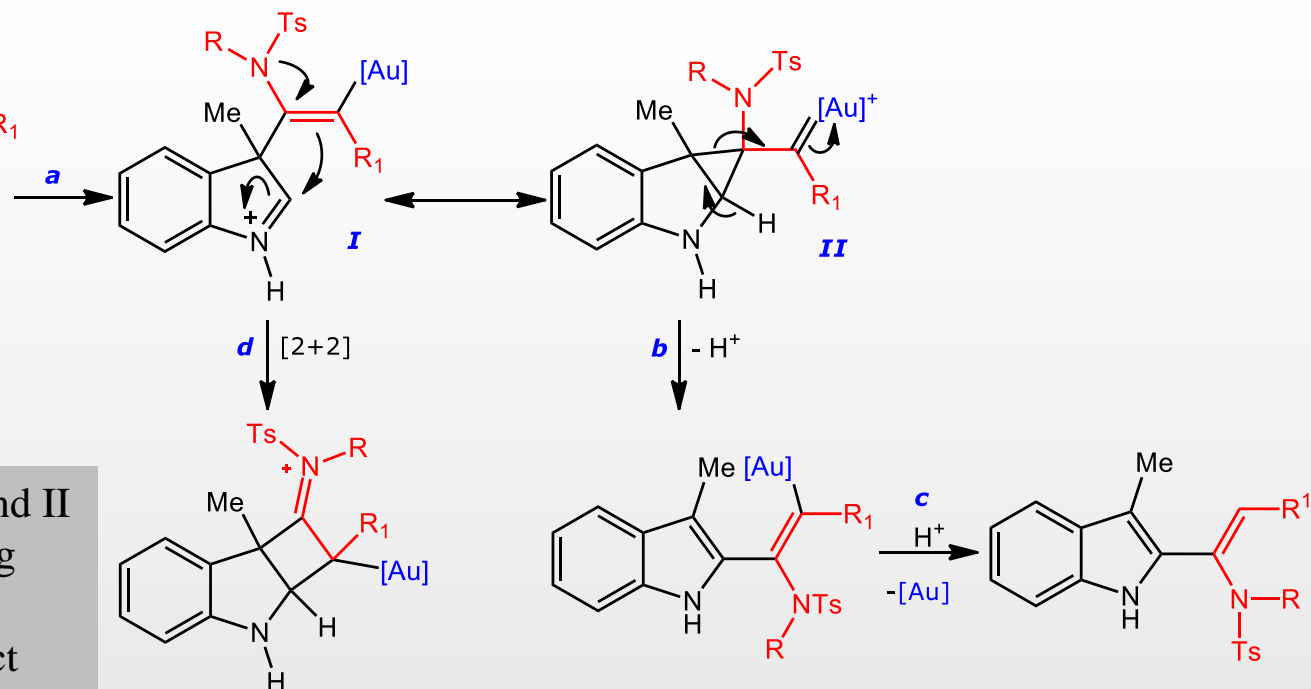
N. Zheng, Y.-Y. Chang, L.-J. Zhang, J.-X. Gong, Z. Yang *Chem. Asian J.* **2016**, *11*, 371.

synthesis of 2-vinylindoles



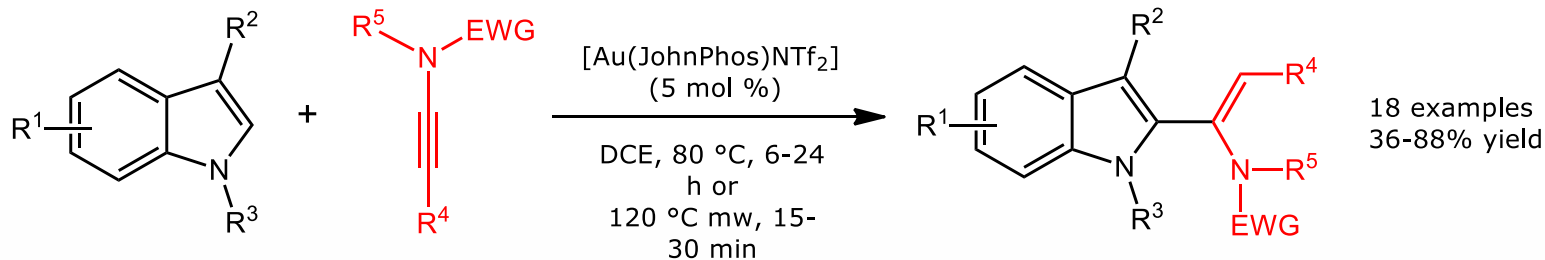
reaction mechanism

- a) formation of intermediates I and II
- b) loss of proton and ring opening
- c) protodeauration step
- d) formation of [2+2] cycloadduct

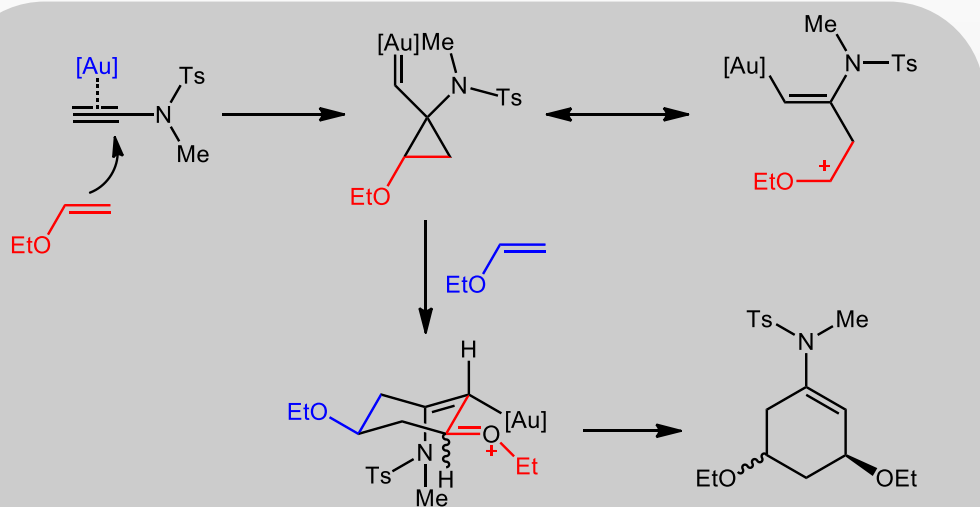


V. Pirovano, M. Negrato, G. Abbiati, M. Dell'Acqua, E. Rossi *Org. Lett.* **2016**, *18*, 4798.

synthesis of 2-vinylindoles



reaction mechanism



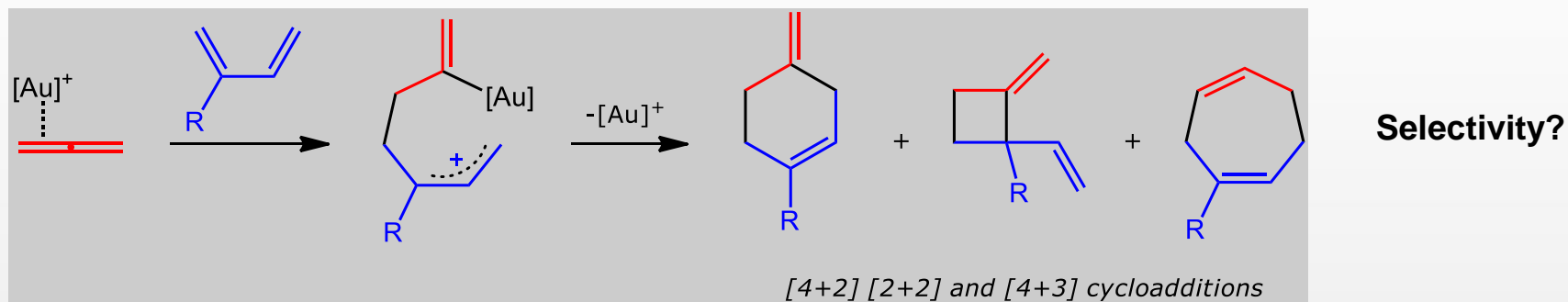
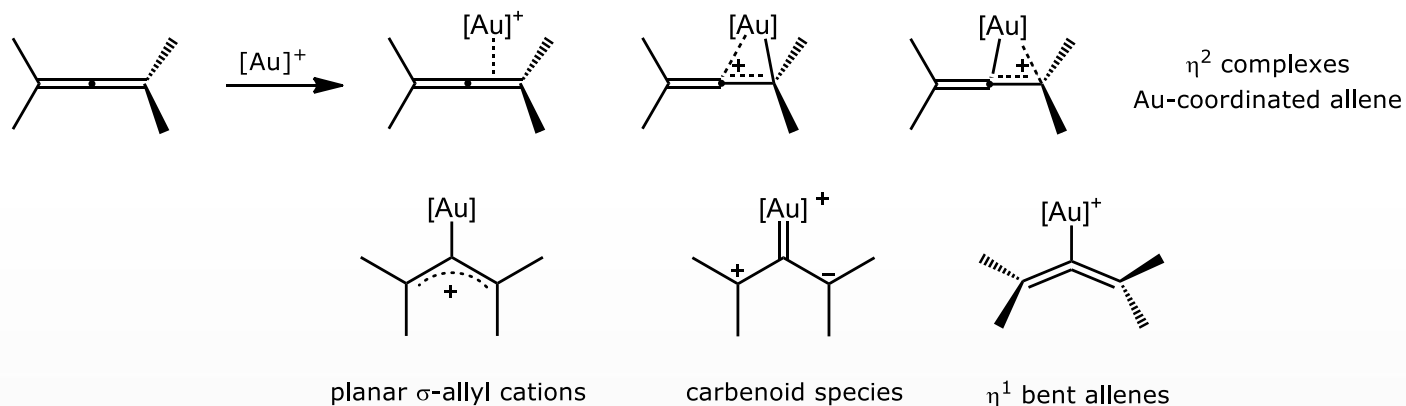
R. B. Dateer, B. S. Shaibu, R.-S. Liu *Angew. Chem. Int. Ed.* **2012**, *51*, 113.

reaction conditions

catalyst (mol%)	Yield (%)
HNTf ₂ (5)	59
AgNTf ₂ (5)	15
AuCl ₃	5
PtCl ₂	-
iPrAuNTf ₂ (5)	57
PPh ₃ AuNTf ₂ (5)	58
JohnPhosAuNTf ₂ (5), 80°C	65
JohnPhosAuNTf ₂ (5), mw, 120°C	70

V. Pirovano, M. Negrato, G. Abbiati, M. Dell'Acqua, E. Rossi *Org. Lett.* **2016**, *18*, 4798.

Allenes and gold catalysis



F. López, J. L. Mascareñas *Beilstein J. Org. Chem.* **2013**, *9*, 2250

S. Montserrat, G. Ujaque, F. López, J. L. Mascareñas, A. Lledós *Top. Curr. Chem.* **2011**, *302*, 225

Intramolecular processes

P. Mauleón, R. M. Zeldin, A. Z. González, F. D. Toste *J. Am. Chem. Soc.*, **2009**, *131*, 6348

B. Trillo, F. López, S. Montserrat, G. Ujaque, L. Castedo, A. Lledós, J. L. Mascareñas *Chem.–Eur. J.*, **2009**, *15*, 3336

B. Trillo, F. López, M. Gulías, L. Castedo, J. L. Mascareñas *Angew. Chem., Int. Ed.*, **2008**, *47*, 951

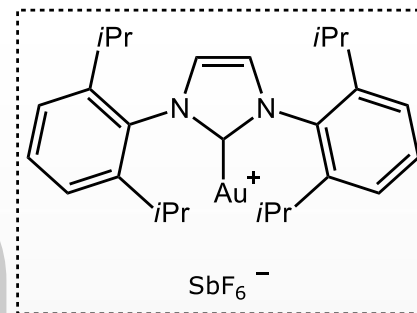
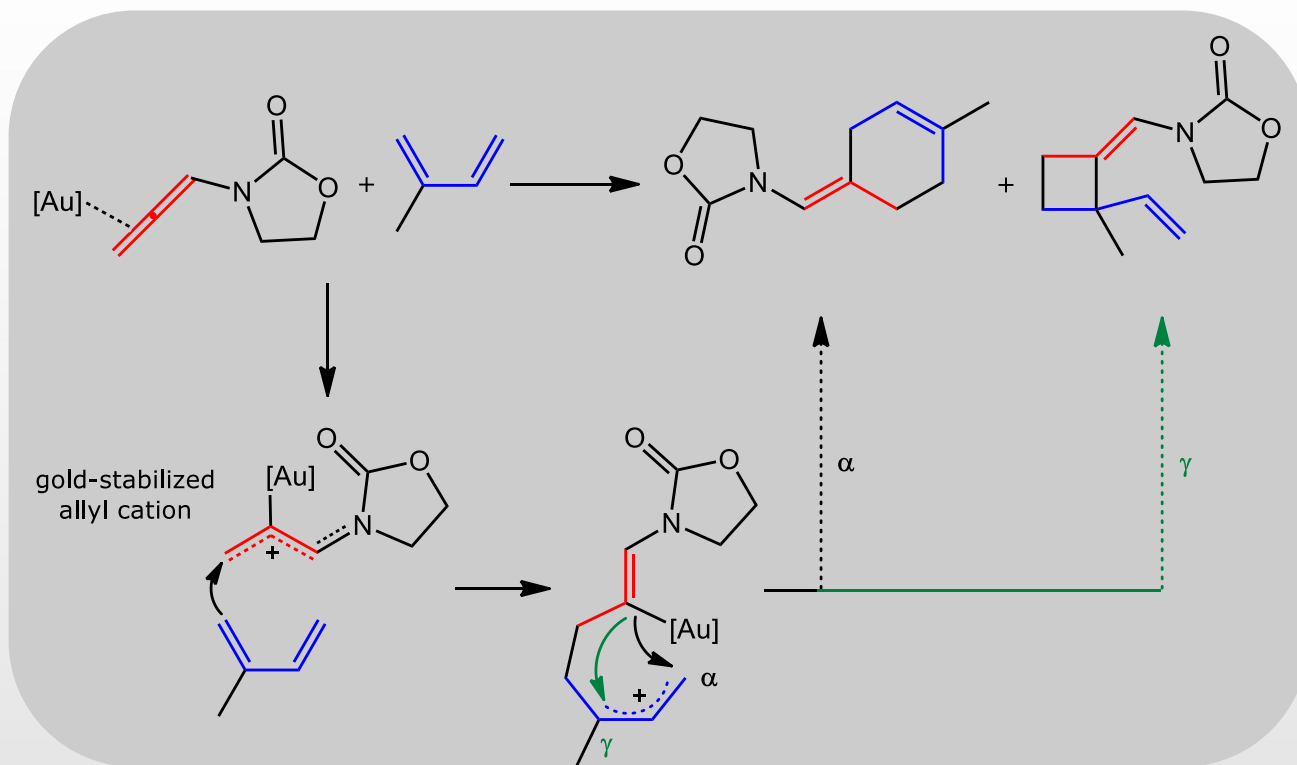
H. Teller, S. Flügge, R. Goddard, A. Fürstner *Angew. Chem., Int. Ed.*, **2010**, *49*, 1949

I. Alonso, B. Trillo, F. López, S. Montserrat, G. Ujaque, L. Castedo, A. Lledós, J. L. Mascareñas *J. Am. Chem. Soc.*, **2009**, *131*, 13020

intermolecular [4+2] cycloaddition reactions involving allenes

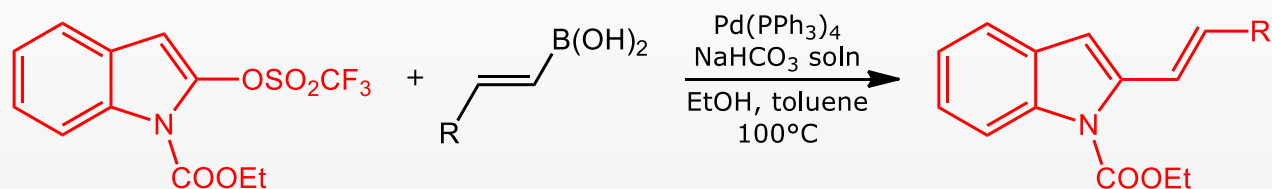
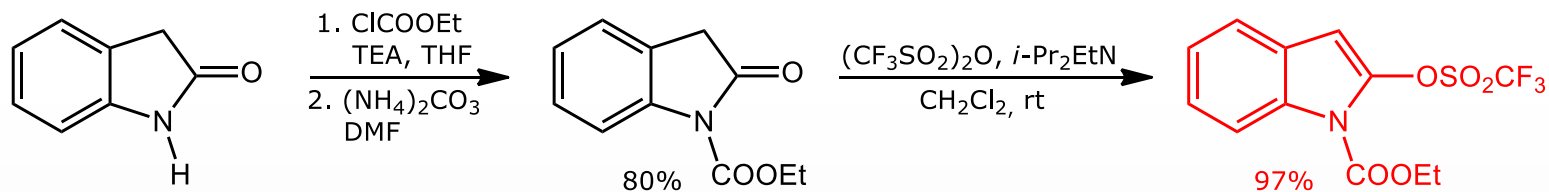


reaction mechanism



H. Faustino, F. López, L. Castedo, J. L. Mascareñas *Chem. Sci.* **2011**, 2, 633

stereospecific synthesis of 2-vinylindoles

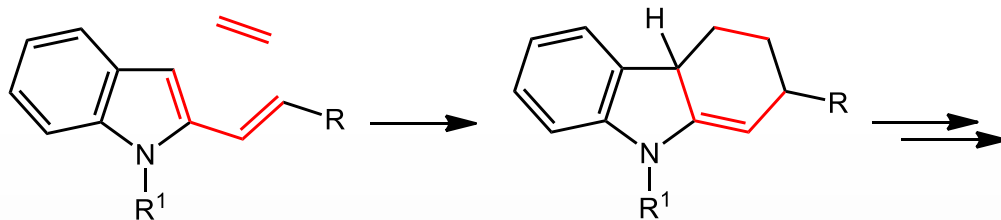


E-2-vinylindoles
12 examples, 75-99%

Suzuki-Miyaura coupling, high yield, stereospecific reaction

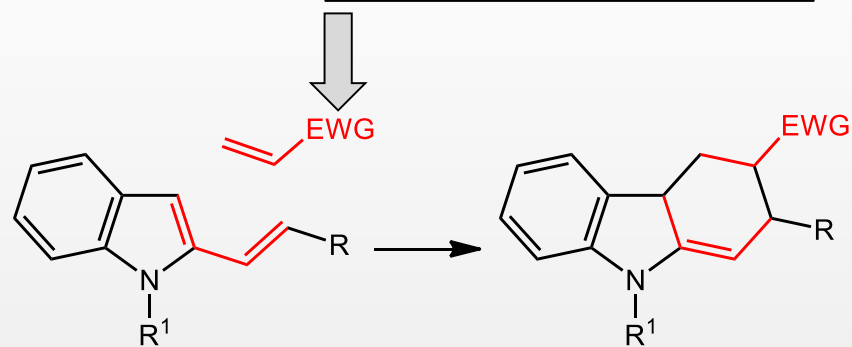
E. Rossi, G. Abbiati, V. Canevari, G. Celentano, E. Magri *Synthesis*, **2006**, 299
V. Pirovano, L. Decataldo, E. Rossi, R. Vicente *Chem. Commun.* **2013**, 49, 3594

2-vinylindoles as 4π-component in cycloaddition reactions

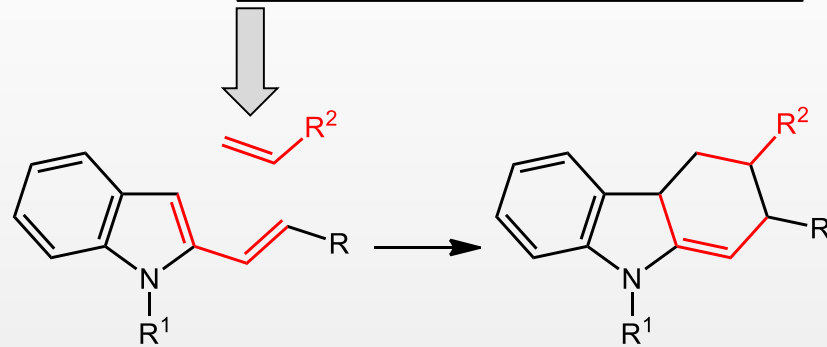


- ✓ Diels-Alder cycloaddition
- ✓ [4+2] cycloaddition
- ✓ [4+2] cyclization

σ-activation by Lewis acids



π-activation by transition metals

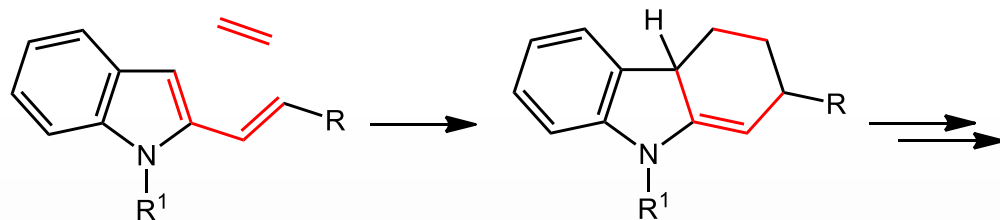


- ✓ Substrates
- ✓ Catalysts
- ✓ Reaction conditions



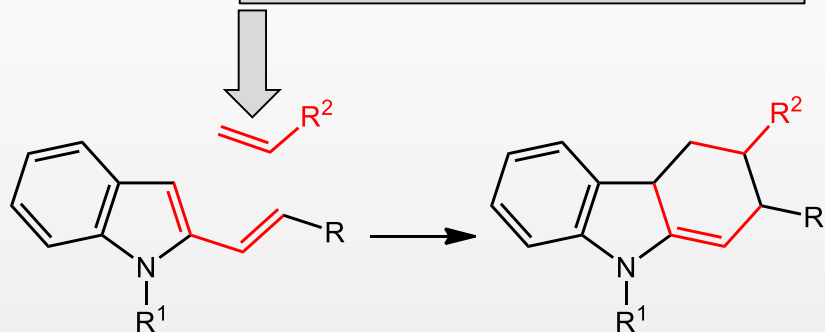
Selectivity

2-vinylindoles as 4π-component in cycloaddition reactions

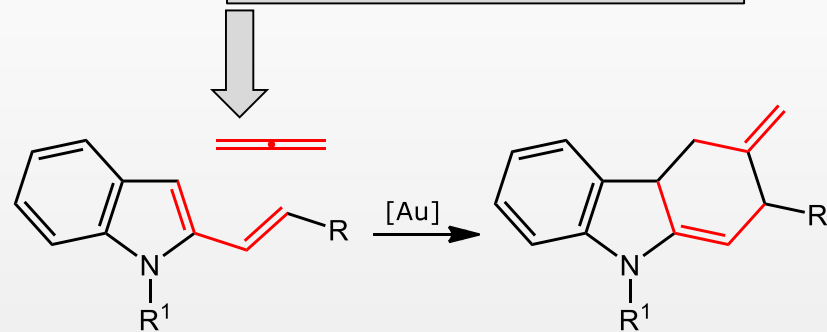


- ✓ Diels-Alder cycloaddition
- ✓ [4+2] cycloaddition
- ✓ [4+2] cyclization

π-activation by transition metals



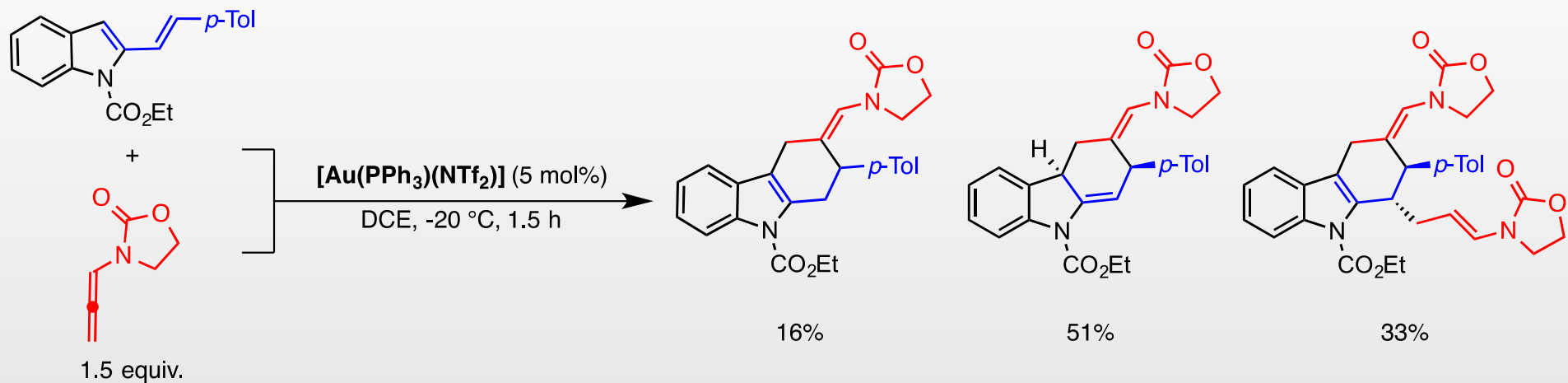
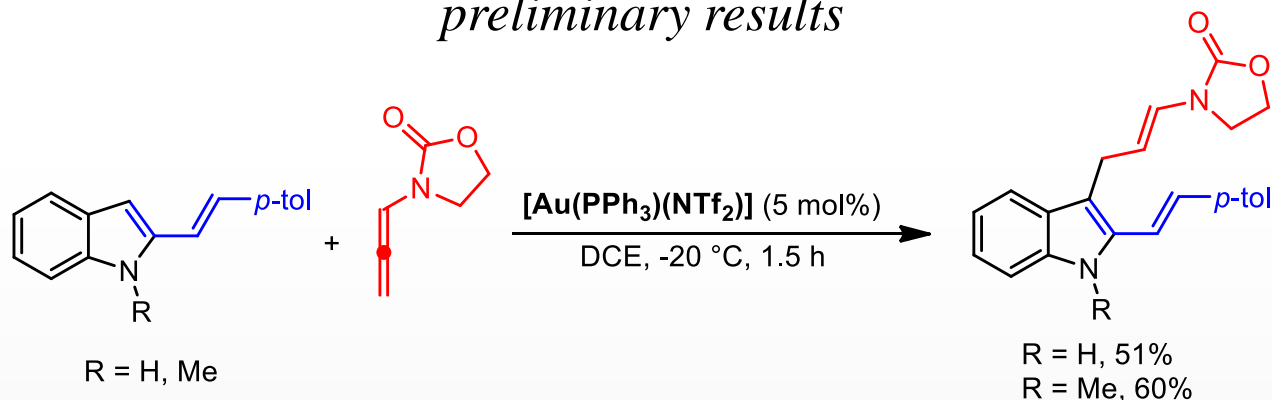
π-activation by gold catalysis



- ✓ Substrates
- ✓ Catalysts
- ✓ Reaction conditions

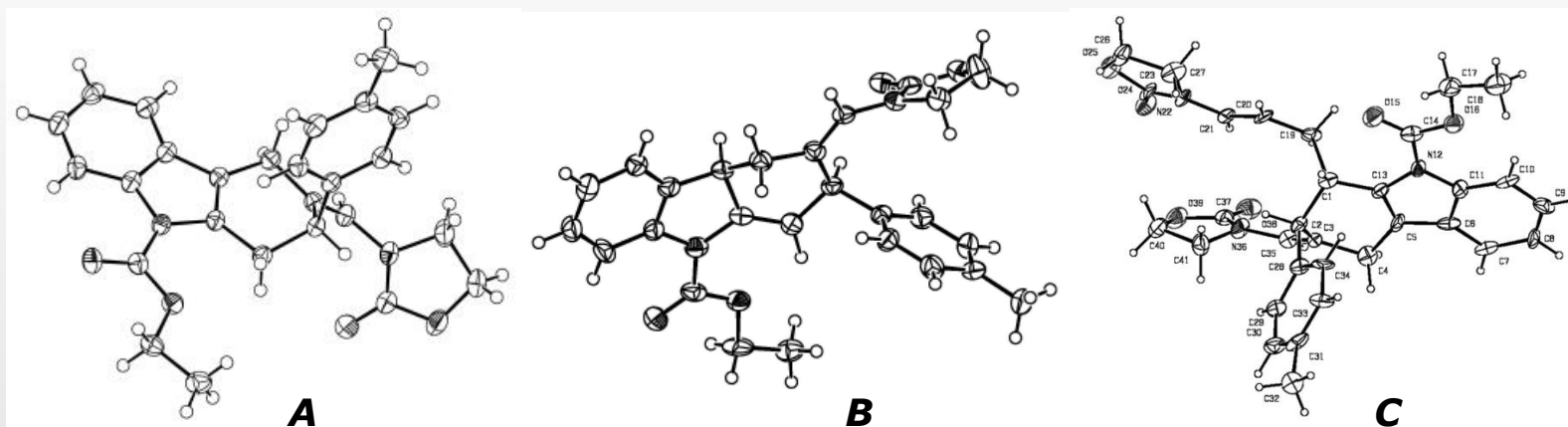
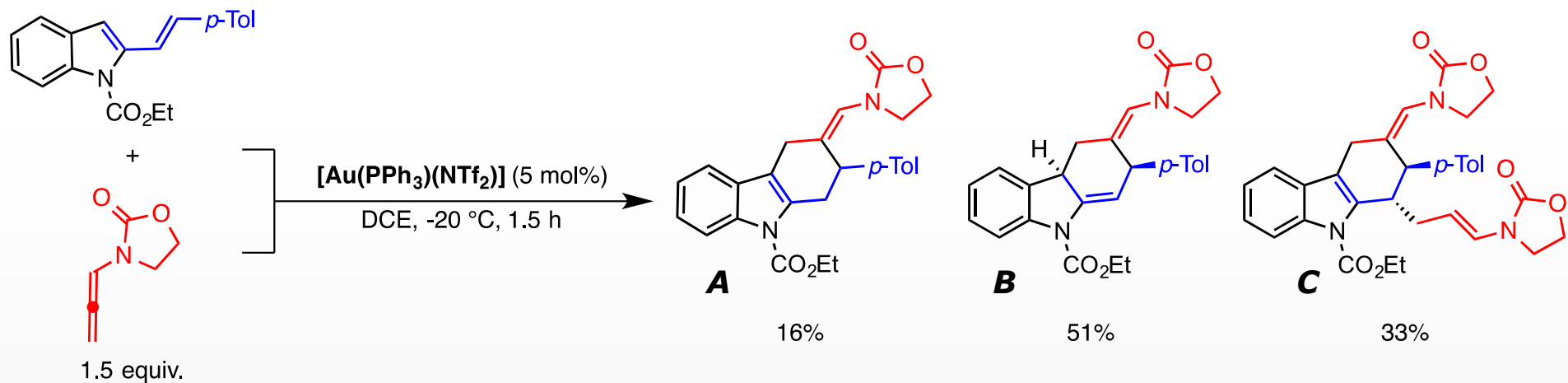
⇒ **Selectivity**

2-vinylindoles and allenamides
preliminary results



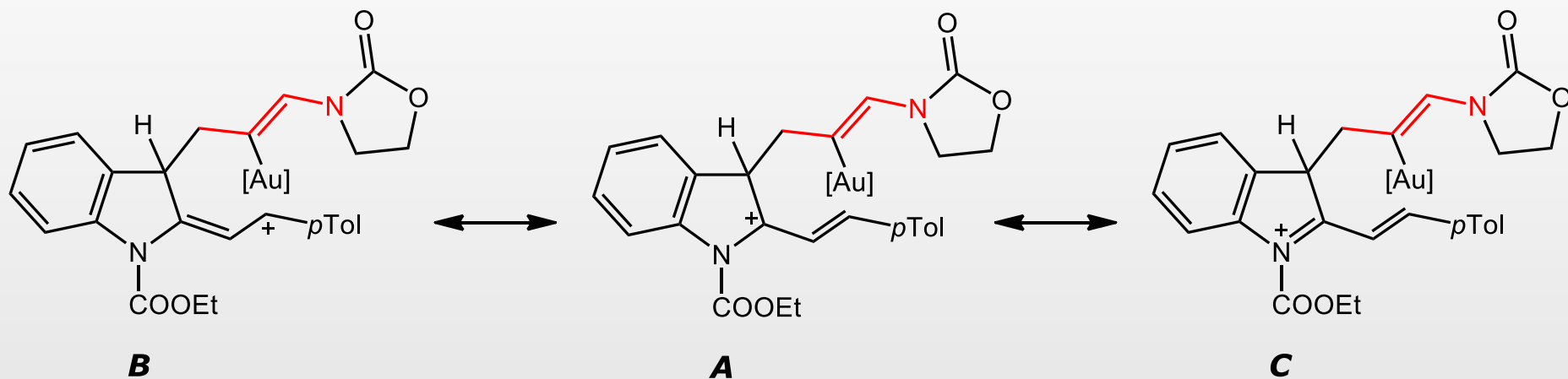
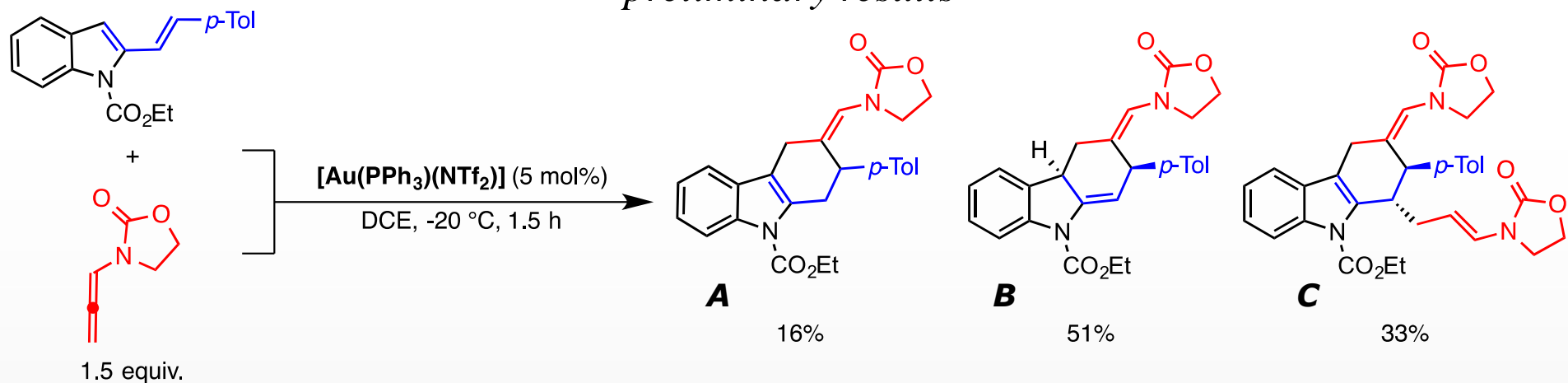
V. Pirovano, L. Decataldo, E. Rossi, R. Vicente *Chem. Commun.* **2013**, 49, 3594

2-vinylindoles and allenamides

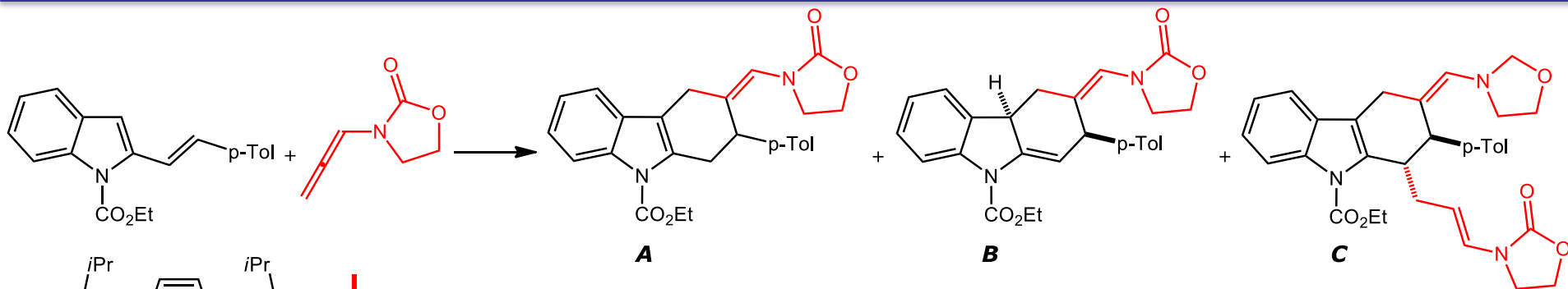


V. Pirovano, L. Decataldo, E. Rossi, R. Vicente *Chem. Commun.* **2013**, 49, 3594

2-vinylindoles and allenamides
preliminary results

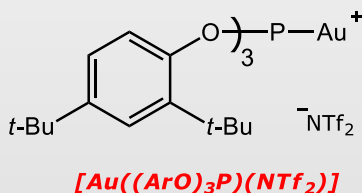
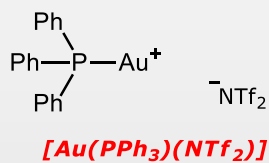
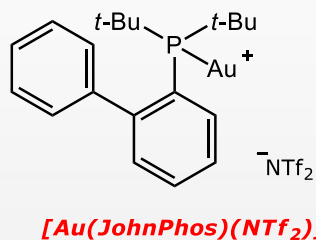
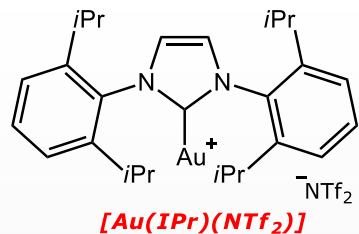


Gold catalysis and indole chemistry



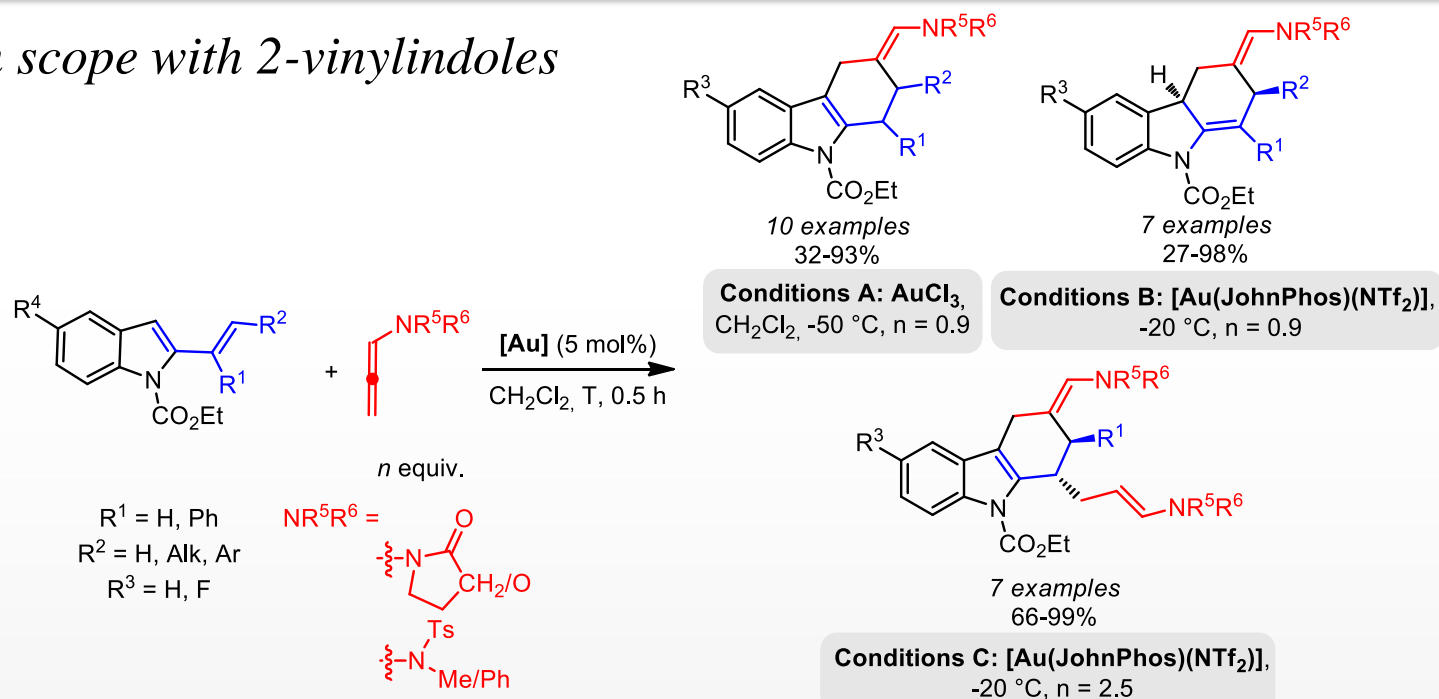
reaction conditions

entry	allene equiv.	Catalyst (5 mol%)	solvent	T (°C)	[M]	A	B	C
1	0.9	[Au(IPr)(NTf ₂)]	DCE	-20	0.1	5	75	-
2	0.9	[Au(JohnPhos)(NTf ₂)]	DCE	-20	0.1	-	81	8
3	0.9	[Au(PPh ₃)(NTf ₂)]	DCE	-20	0.1	54	-	9
4	0.9	[Au((ArO) ₃ P)(NTf ₂)]	DCE	-20	0.1	65	18	-
5	0.9	[Au(JohnPhos)(NTf ₂)]	DCM	-20	0.05	-	80	-
6	2.5	[Au(JohnPhos)(NTf ₂)]	DCM	-20	0.05	-	-	95
7	0.9	AuCl ₃	DCM	-50	0.1	83	-	-

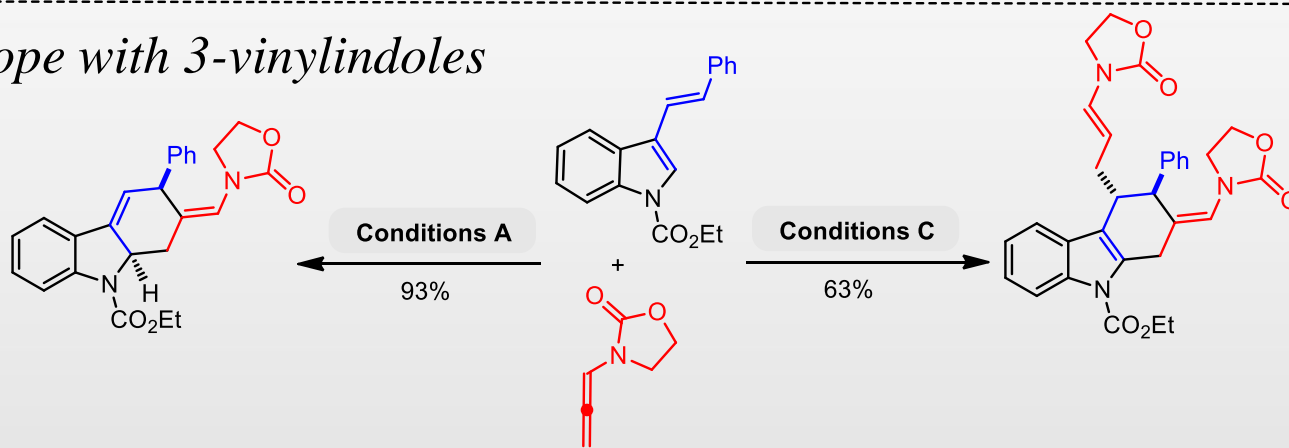


V. Pirovano, L. Decataldo, E. Rossi, R. Vicente *Chem. Commun.* **2013**, 49, 3594

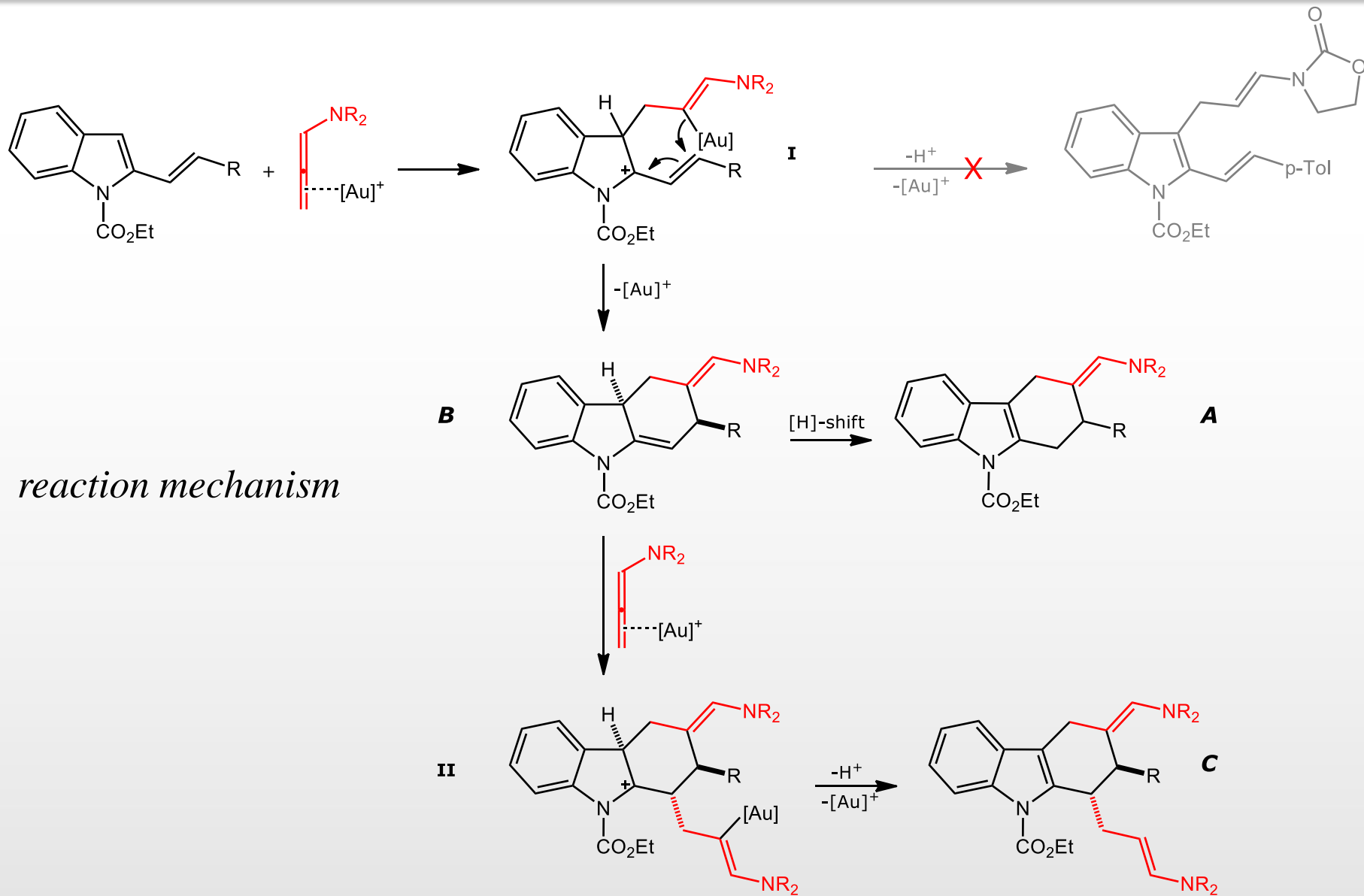
reaction scope with 2-vinylindoles



reaction scope with 3-vinylindoles

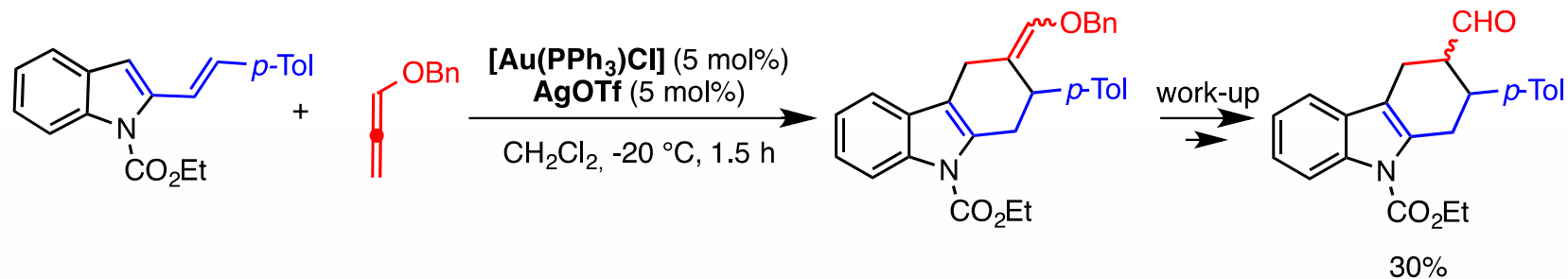


V. Pirovano, L. Decataldo, E. Rossi, R. Vicente *Chem. Commun.* **2013**, 49, 3594

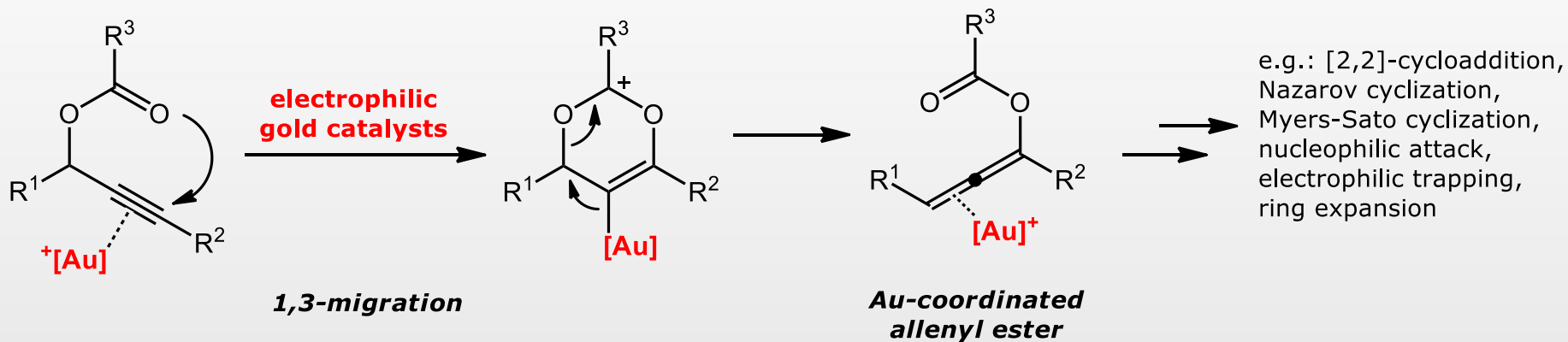


V. Pirovano, L. Decataldo, E. Rossi, R. Vicente *Chem. Commun.* **2013**, 49, 3594

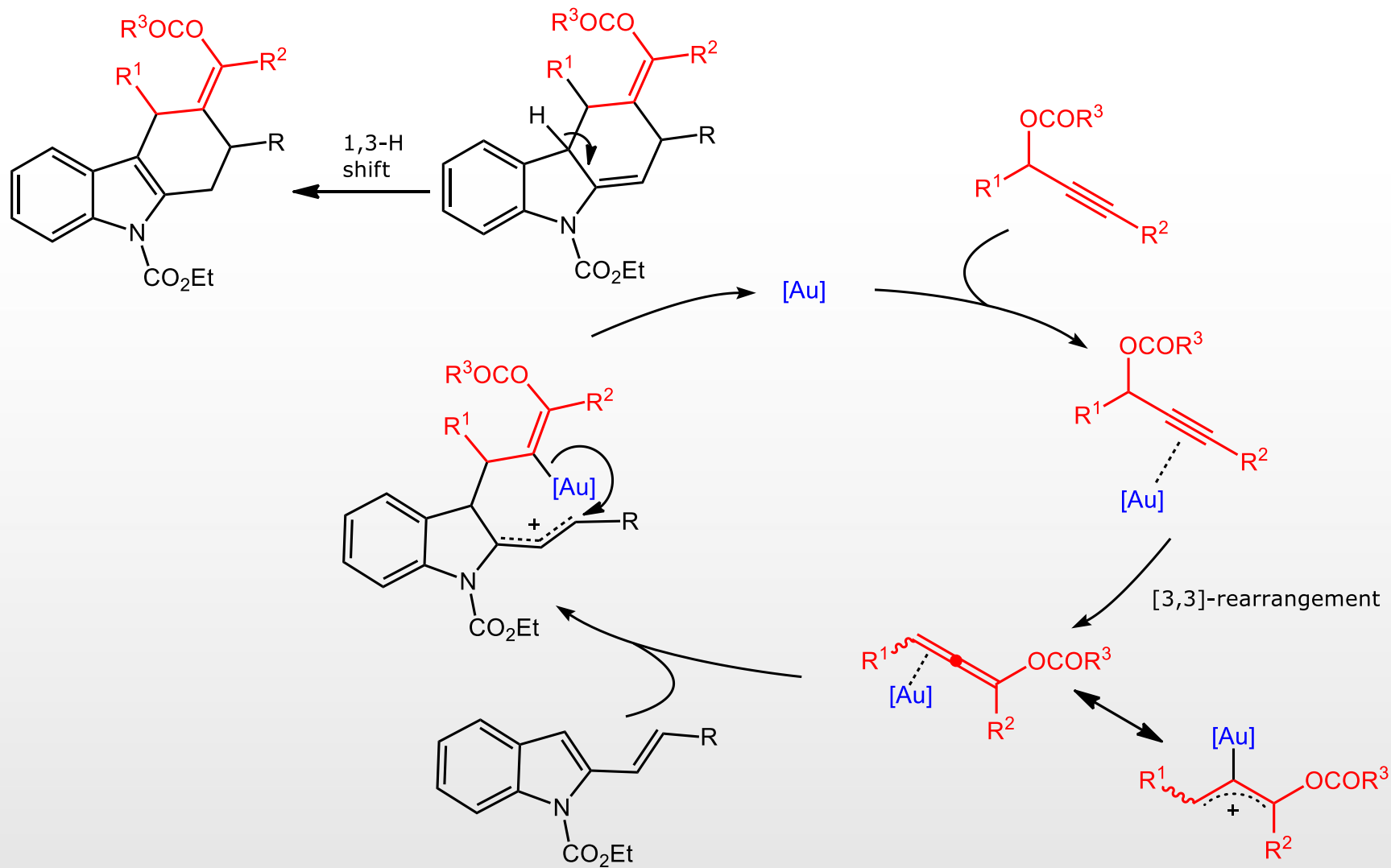
2-vinylindoles and allenylethers



gold catalyzed propargylic esters rearrangement

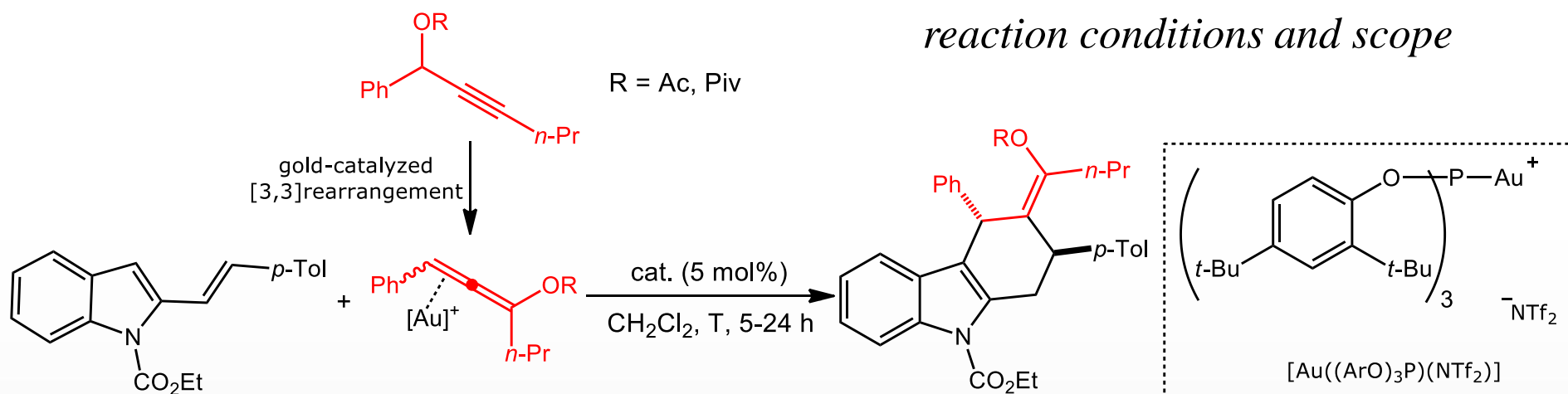


2-vinylindoles and allenylesters, cascade reaction

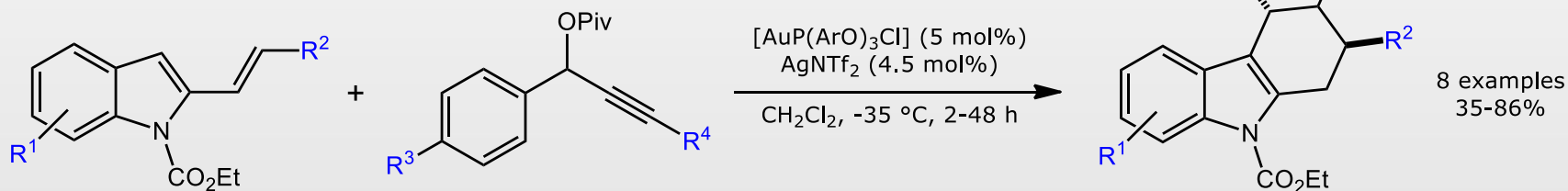


V. Pirovano, E. Arpini, M. Dell'Acqua, R. Vicente, G. Abbiati, E. Rossi *Adv. Synth. Catal.* **2016**, 358, 403

reaction conditions and scope

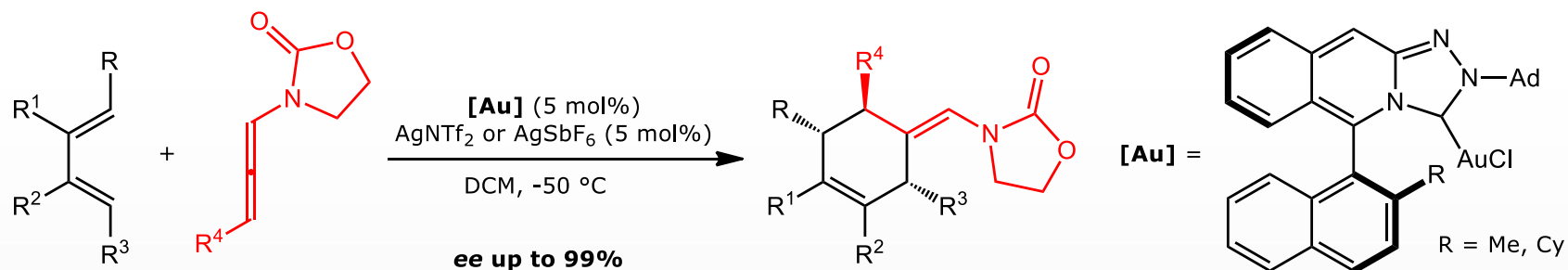


entry	allene (equiv.)	Catalyst (5 mol%)	T (°C)	t [h]	Yield [%]
1	R = Ac (1.1)	[Au((ArO) ₃ P)(NTf ₂)]	-20	24	31
2	R = Ac (1.1)	[Au((ArO) ₃ P)(NTf ₂)]	-20	24	52
3	R = Ac (1.1)	[Au((ArO) ₃ P)(NTf ₂)]	-35	24	64
4	R = Piv (1.5)	[Au((ArO) ₃ P)(NTf ₂)]	-35	5	75

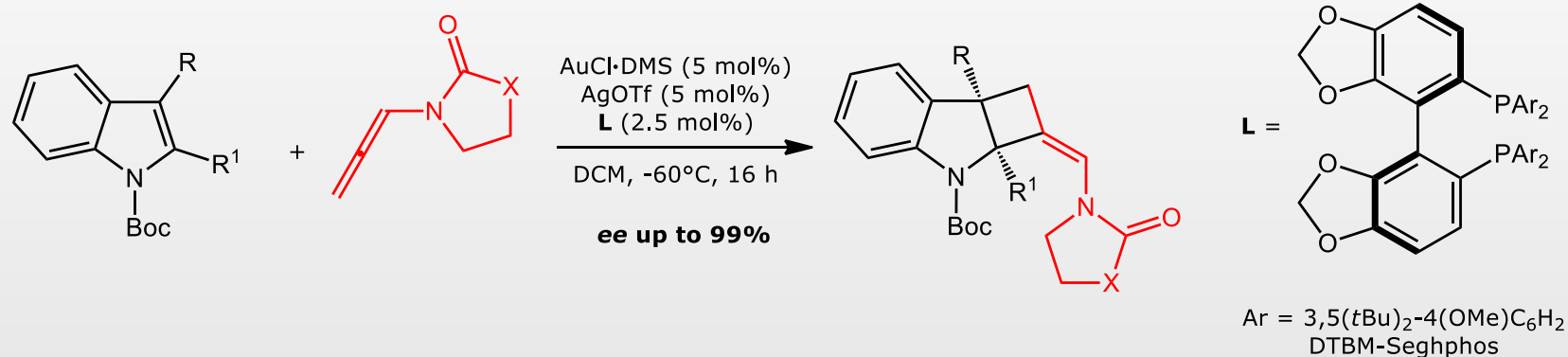


V. Pirovano, E. Arpini, M. Dell'Acqua, R. Vicente, G. Abbiati, E. Rossi *Adv. Synth. Catal.* **2016**, 358, 403

enantioselective cycloaddition reactions with allenamides

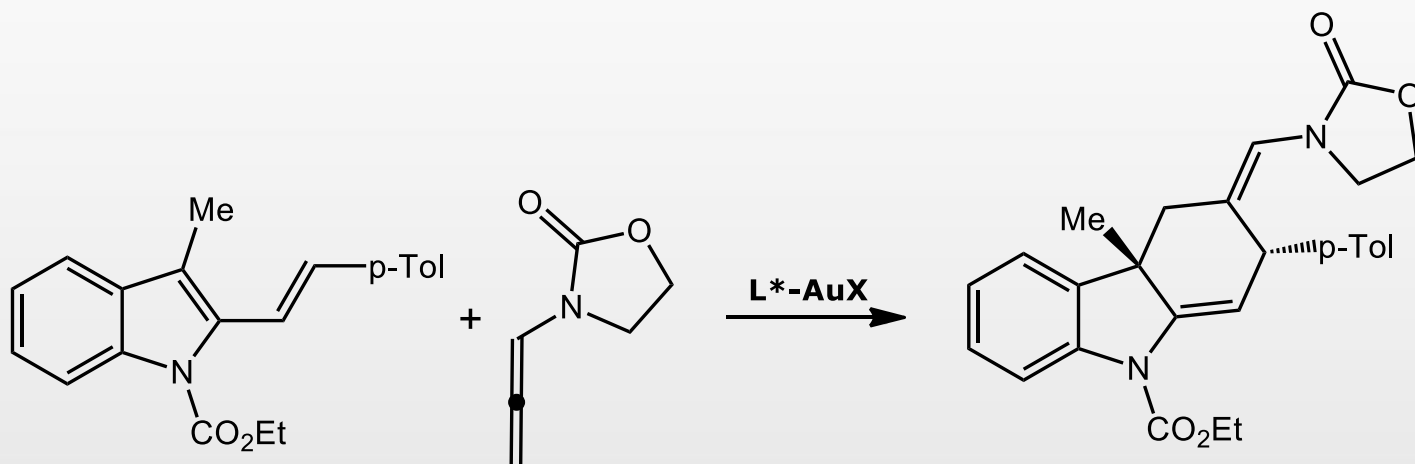
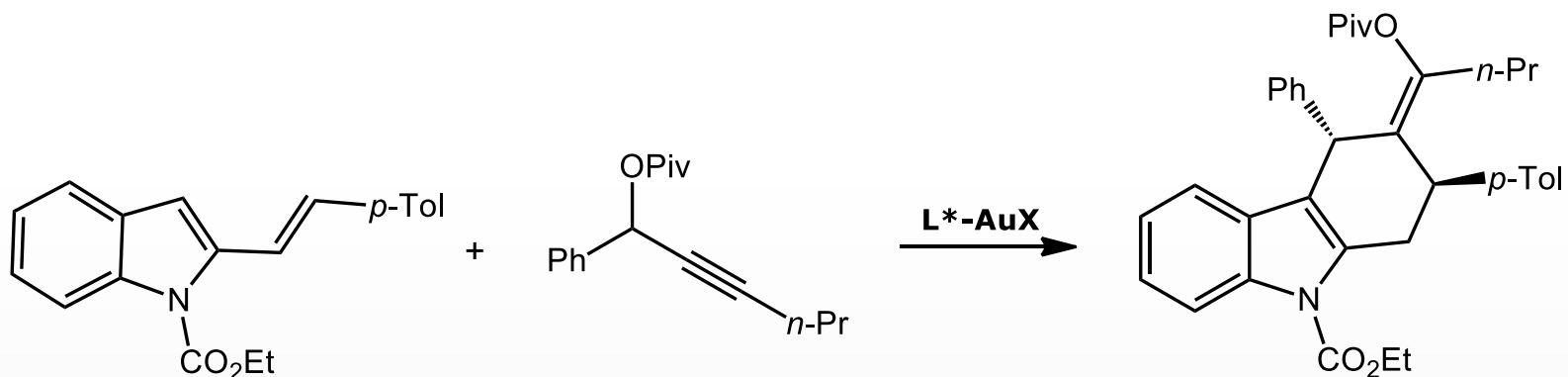


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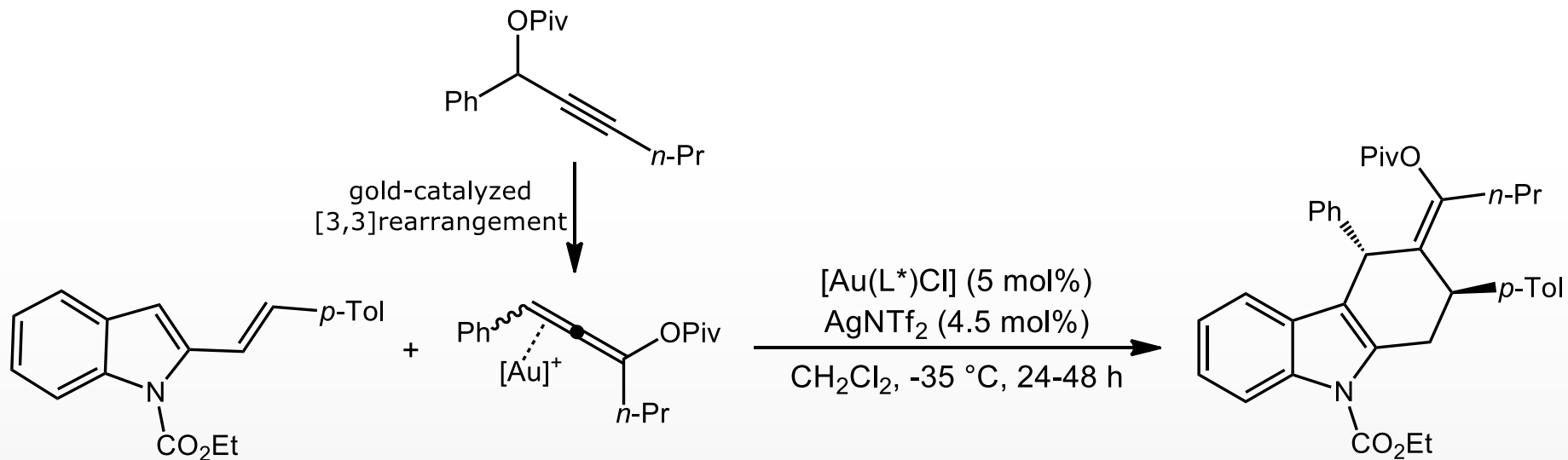
enantioselective cycloaddition reactions of 2-vinylindoles and allenes



V. Pirovano, E. Arpini, M. Dell'Acqua, R. Vicente, G. Abbiati, E. Rossi *Adv. Synth. Catal.* **2016**, 358, 403 – VIP.

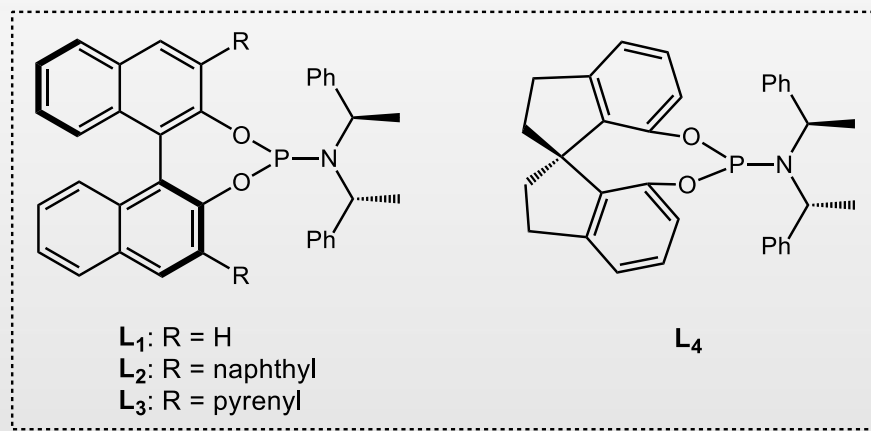
V. Pirovano, M. Borri, S. Rizzato, G. Abbiati, E. Rossi *Adv. Synth. Catal.* **2017**, 359, 1912.

enantioselective cycloaddition reactions of 2-vinylindoles and allenylesters



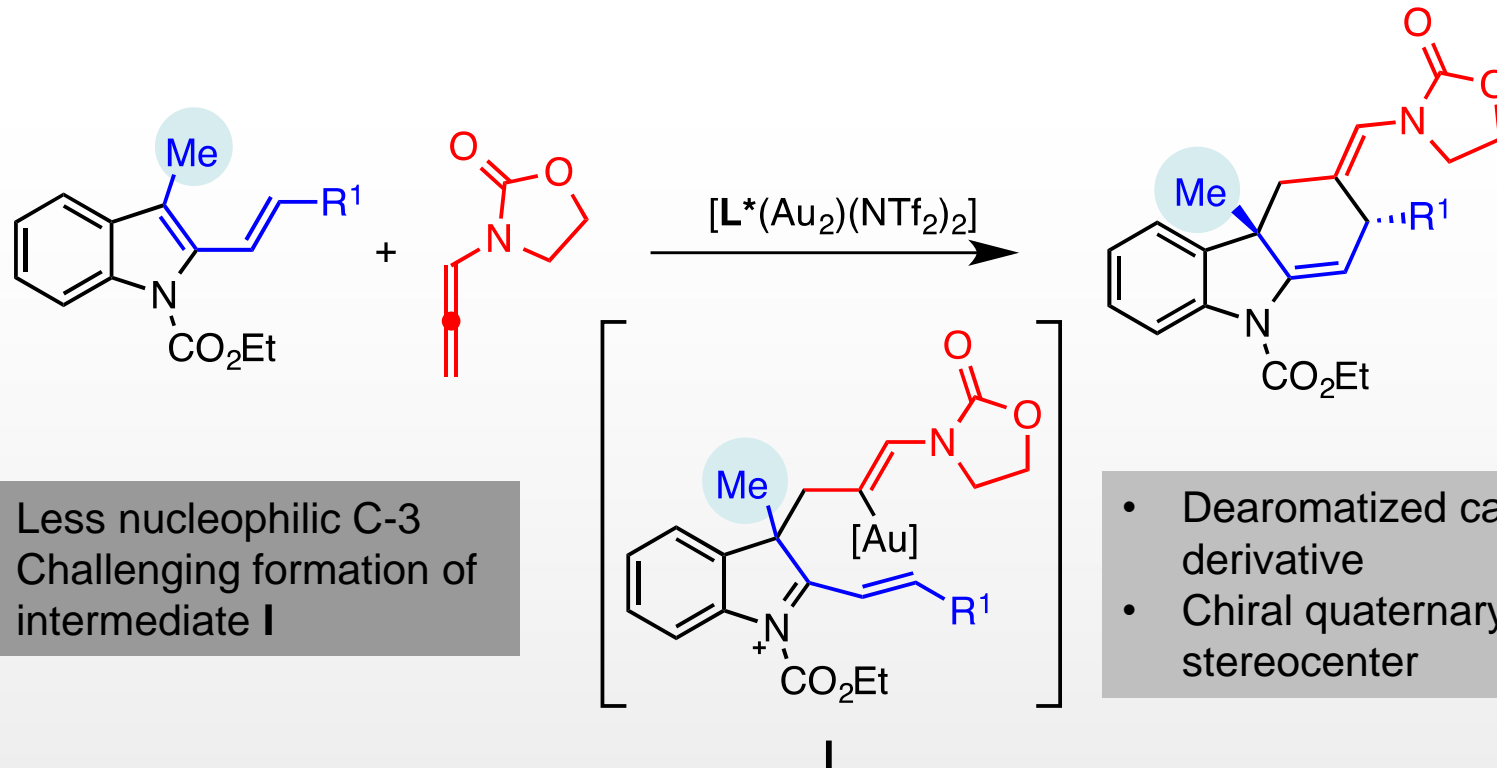
reaction conditions

entry	L*	Additive	t [h]	Yield [%]	e.r.
1	L1	-	24	90	40:60
2	L2	-	24	42	80:20
3	L3	-	48	47	81:19
4	L4	-	48	42	84:16
5	L4	4 Å ms	48	62	85:15



V. Pirovano, E. Arpini, M. Dell'Acqua, R. Vicente, G. Abbiati, E. Rossi *Adv. Synth. Catal.* **2016**, 358, 403 - VIP

enantioselective cycloaddition reactions of 2-vinyl-3-methylindoles and allenamides



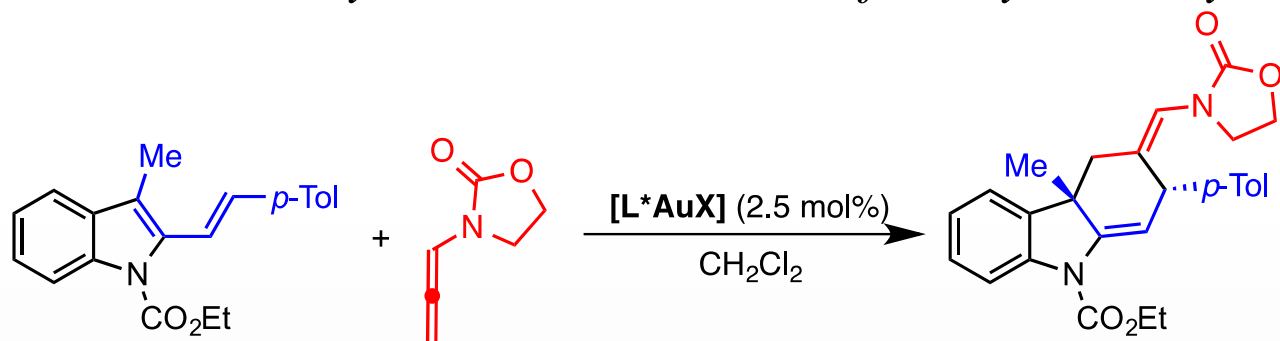
- Less nucleophilic C-3
- Challenging formation of intermediate I

- Dearomatized carbazole derivative
- Chiral quaternary stereocenter

L^* = *dinuclear chiral gold(I) catalyst*

V. Pirovano, M. Borri, S. Rizzato, G. Abbiati, E. Rossi *Adv. Synth. Catal.* **2017**, 359, 1912.

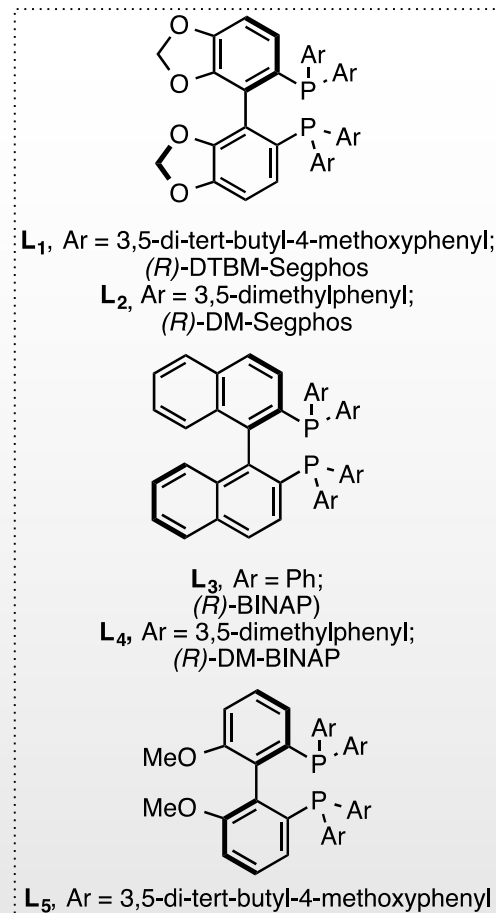
enantioselective cycloaddition reactions of 2-vinyl-3-methylindoles and allenamides



reaction conditions

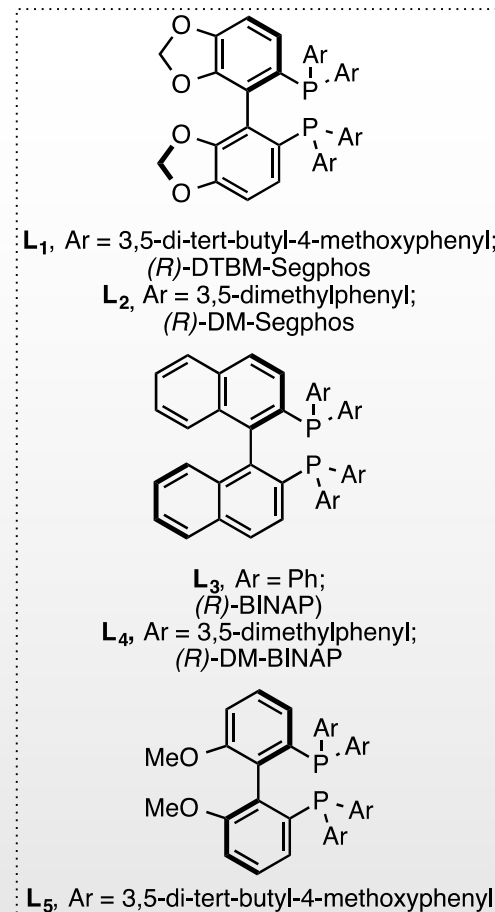
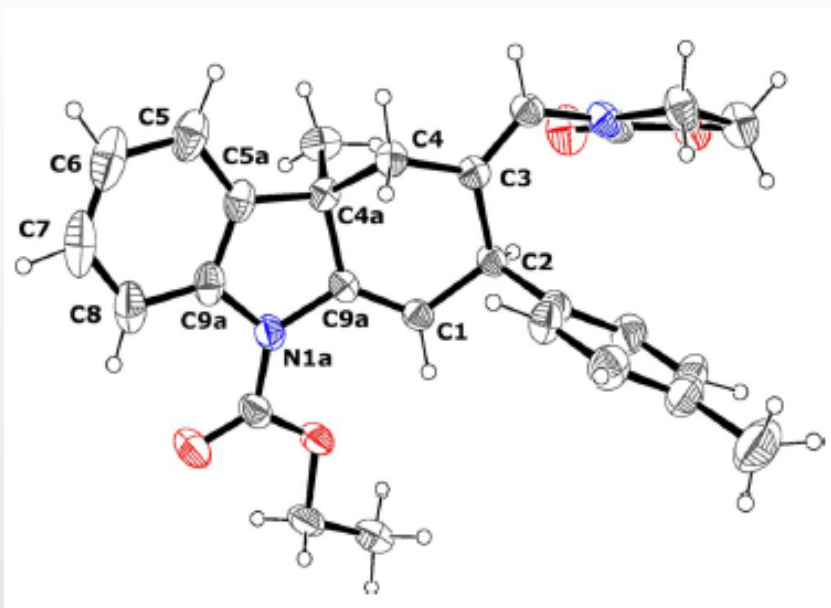
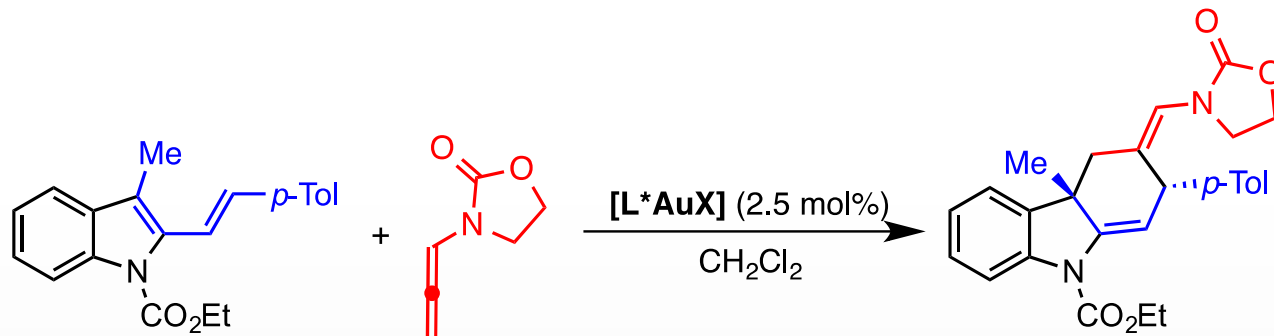
Entry	allene equiv.	Catalyst	T (°C)	t (h)	Yield (%)	e.r.
1	0.9	$[L_1Au_2Cl_2]/AgNTf_2$	-20	1	72	7:93
2	1.2	$[L_1Au_2(NTf_2)_2]$	-40	18	75	4:96
3	1.2 (0.05 M)	$[L_1Au_2(NTf_2)_2]$	-40	18	79	4:96*
4	0.9	$[L_2Au_2Cl_2]/AgNTf_2$	-20	1	54	27:73
5	0.9	$[L_3Au_2Cl_2]/AgNTf_2$	-20	1	52	37:63
6	0.9	$[L_4Au_2Cl_2]/AgNTf_2$	-20	1	75	20:80
7	0.9	$[L_5Au_2Cl_2]/AgNTf_2$	-20	1	48	12:88

*: increased to 1:99 after a single recrystallization from CH_2Cl_2 /pentane



V. Pirovano, M. Borri, S. Rizzato, G. Abbiati, E. Rossi *Adv. Synth. Catal.* **2017**, 359, 1912.

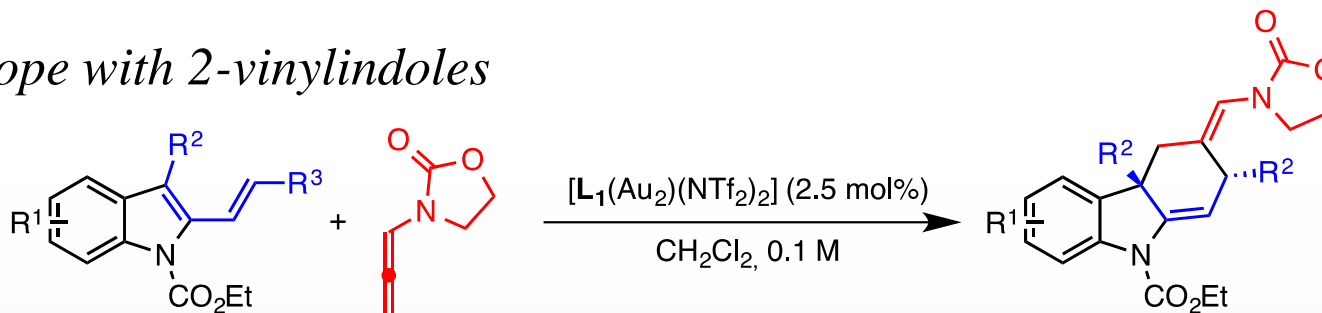
enantioselective cycloaddition reactions of 2-vinyl-3-methylindoles and allenamides



V. Pirovano, M. Borri, S. Rizzato, G. Abbiati, E. Rossi *Adv. Synth. Catal.* **2017**, 359, 1912.

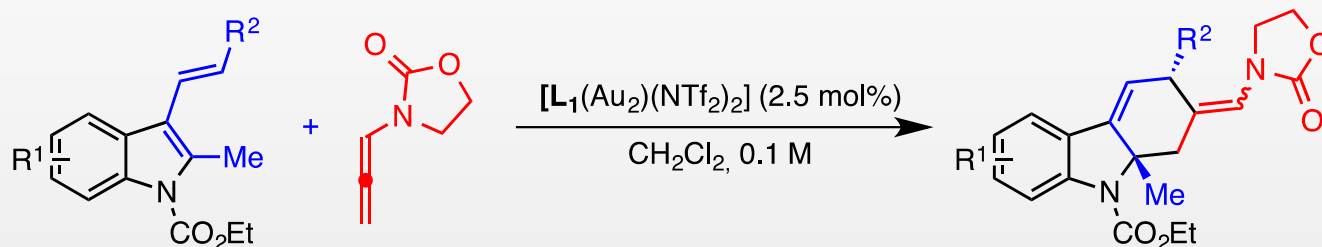
enantioselective cycloaddition reactions of 2-vinyl-3-methylindoles and allenamides

reaction scope with 2-vinylindoles



8 examples
40-90%
e.r. up to 97:3

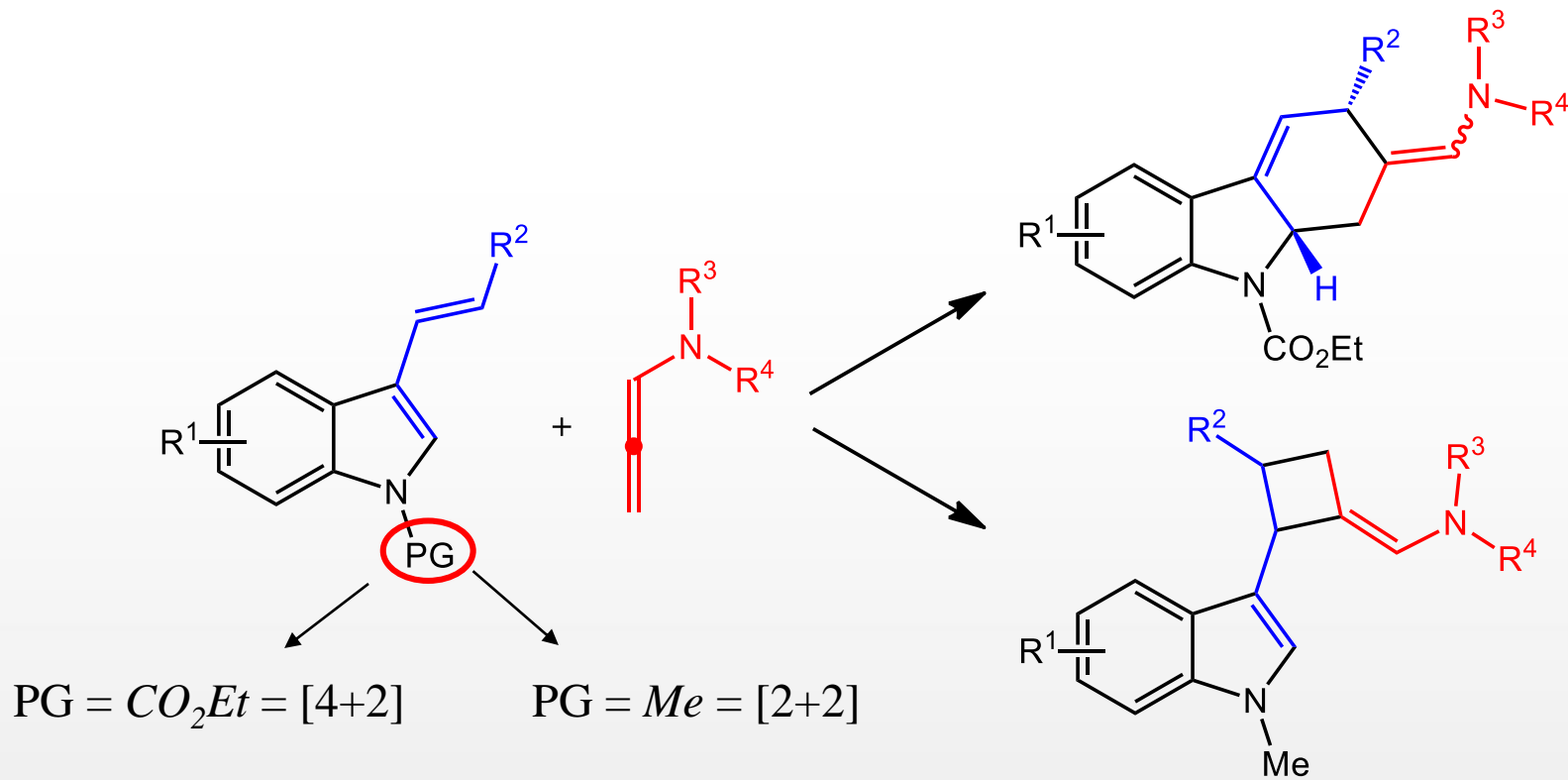
reaction scope with 3-vinylindoles



4 examples
80-97%
(E)/(Z) up to 20:1
e.r. (E) up to 98:2
e.r. (Z) up to 99:1

V. Pirovano, M. Borri, S. Rizzato, G. Abbiati, E. Rossi *Adv. Synth. Catal.* **2017**, 359, 1912.

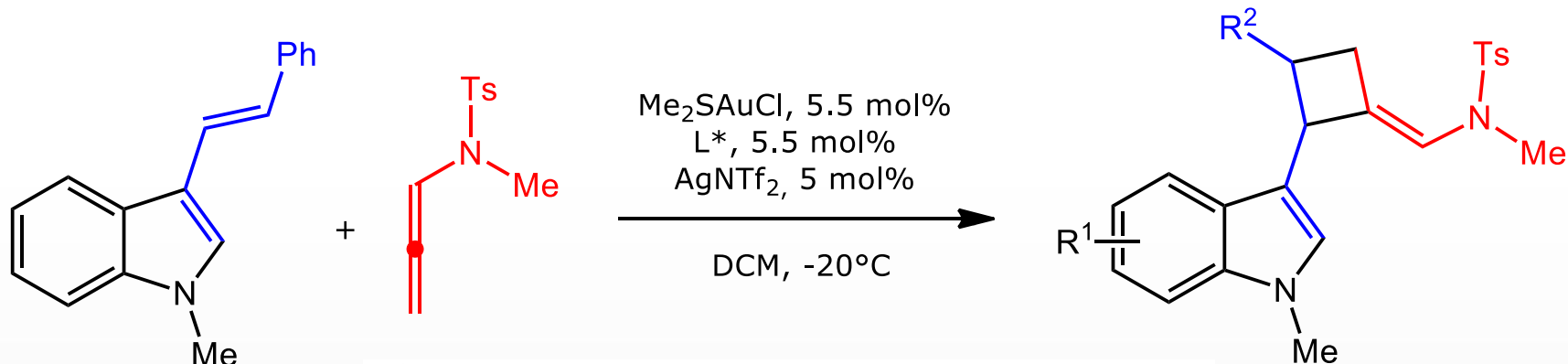
enantioselective cycloaddition reactions of 3-vinylindoles and allenamides



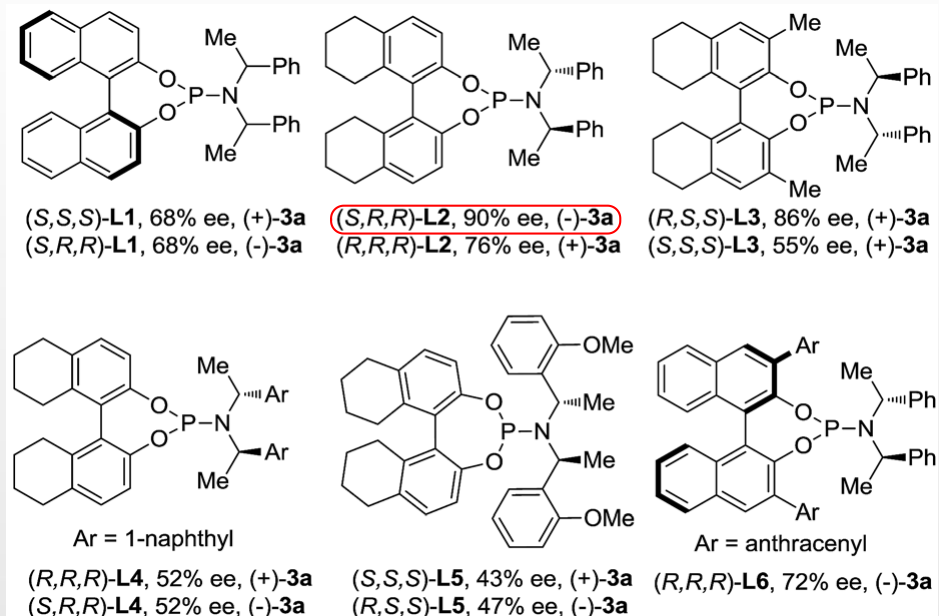
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enantioselective cycloaddition reactions of 3-vinylindoles and allenamides



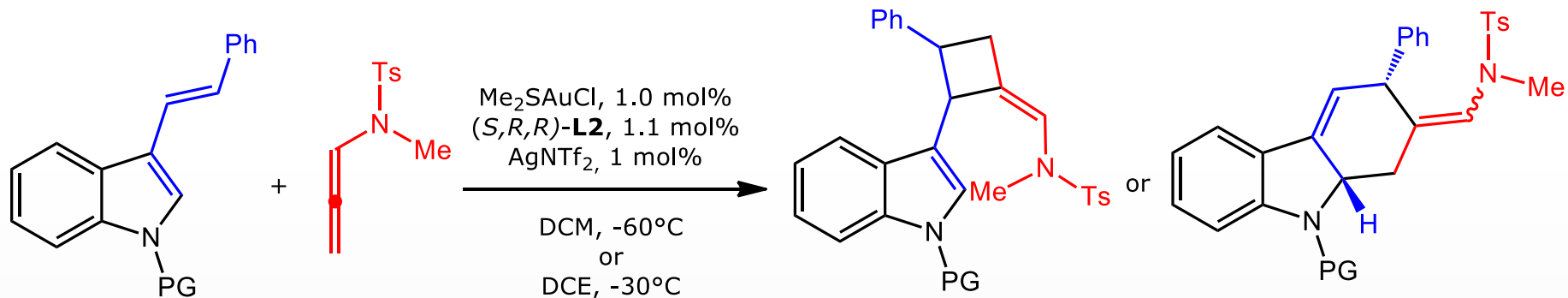
reaction conditions



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enantioselective cycloaddition reactions of 3-vinylindoles and allenamides



reaction conditions and scope

PG	[]	condition	yield (%)	Z/E	ee (%)
Me	[2+2]	A	99	1:0	96
Bn	[2+2]	A	94	1:0	92
Allyl	[2+2]	A	93	1:0	91
H	[2+2]	A	95	1:0	72
CO ₂ Et	[4+2]	B	95	5.3:1	95
Ts	[4+2]	B	86	4.2:1	91
Ac	[4+2]	B	67	7.3:1	94

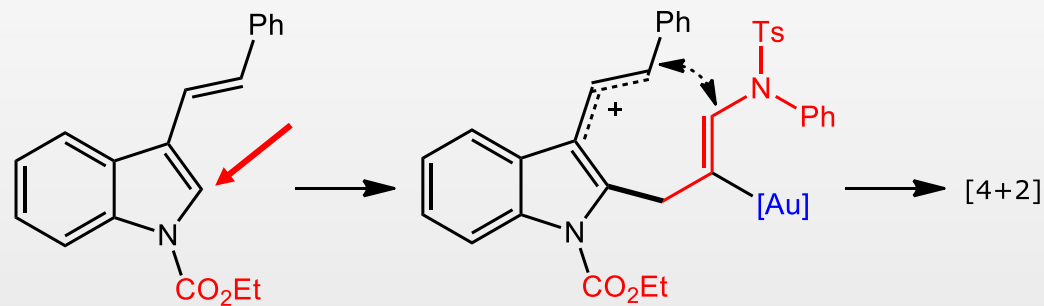
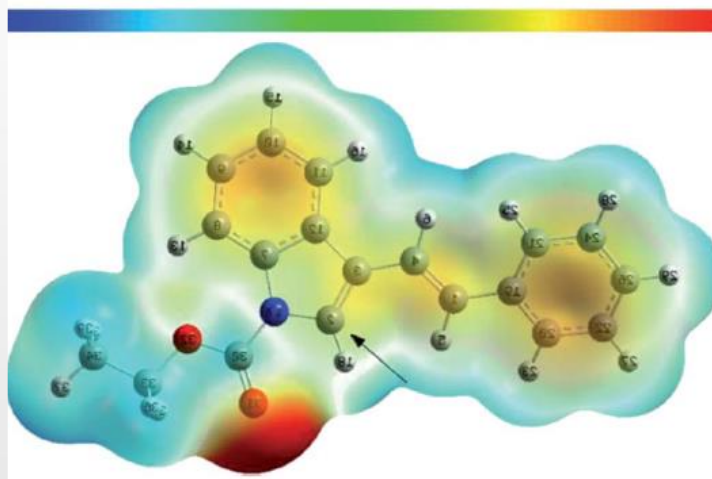
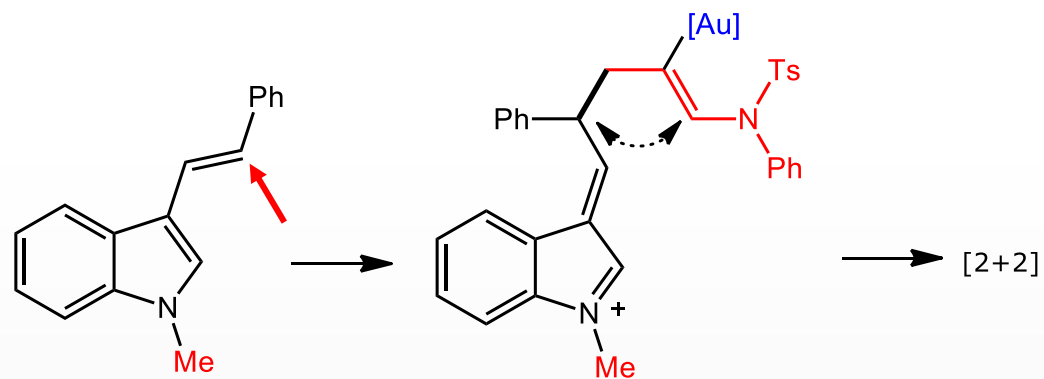
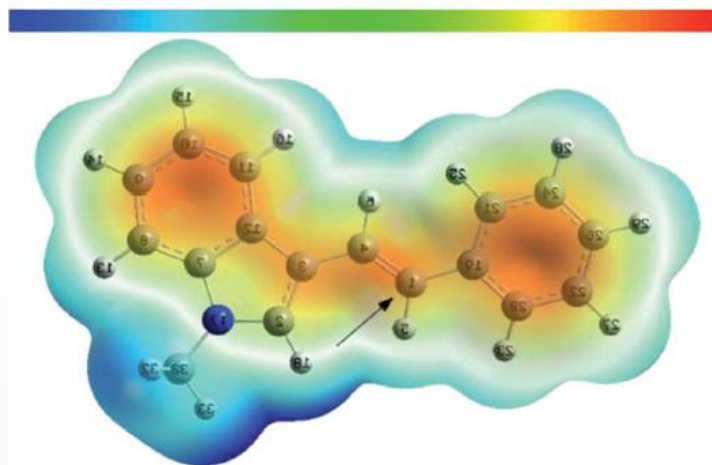
13 examples
58-99%
Z/E 1:0
ee 82-96%

12 examples
82-99%
Z/E 3-6.1:1
ee (Z) 89-97% ee (E) 88-92%

Y. Wang, P. Zhang, Y. Liu, F. Xia, J. Zhang *Chem. Sci.* **2015**, 6, 5564.

H. Hu, Y. Wang, D. Qian, Z.-M. Zhang, L. Liu, J. Zhang *Org. Chem. Front.* **2016**, 3, 759.

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